

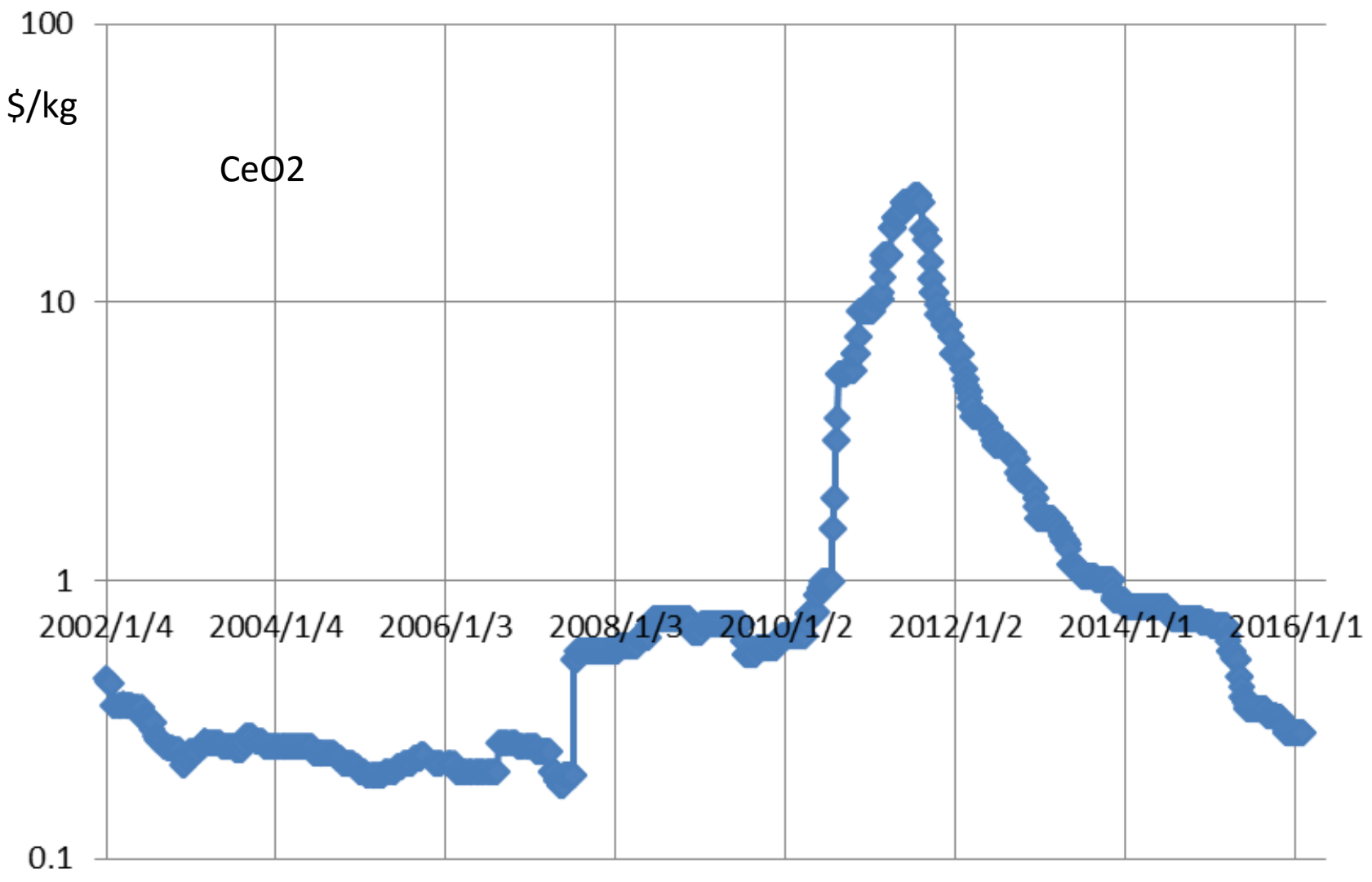
Life-cycle Analysis and Criticality of Ceramics and Elements

Kohmei HALADA

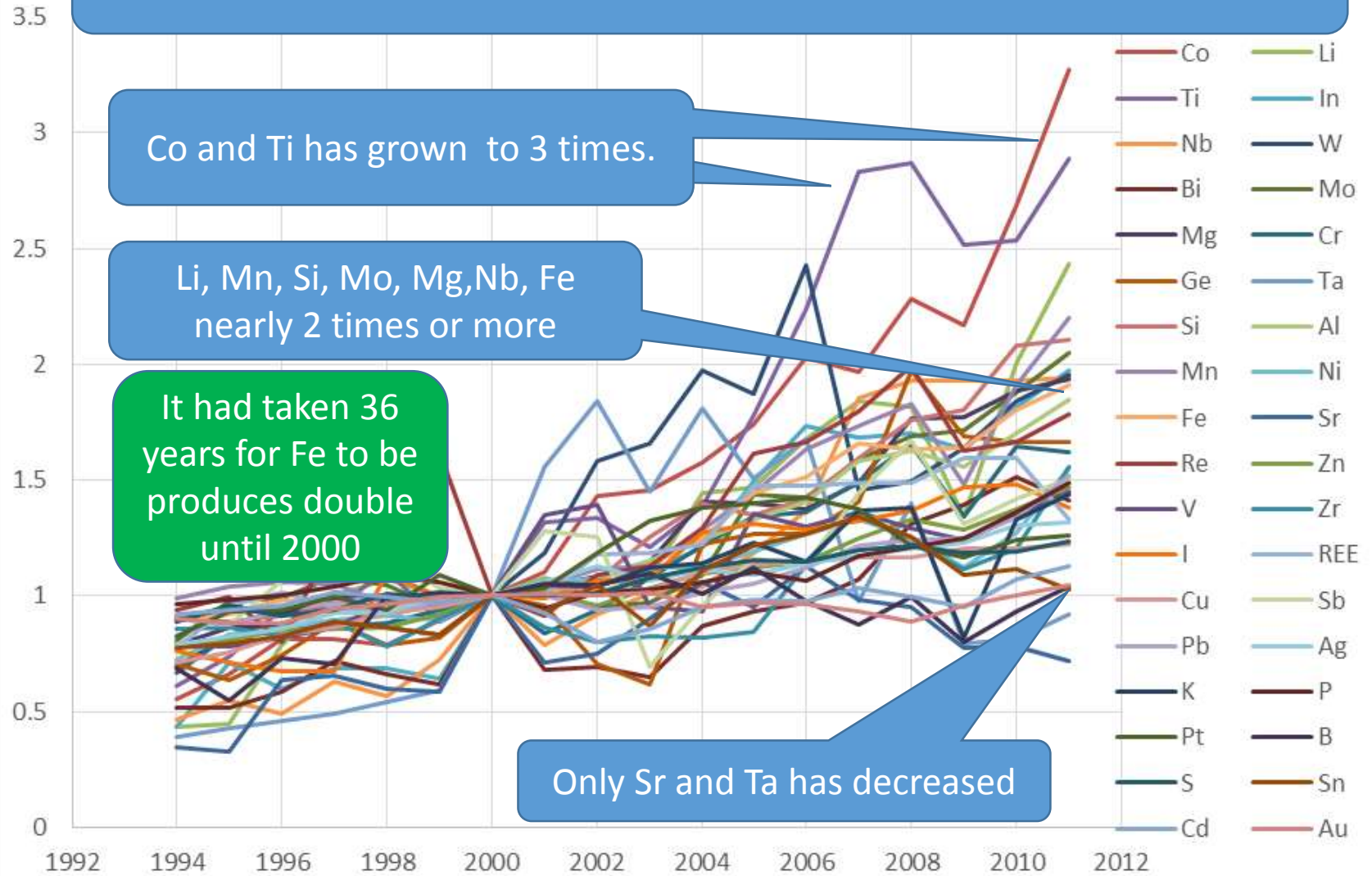
National Institute for Materials Science

World Academy of Ceramics Forum 2016

CeO2



These 15 years was turbulent fifteen years for strategic elements



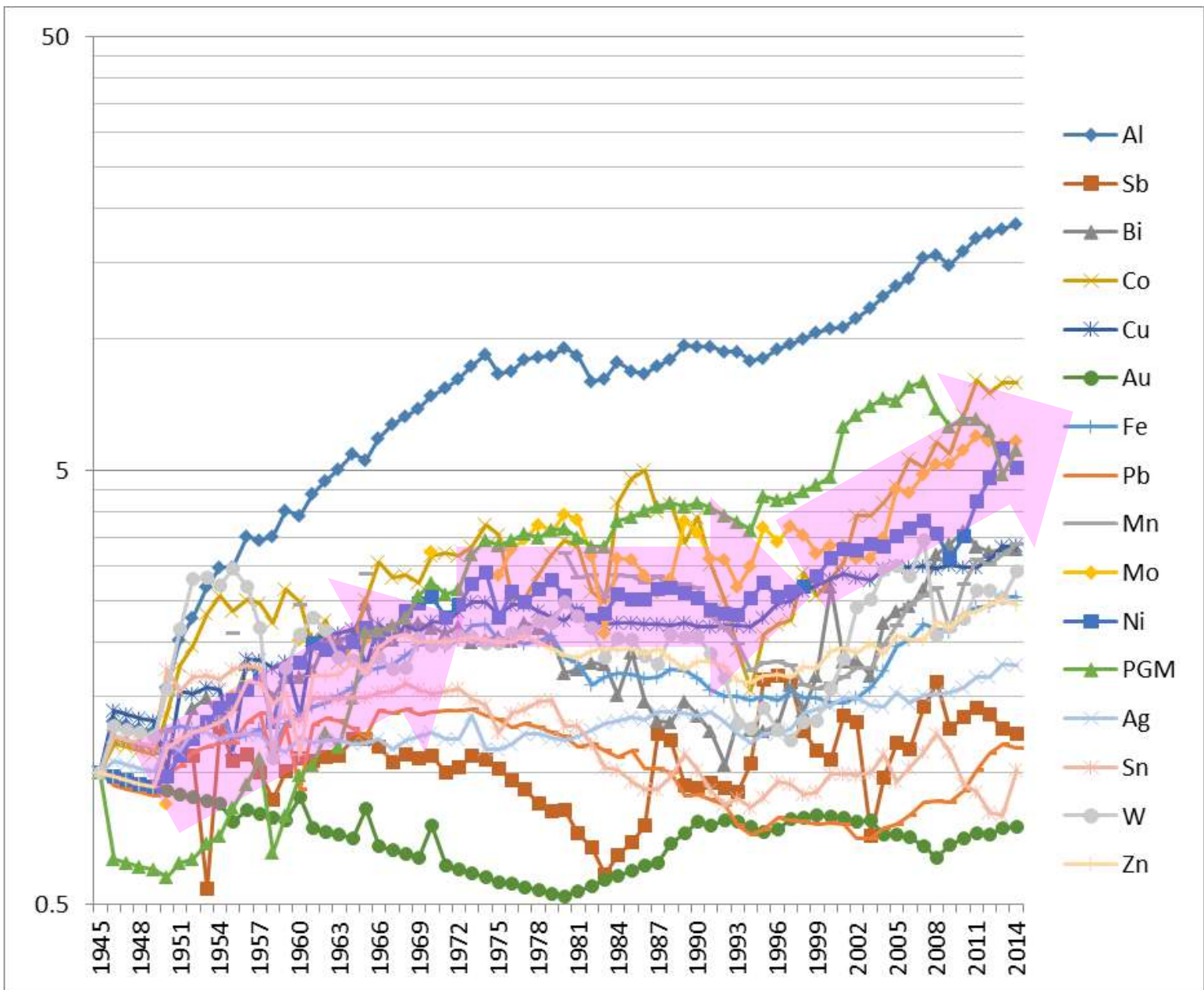


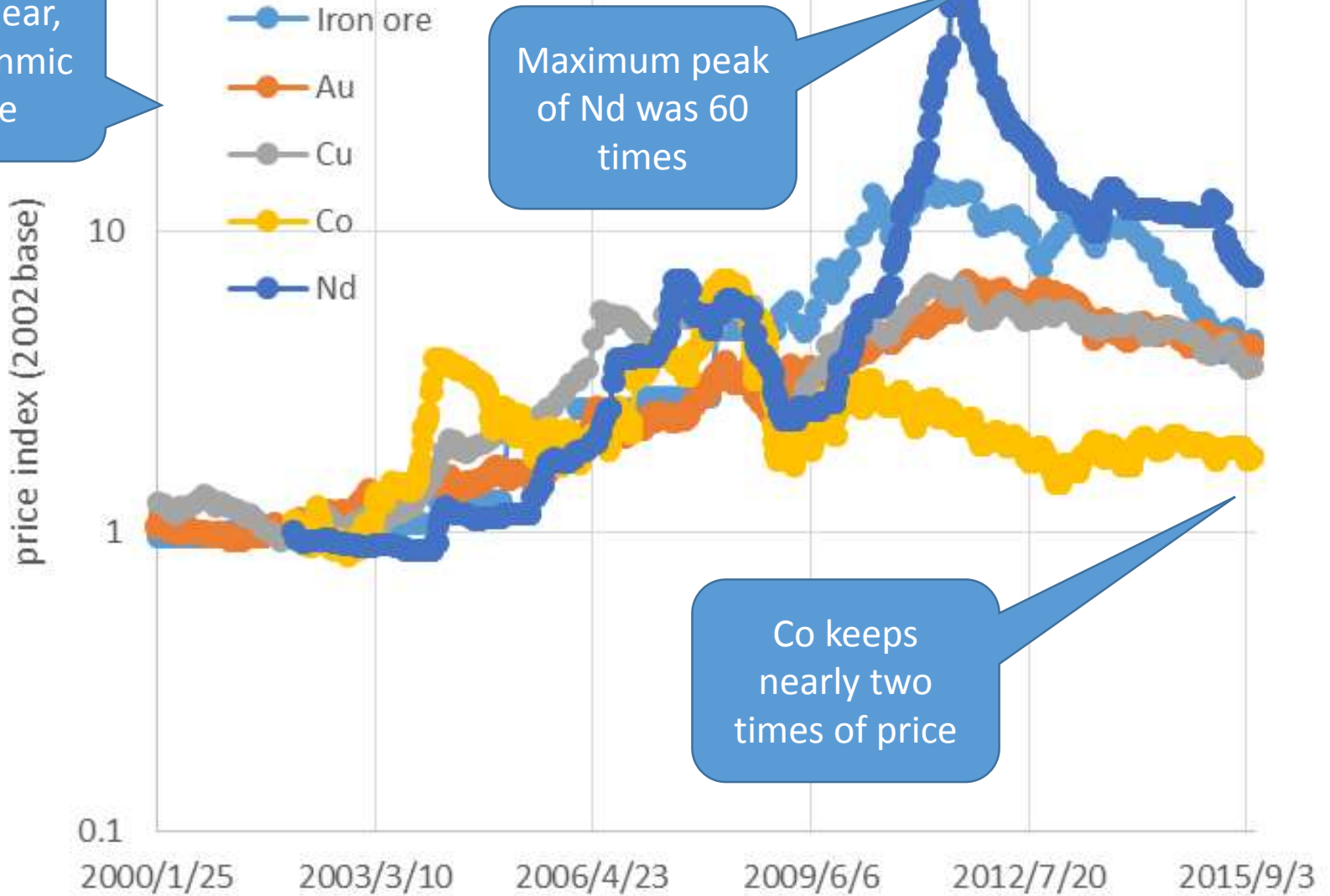
Fig.1 meta production index (1945base)

Prices have changed more drastically

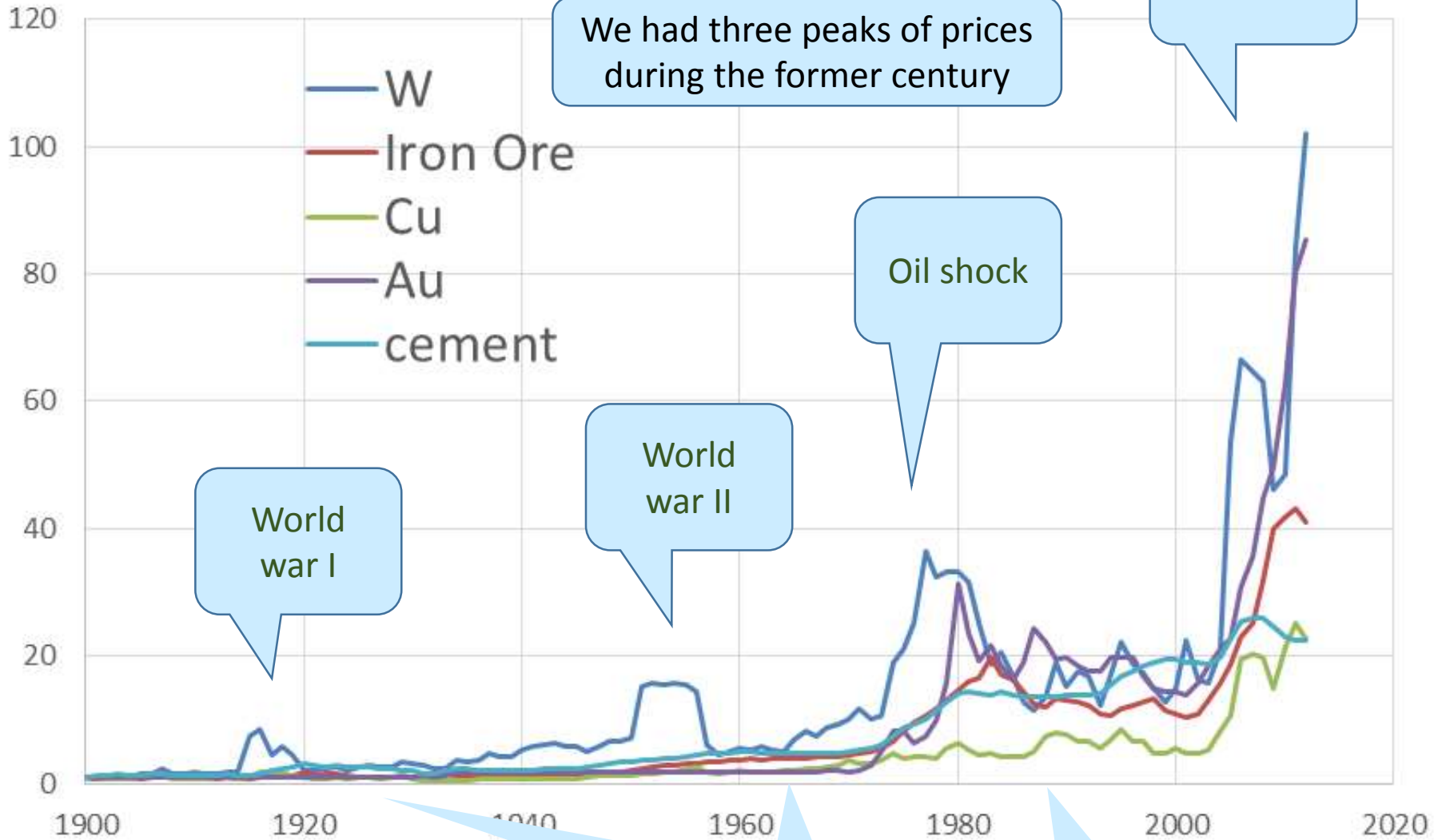
Not linear,
Logarithmic
scale

Maximum peak
of Nd was 60
times

























Co keeps
nearly two
times of price

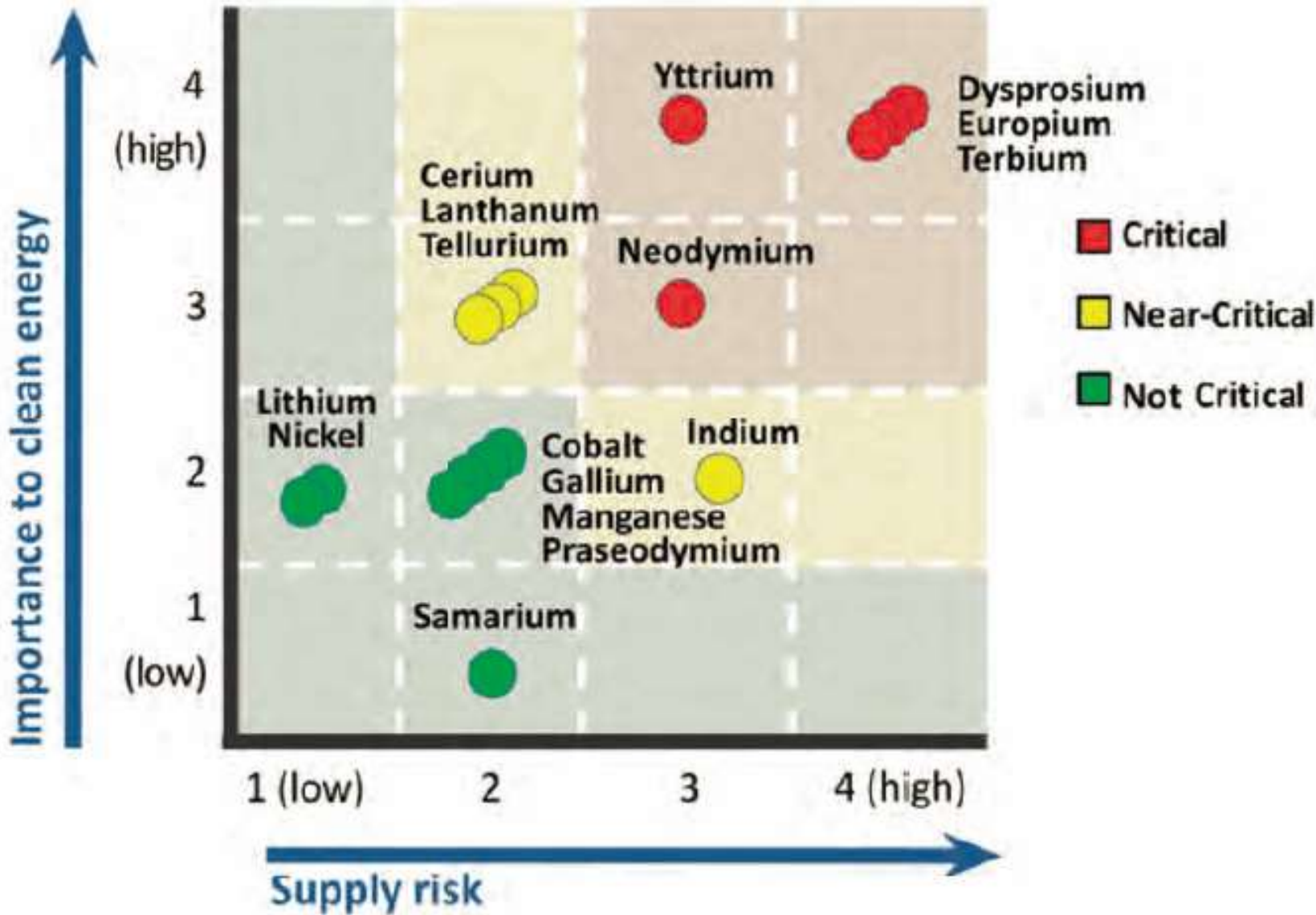


Historical resource price from 1900



After the peak, prices shifted higher levels

H	white: strategic rare metals (JP)															He						
 Li リチウム	Be ベリウム	 Key materials , DOE(US)															B ホウ素	C	N	O	F	Ne
Na	Mg	 Critical materials (EU)															Al	Si	P	S	Cl	Ar
K	Ca	 Sc スカンジウム	Ti チタン	V バナジウム	Cr クロム	 Mn マンガン	Fe	 Co コバルト	 Ni ニッケル	Cu	Zn	 Ga ガリウム	 Ge ゲルマニウム	As	 Se セレン	Br	Kr					
Rb ルビウム	Sr ストロンチウム	 Y イットリウム	Zr ジルコニウム	Nb ニオブ	Mo モリブデン	Tc	 Ru	Rh	 Pd パラジウム	Ag	Cd	 In インジウム	Sn	Sb アンチモン	 Te テルル	I	Xe					
Cs セシウム	Ba バリウム	(Ln) ランタノイド	Hf ハフニウム	Ta タンタル	W タングステン	Re レニウム	Os	Ir	 Pt 白金	Au	Hg	Tl タリウム	Pb	Bi ビスマス	Po	At	Rn					
Fr	Ra	(An)	 La ランタン	 Ce セリウム	 Pr プラセチウム	 Nd ネオジム	Pm	 Sm サマリウム	 Eu ユーロピウム	Gd ガドリウム	 Tb テルビウム	 Dy ジスプロシウム	Ho ホルミウム	Er エルビウム	Tm ツリウム	Yb イットルビウム	Lu ルテチウム					
Ac	Th	Pa	U																			

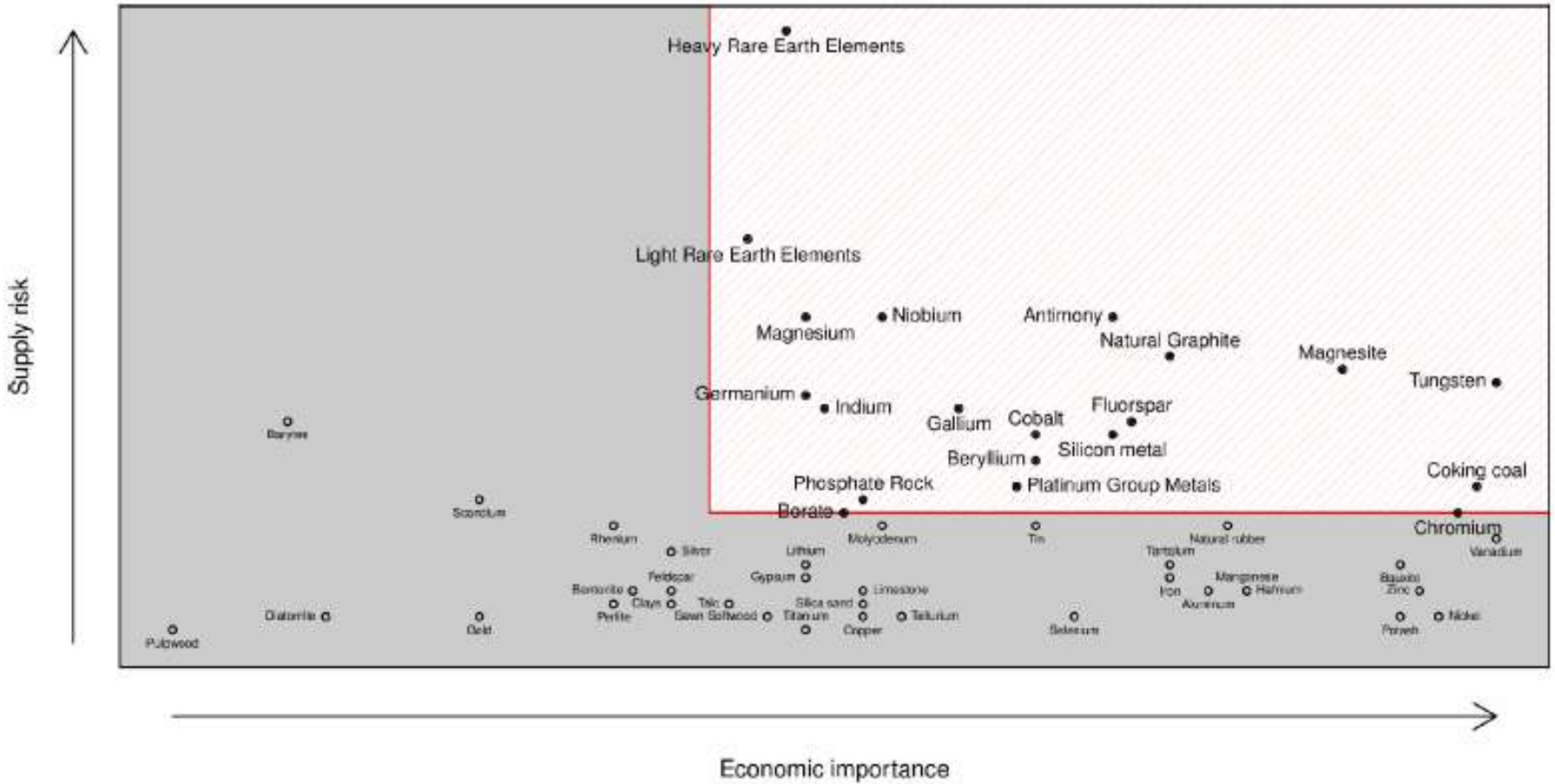


(U.S. DEPARTMENT OF ENERGY: "Critical Materials Strategy" (Dec. 2011))

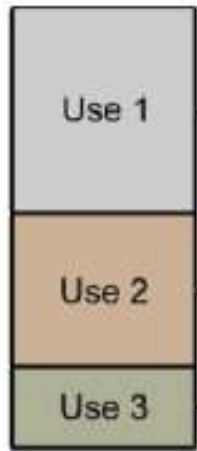
DOE's evaluation method

		Weight: 0.75			0.25		Weight: 0.4		0.1	0.2	0.1	0.2
	Atomic #	Importance to Clean Energy (Rounded Score)	Clean Energy Demand	Substitutability Limitations	Supply Risk (Rounded Score)	Basic Availability	Competing Technology Demand	Political, Regulatory, and Social Factors	Co-Dependence with Other Markets	Producer Diversity		
Short Term												
	lithium	3	2	2	2	1	1	1	1	1	1	
	manganese	25	2	2	2	2	2	1	1	1	2	
	cobalt	27	2	2	2	2	1	2	3	2	2	
	nickel	28	2	2	2	1	1	2	1	1	1	
	gallium	31	2	2	2	2	2	2	1	3	1	
	yttrium	39	4	4	4	3	3	2	4	2	4	
	indium	49	2	2	2	3	4	3	1	3	1	
	tellurium	52	3	3	2	2	2	2	1	3	1	
	lanthanum	57	3	3	3	2	2	2	3	2	3	
	cerium	58	3	3	2	2	2	2	3	2	3	
	praseodymium	59	2	2	1	2	2	1	3	3	3	
	neodymium	60	3	3	3	3	2	3	3	2	4	
	samarium	62	1	1	1	2	2	1	3	3	3	
	europium	63	4	4	4	4	4	2	4	3	4	
	terbium	65	4	4	4	4	4	2	4	4	4	
	dysprosium	66	4	4	3	4	4	2	4	3	4	
	Atomic #	Importance to Clean Energy (Rounded Score)	Clean Energy Demand	Substitutability Limitations	Supply Risk (Rounded Score)	Basic Availability	Competing Technology Demand	Political, Regulatory, and Social Factors	Co-Dependence with Other Markets	Producer Diversity		
Medium Term												
	lithium	3	3	3	2	2	2	2	1	1	1	
	manganese	25	2	2	2	2	2	1	1	1	2	
	cobalt	27	2	2	2	2	1	2	2	2	2	
	nickel	28	2	2	2	1	1	2	1	1	1	
	gallium	31	2	2	2	2	3	3	1	3	1	
	yttrium	39	3	3	4	3	3	2	4	2	4	
	indium	49	2	2	2	2	3	3	1	3	1	

(REPORT ON CRITICAL RAW MATERIALS FOR THE EU 2014)



End uses of a raw material



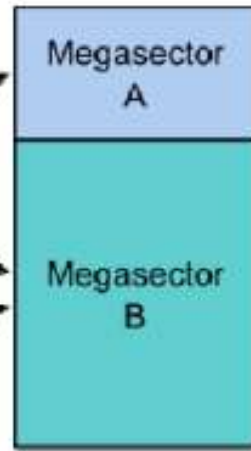
$\Sigma = 100\%$

Assign end uses to pertinent "megasectors"

A_{is}

Gross value added of end use "megasectors" in the EU

Q_s



$$El_i = \frac{1}{GDP} \sum_s A_{is} Q_s$$

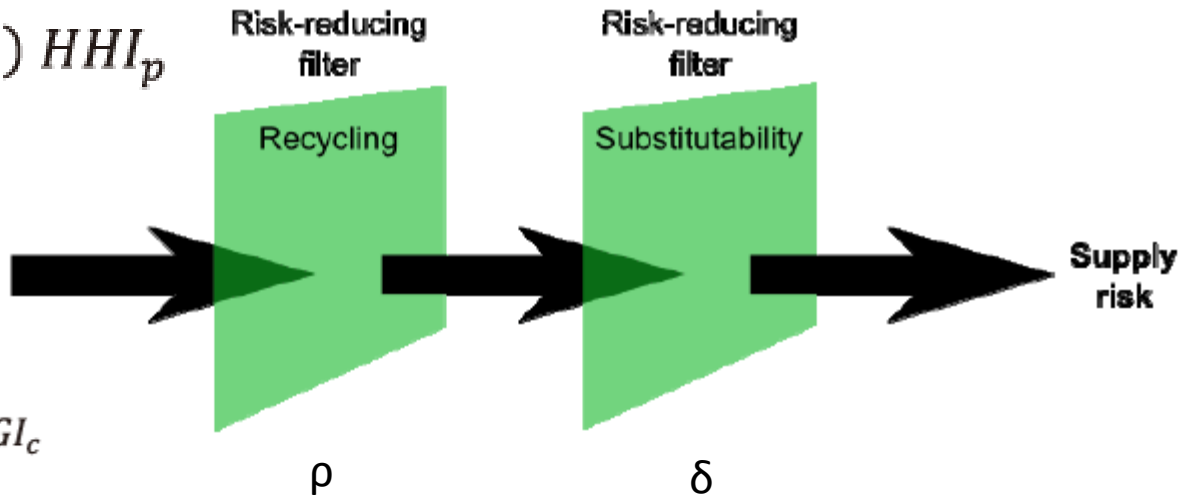


Economic importance

Multiply each % end use by GVA of megasector and build weighted sum

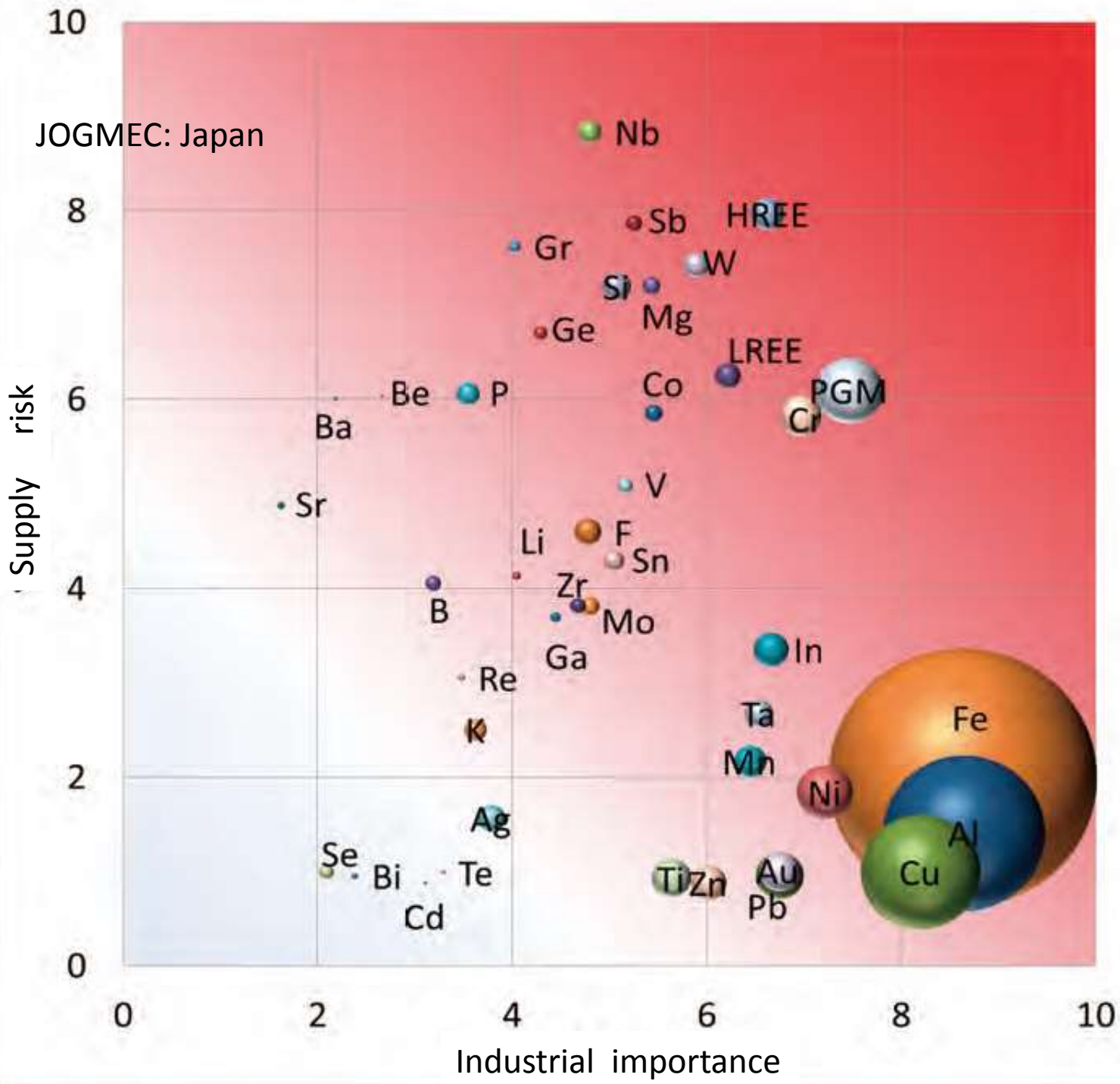
$$SR_i = \delta_i (1 - \rho_i) HHI_p$$

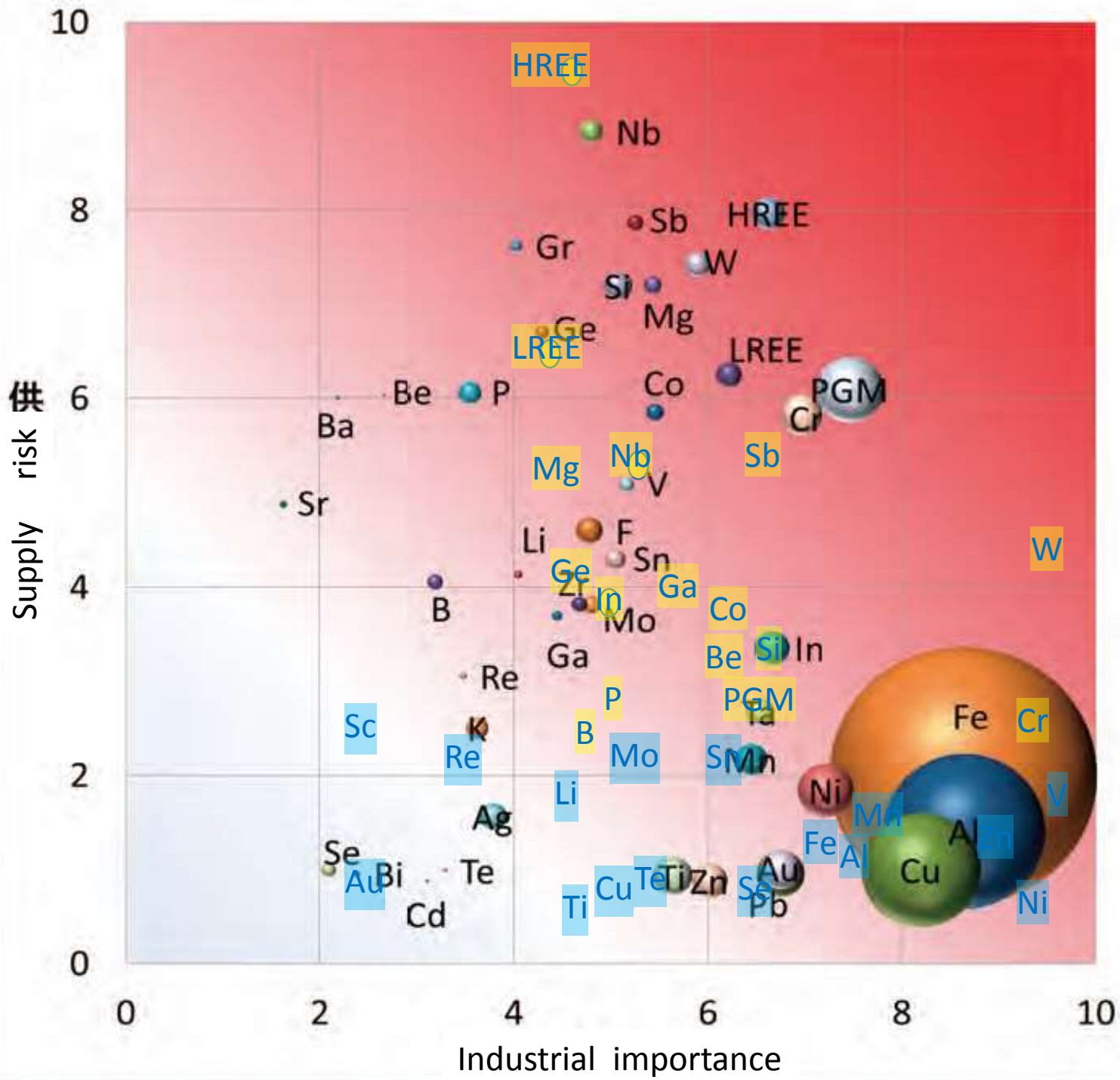
Risk from concentrated primary production in countries with poor governance



$$HHI_p = \sum_c (S_{ic})^2 WGI_c$$

WGI: world governance indicator
S : share





4 types + 1 of resource constraints

Absolute mass Reserve amount is not so enough.
Au, Cu, Zn

Geological, Political unevenly distributed
Pt, Nb, Dy Co

Energy depending on energy situation
Al, Mg, Si

Environmental cost of waste management
REE

plus1 speed of supply
by-products such as LI

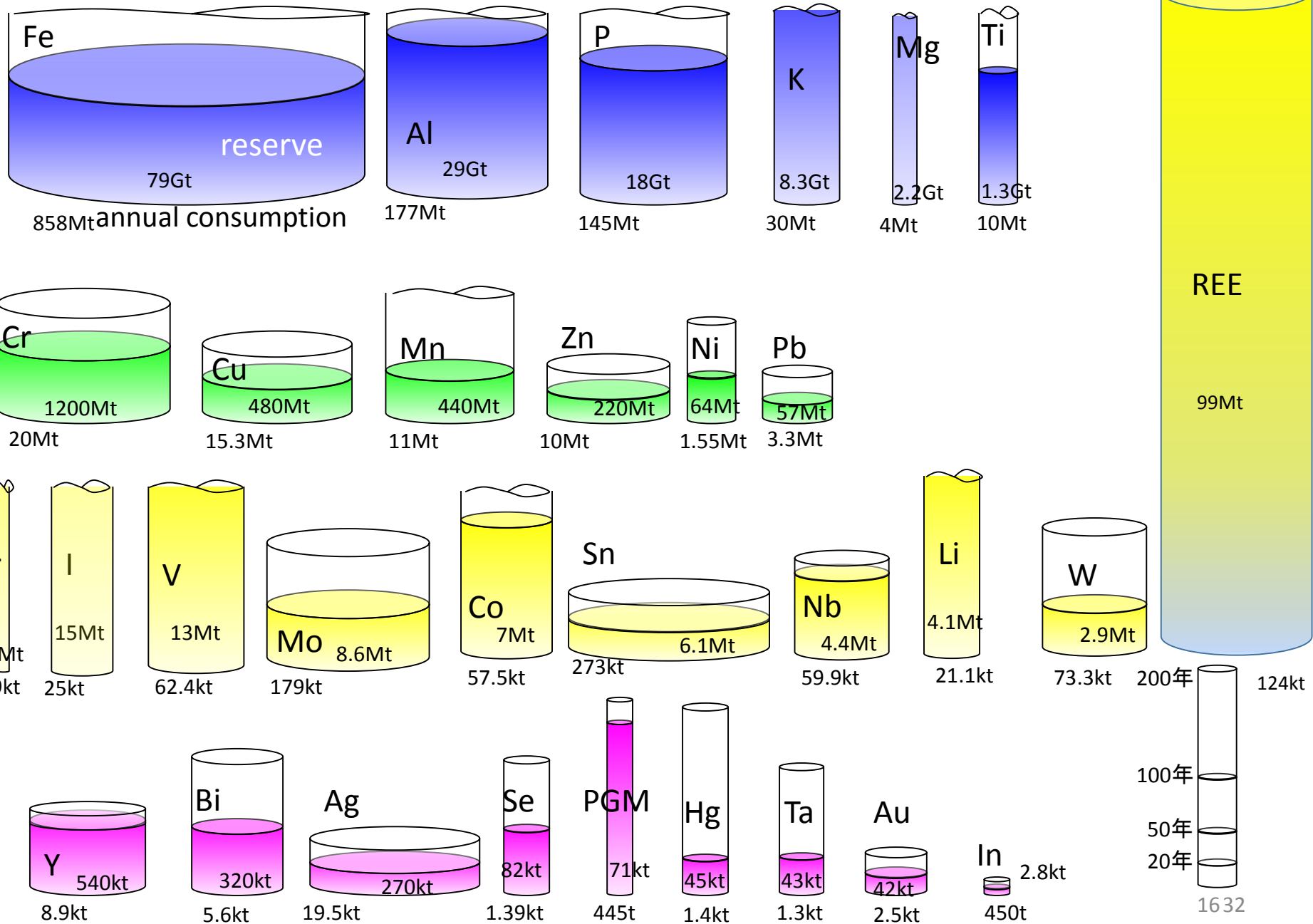
The Elements with sustainability parameters

- Durable years: (reserve)/(annual consumption)
- Resource-view weight: tons of TMR for 1kg of metal production
- Share % Of top country of production, country code
- Increase of production from 1999 to 2009, (%)

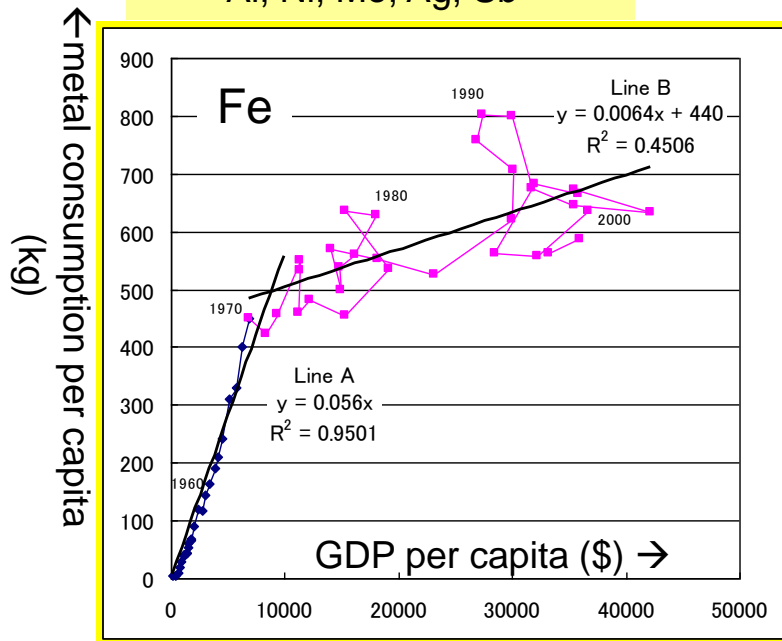
H scarcity TMR domination acceleration												B 0.14 47TK 101						He	
Li 194 1.5 41CL 120		Be 2.5 86US 42		● Magnet, motor ● Optical function ● Display & its porishing ● Batteries ● Information media ● Fire retardant ● IC tips and parts ● Thermoelectric, ● Solar cell ● Electric wiring ● Catalyst, electrode ● lightning ● Structural material										Al 164 0.05 31CN 163	Si 0.03 65CN 169	P 124 35CN 114	S 126	Cl 130	Ar
Na 0.4 56 100	Mg 5500 0.07 82CN 215	K 2800 26CA 99	Ca 0.09 237	Sc 2.	Ti 1300 0.04 23AU 220	V 208 1.5 37CN 135	Cr 60 0.03 42ZA 180	Mn 40 0.01 22CN 163	Fe 92 0.008 39CN 165	Co 122 0.61 40CG 219	Ni 41 0.26 19RU 125	Cu 31 0.36 34CL 125	Zn 22 0.04 28CN 131	Ga 7.3 157	Ge 32 71CN 241	As 0.03 47 129	Se 59 0.45 50JP 119	Br 38IL 86	Kr
Rb 0.13	Sr 0.51 48ES 133	Y6 1 2.7 271	Zr 4200 0.55 41AU 151	Nb 73 0.64 92BR 335	Mo 48 0.75 25US 155	Tc	Ru 79 79ZA 119	Rh 160 2300 79ZA 85	Pd 160 810 41ZA 156	Ag 14 4.8 18PL 134	Cd 0.07 23CN 94	In 24 12 50CN 250	Sn 22 2.5 37CN 153	Sb 0.06 91CN 136	Te 10 44JP 88	I 600 59CL 159	Xe		
Cs 0.01	Ba 31 0.51 147	(Ln) 800 - 97CN 162	Hf 10 151	Ta 33 6.8 48AU 245	W 40 0.2 81CN 185	Re 18 48CL 118	Os 540 79ZA	Ir 400 79ZA 40	Pt 160 530 79ZA 118	Au 17 1100 13CN 101	Hg 32 2 63CN 56	Tl 0.4 67	Pb 17 0.03 43CN 128	Bi 57 0.22 62CN 221	Po	At	Rn		
Fr	Ra	(An)	La 1600 8.2 371*	Ce 770 18 295*	Pr 7.9	Nd 420 12 90*	Pm	Sm 16	Eu 188 33	Gd 17	Tb 244 55	Dy 209 16	Ho 30	Er 12	Tm 32	Yb 32	Lu 32		
Ac	Th	Pa	U 22																



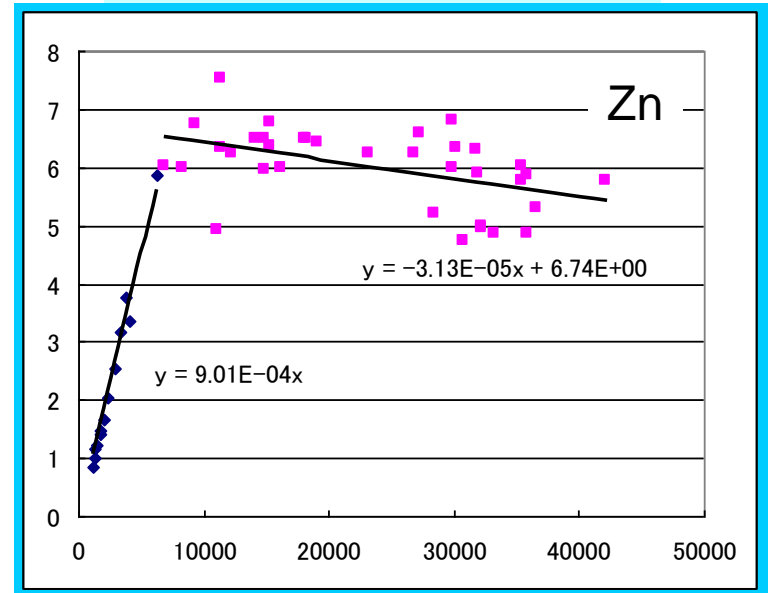
* Estimated by import of Japan, () amount in crust is less than in sea water
 Data form 米国鉱山局データ USGS minerals information
 工業レアメタル (Kogyo rare metal) Japanese journal
 「概説 資源端重量」 NIMS-EMC data on mat. & env. No.18
 Halada, Katagiri, Proc. of EcoBalance 2010 p609



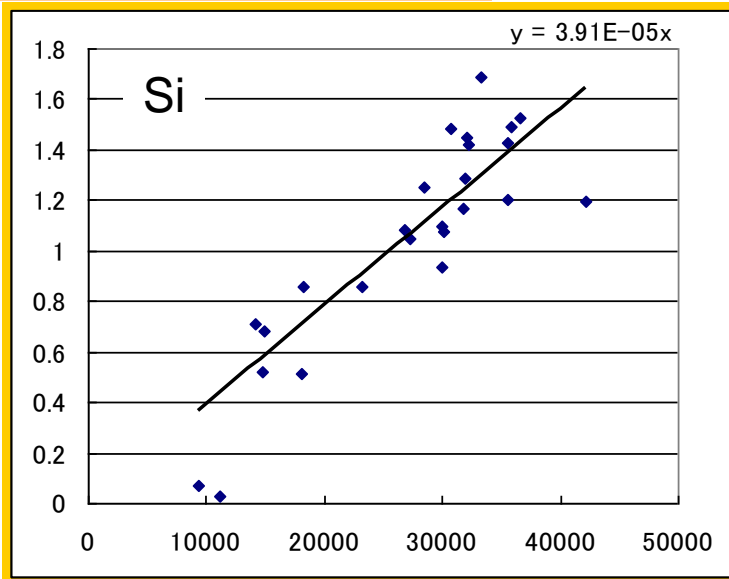
Fe-type: weakly de-coupled
Al, Ni, Mo, Ag, Sb



Zn-type: de-coupled
Cu, Sn, Pb, W, Cr, Mn, Au

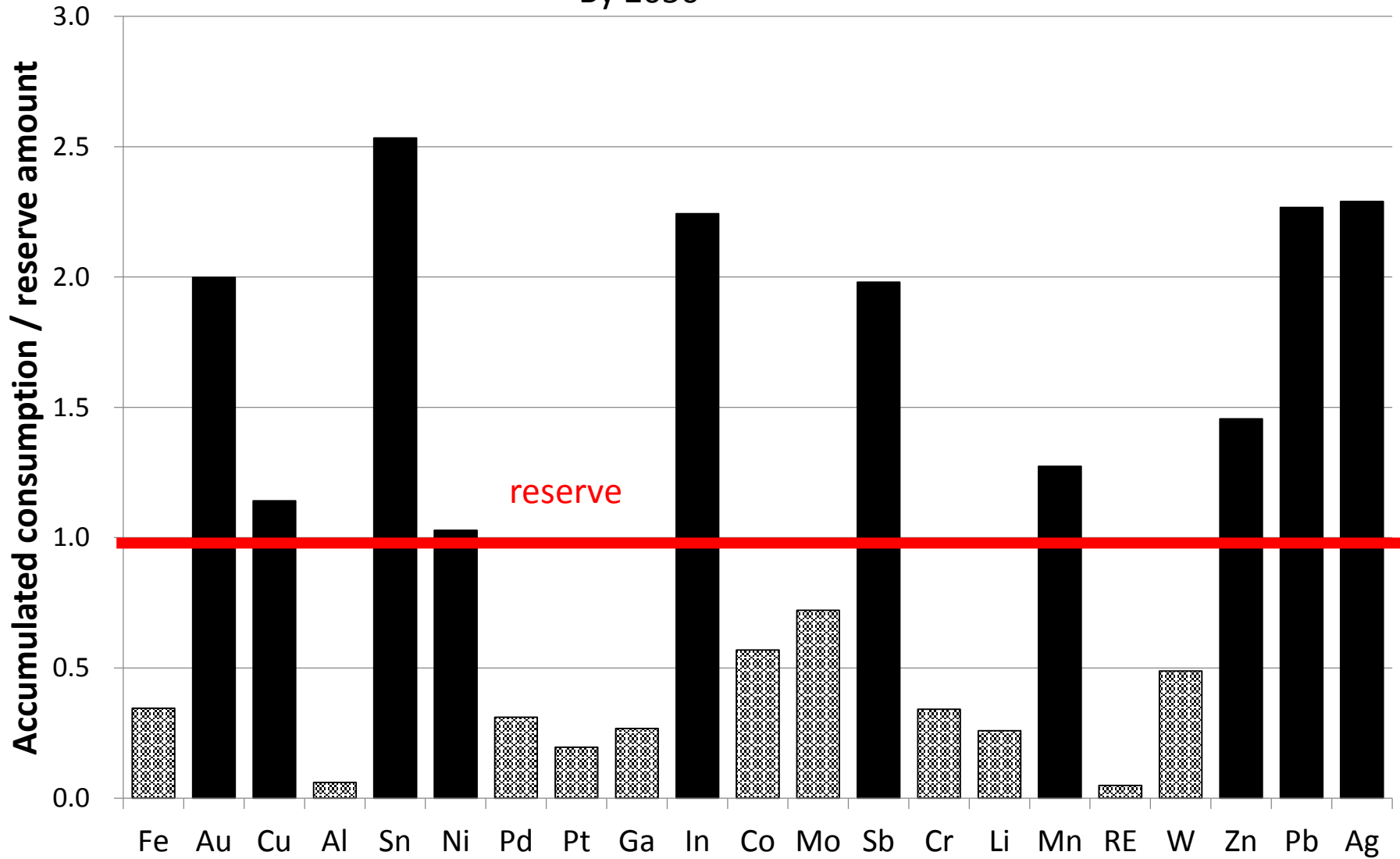


Si-type: still coupling Pt,



Four types of the two step line model of metal consumption v.s. GDP per capita

By 2050



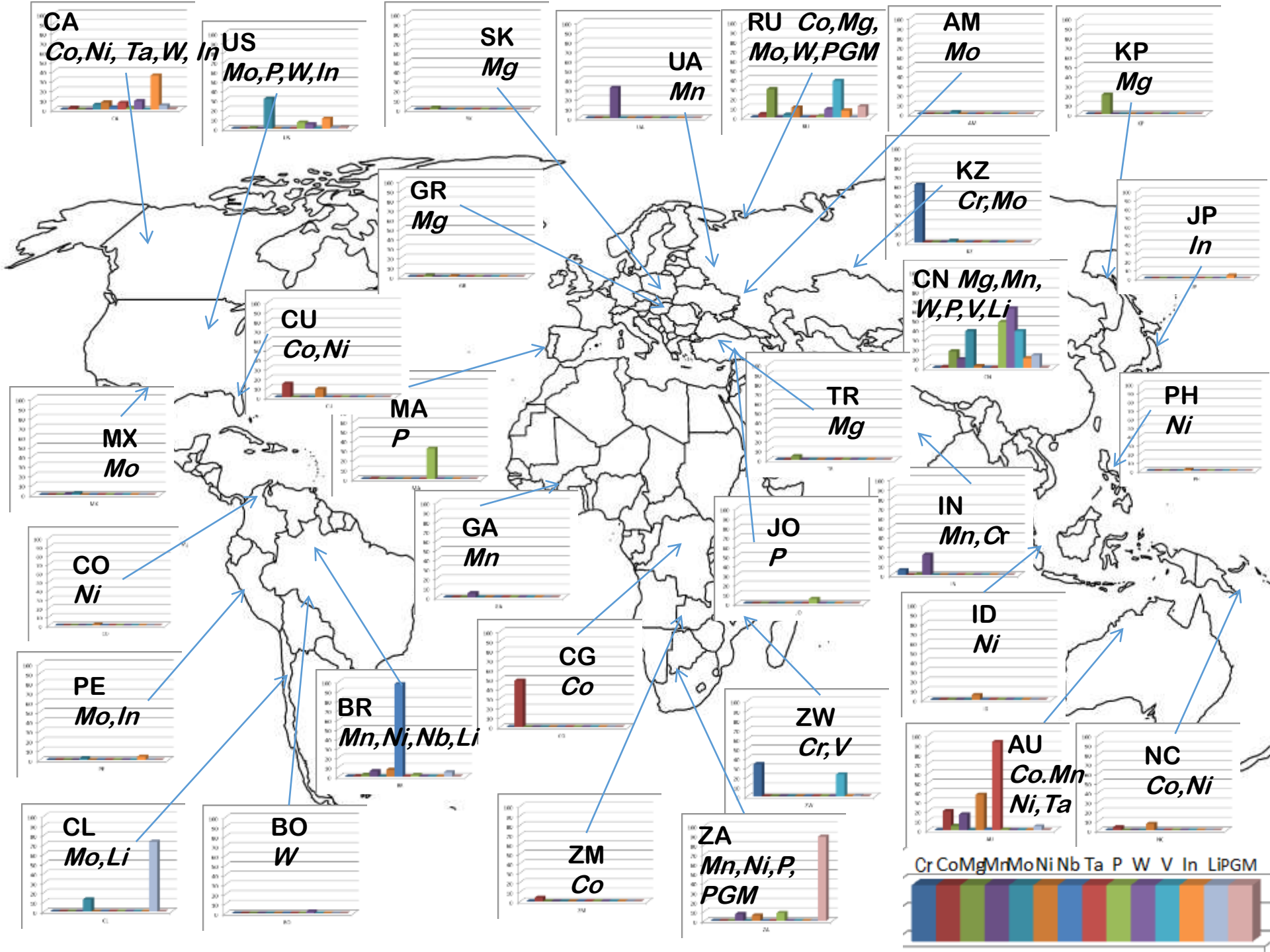
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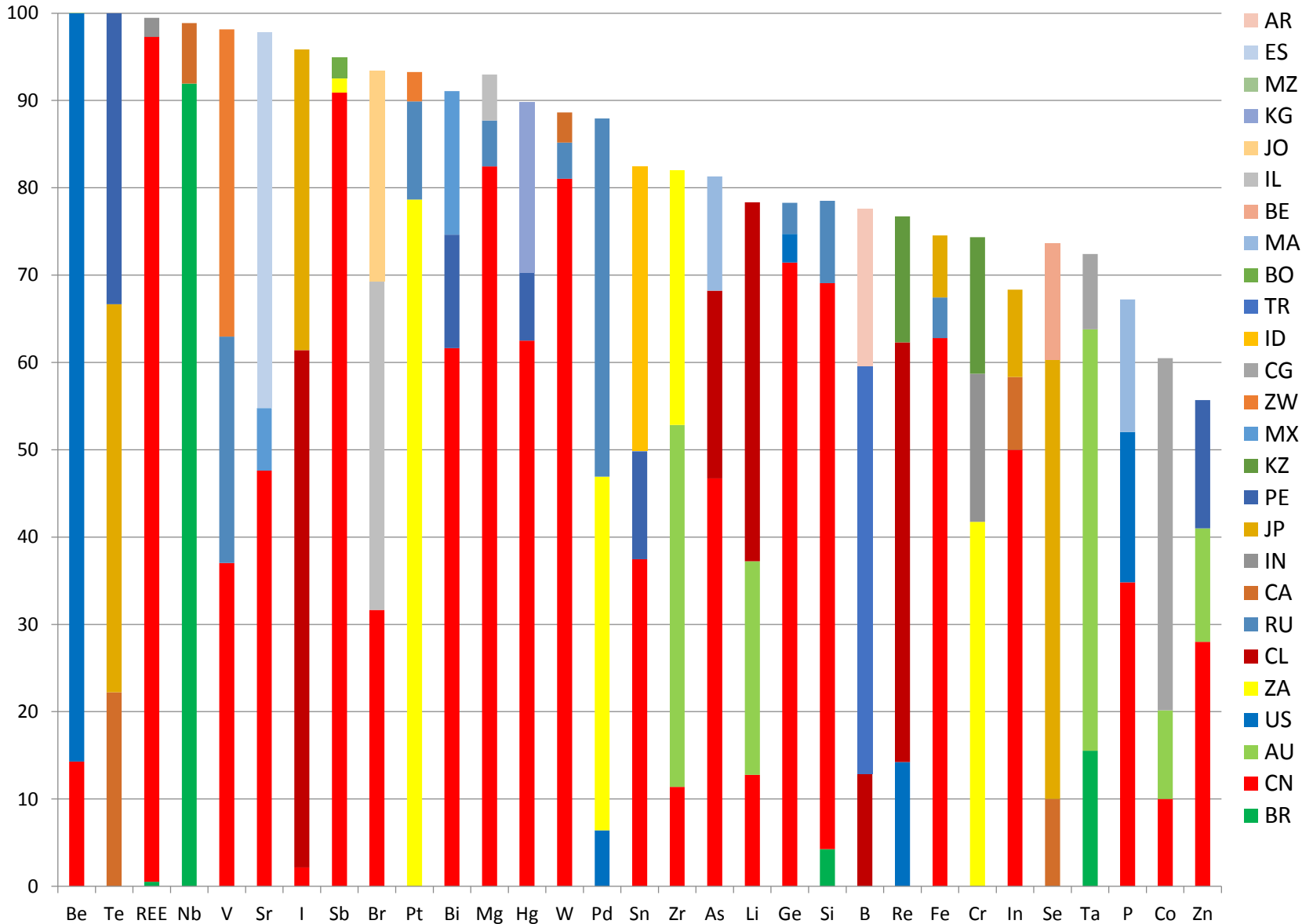
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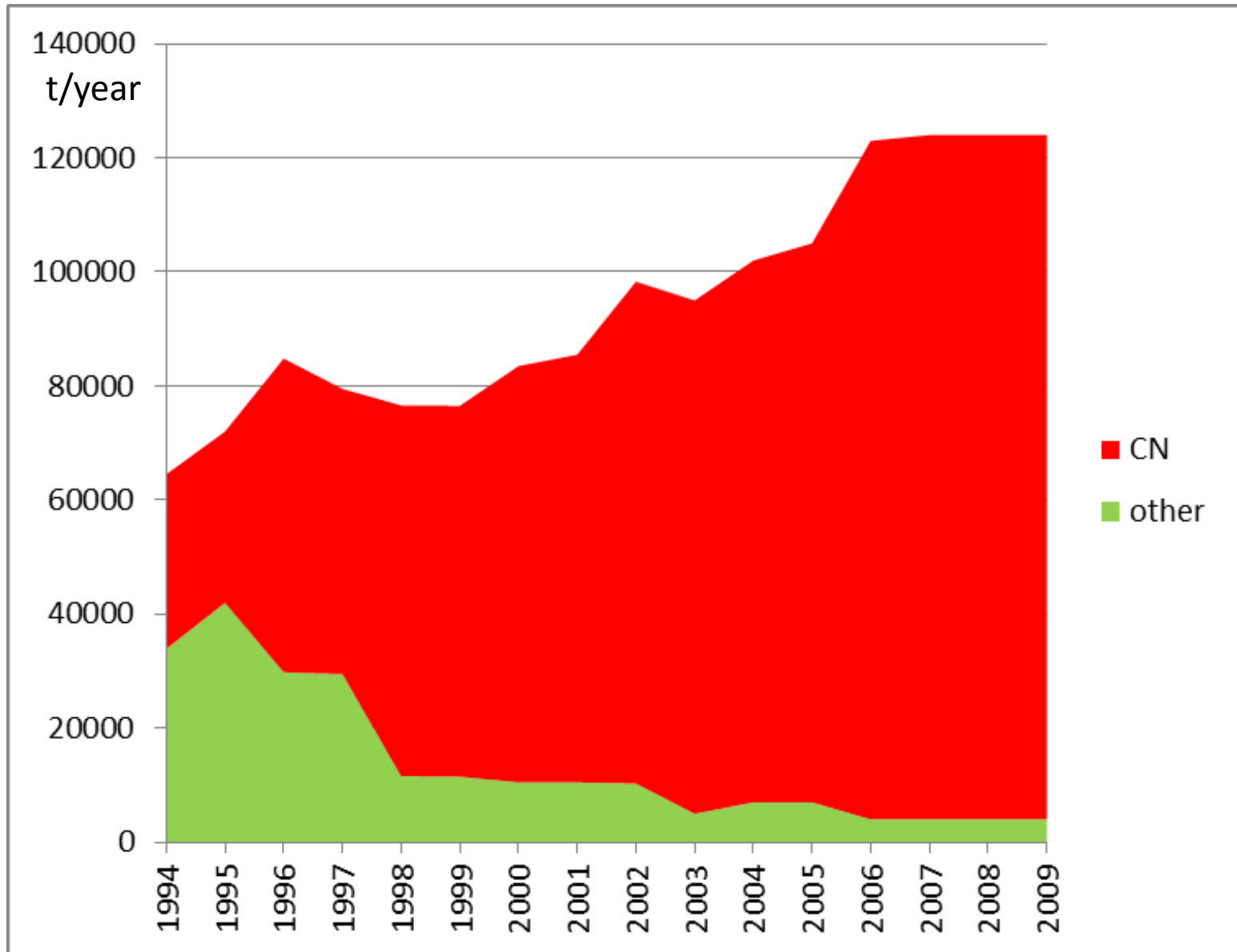
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Data form 米国鉱山局データ USGS minerals information
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Three top country produce more than 80% In 18 elements in 31





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- Magnet, motor
- Batteries
- IC tips and parts
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- lightning
- Optical function
- Information media
- Thermolectric,
- Catalyst, electrode
- Structural material
- Display & its porishing
- Fire retardant
- Solar cell

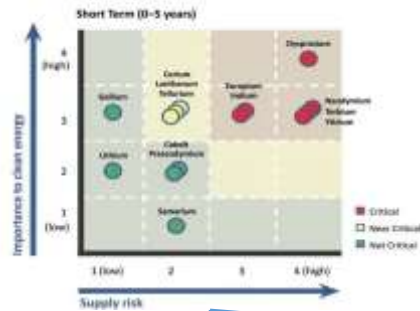


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Progress of discussion on criticality index of metals

Department of Energy; Criticality Matrix

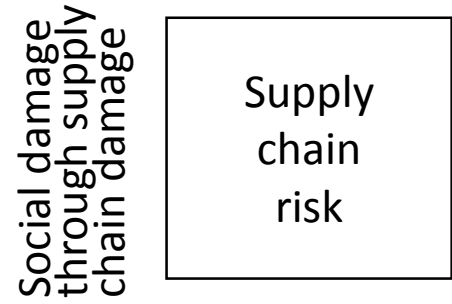
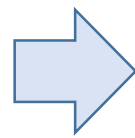
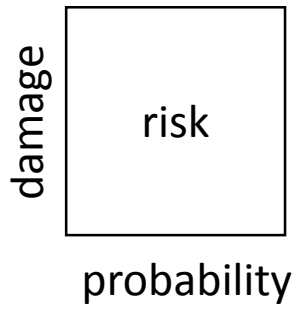


DOE matrix
(importance)
x (supply risk)

Criticality has Different two concepts

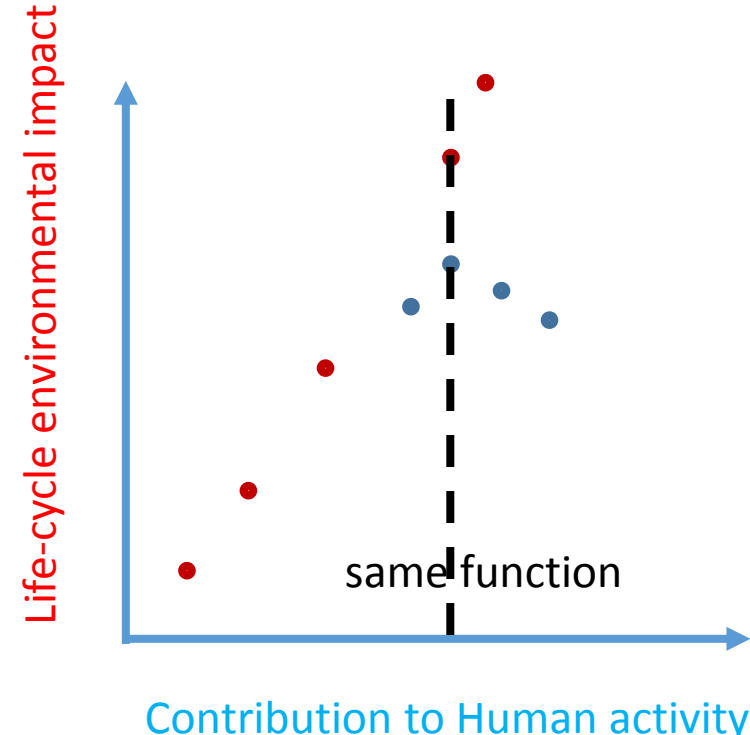
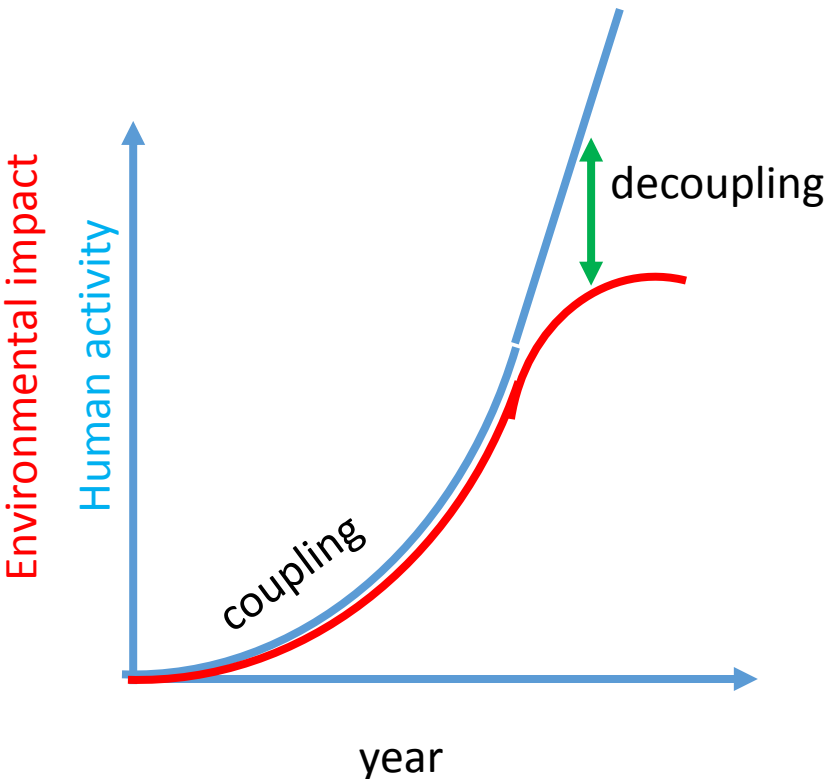
Criticality for supply chain
= supply chain risk

Criticality for global sustainability

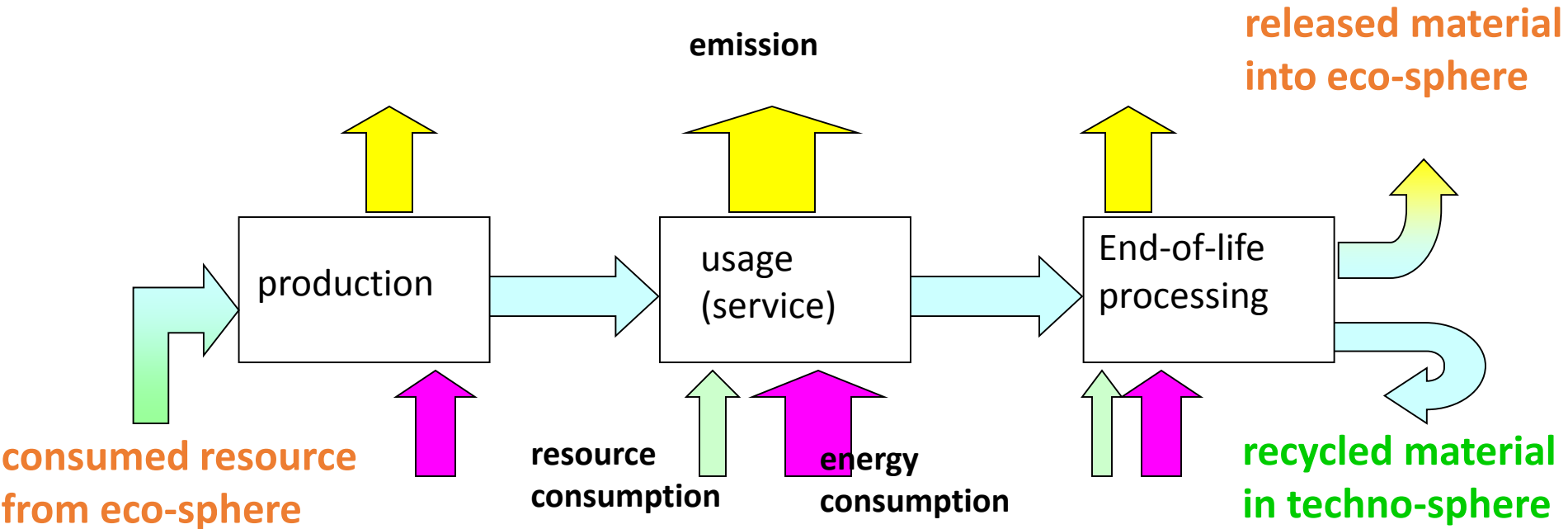


Global environment

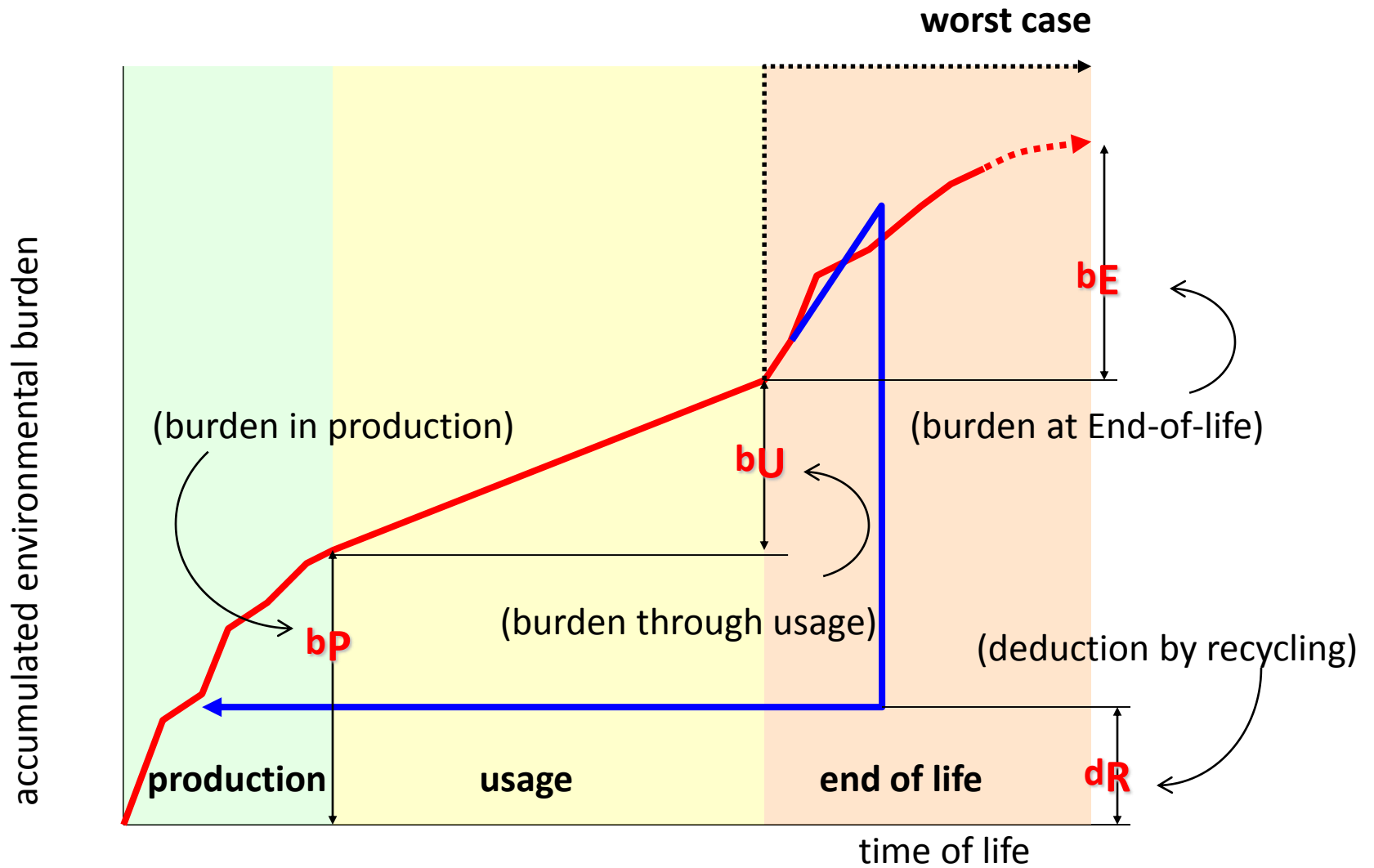


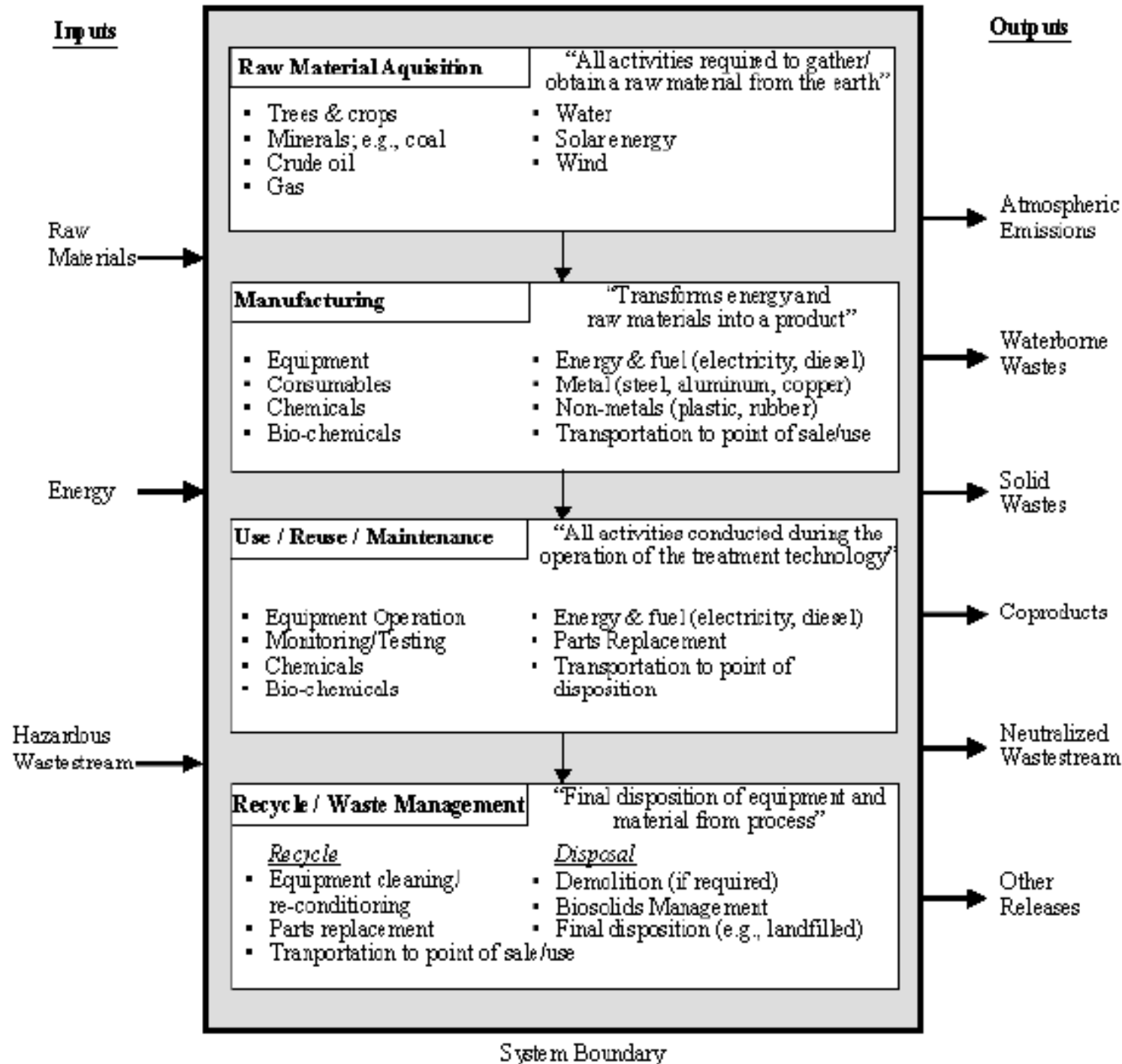


environmental burden = (emission and release of material)
+ (consumption of material and energy)



life-cycle environmental burden =
burden in production + burden through usage
+ burden of End-of-Life - deduction by recycling





In Toyama framework (G7 summit Ise-Shima)

Policy Guidance on Resource Efficiency



Key recommendations

Going for green growth and establishing a resource efficient economy is a major environmental, development and economic challenge today. In this context, improving resource productivity and putting in place policies that implement the principles of reduce, reuse, recycle (the 3Rs) is crucial, as recognised by G7 Leaders in the Schloss Elmau's declaration in June 2015. This report responds to the request by G7 Leaders at the Schloss Elmau Summit asking the OECD to develop policy guidance for resource efficiency. Key findings and recommendations from this report include the following considerations.

Although resource efficiency is first and foremost a matter of national policy decisions, only collective action and co-ordinated efforts will ensure widespread benefits amongst countries. The G7 has an important role to play in this regard.

The G7 can highlight best practices and provide a platform for sharing of experiences both within and beyond its membership. Two key messages from this Guidance are that:

- Resource efficiency policies should target the **entire life-cycle of products**.
- National policies should put more emphasis on aligning sectoral policies in diverse areas like innovation, investment, trade, education and skills development with resource efficiency objectives.

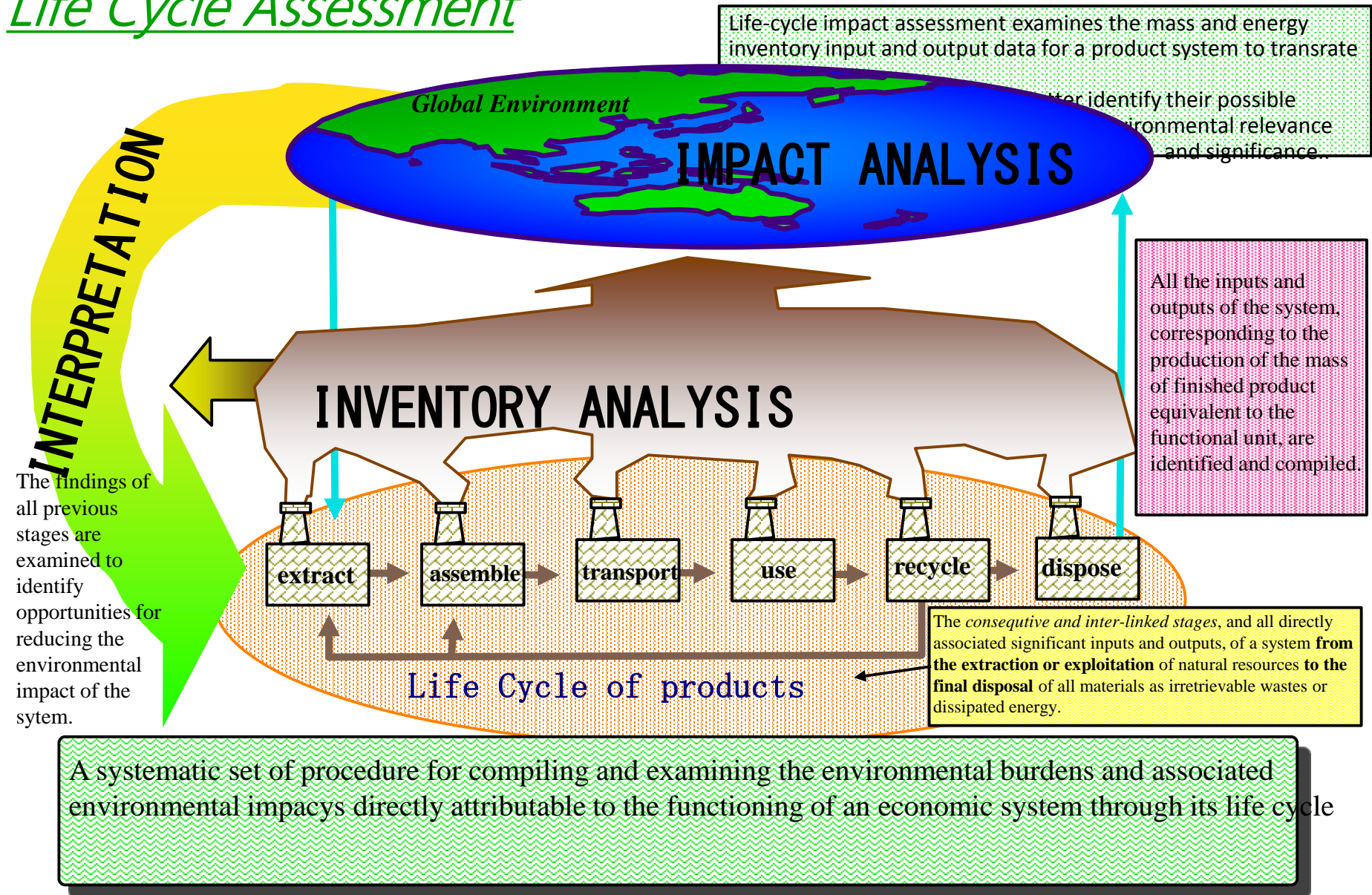
These broader messages on the life-cycle approach and policy coherence could be explicitly supported by the G7.

The G7 can also strengthen co-ordination and co-operation at the international level by:

- Facilitating integration of resource efficiency considerations in Global Value Chains by supporting businesses in their supply chain management efforts.
- Addressing trade and investment related obstacles to resource efficiency in supply chains, including export restrictions on secondary raw materials, restrictions on trade in used products, and barriers to trade in environmental goods and services.
- Calling for some degree of harmonisation in the growing field of environmental labelling and information schemes, with the aim of maintaining high standards, allowing for increased mutual recognition of schemes, and countering increased costs associated with scheme multiplication across international markets.

Finally, the G7 can help address key information gaps related to material flows and resource efficiency. These gaps include harmonised data on indirect material flows associated with international trade, information on flows of secondary raw materials, disaggregated information on resource use by industry, and information on the quality and deterioration of natural resource stocks. Similarly, the G7 can support internationally co-ordinated efforts to improve economic analysis of resource efficiency, an area that has currently received very little attention in research.

Life Cycle Assessment



Size: 9" dinner plate

Weight: 0.5681 kg/plate

Functional Unit: 2960 plates

utilization

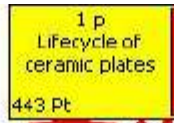
Details	Quantities
Model name	Model 66 DRWSPW-champion company
Capacity	148 racks/hr
Water consumption	302.5298 kg
Detergent dosage	840.64 kg
Electricity consumption	16 MJ

firing

Inputs	Quantity
	Per 2960 plates
Casted, moulded, pressed dough	2228.457312 kg
Natural gas (heat)	35155.4112 MJ
Electricity	7392 MJ
Emissions to air	In kg/2960 plates
Carbon dioxide	2.29×10^3
Carbon monoxide	7.08×10^{-1}
Nitrogen dioxide	4.23×10^0
Sulphur dioxide	2.75×10^0
Hydrocarbons	2.67×10^0
Particulates	2.23×10^{-1}
Aldehydes	2.83×10^{-3}
Organic compounds	5.11×10^{-3}
Ammonia	5.35×10^{-4}
Fluoride	4.36×10^{-2}

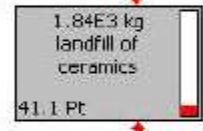
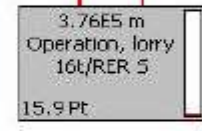
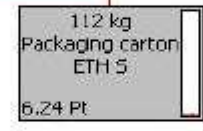
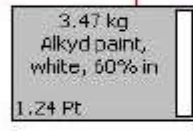
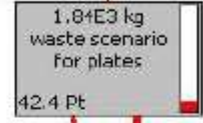
Raw material

Inputs	Quantity
	In kg/2960 plates
Kaolinite	840
Silicon	420
Feldspar	420
Water	3108



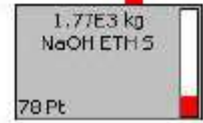
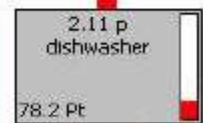
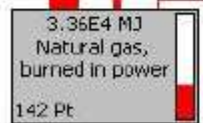
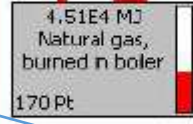
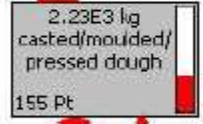
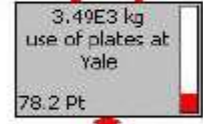
Package & transport

Inputs	Quantity
	In kg/2960 plates
Fired plates	1721.028952
Paint	3.47008
Packaging carton	112
Transportation by truck	346 km



pressing

Inputs	Quantity
	Per 2960 plates
Ceramic dough	4765.6 kg
Natural gas (heat)	9915.6288 MJ
Electricity	26208 MJ
Emissions to air	In kg/2960 plates
Carbon dioxide	6.47×10^2
Carbon monoxide	2.00×10^{-1}
Nitrogen dioxide	1.19×10^0
Sulphur dioxide	7.76×10^{-1}
Hydrocarbons	7.54×10^{-1}
Particulates	6.28×10^{-2}
Aldehydes	7.98×10^{-4}
Organic compounds	1.44×10^{-3}
Ammonia	1.51×10^{-4}
Fluoride	1.23×10^{-2}
Emissions to water	in kg/2960 plates
Suspended substances	2.07×10^{-4}
COD	6.23×10^{-4}
Oils	6.62×10^{-3}
Metallic ions	4.20×10^{-6}
Fluoride	1.86×10^{-3}
Ammonia	8.67×10^{-4}



Life-cycle of ceramic plate

Size: 9" dinner plate

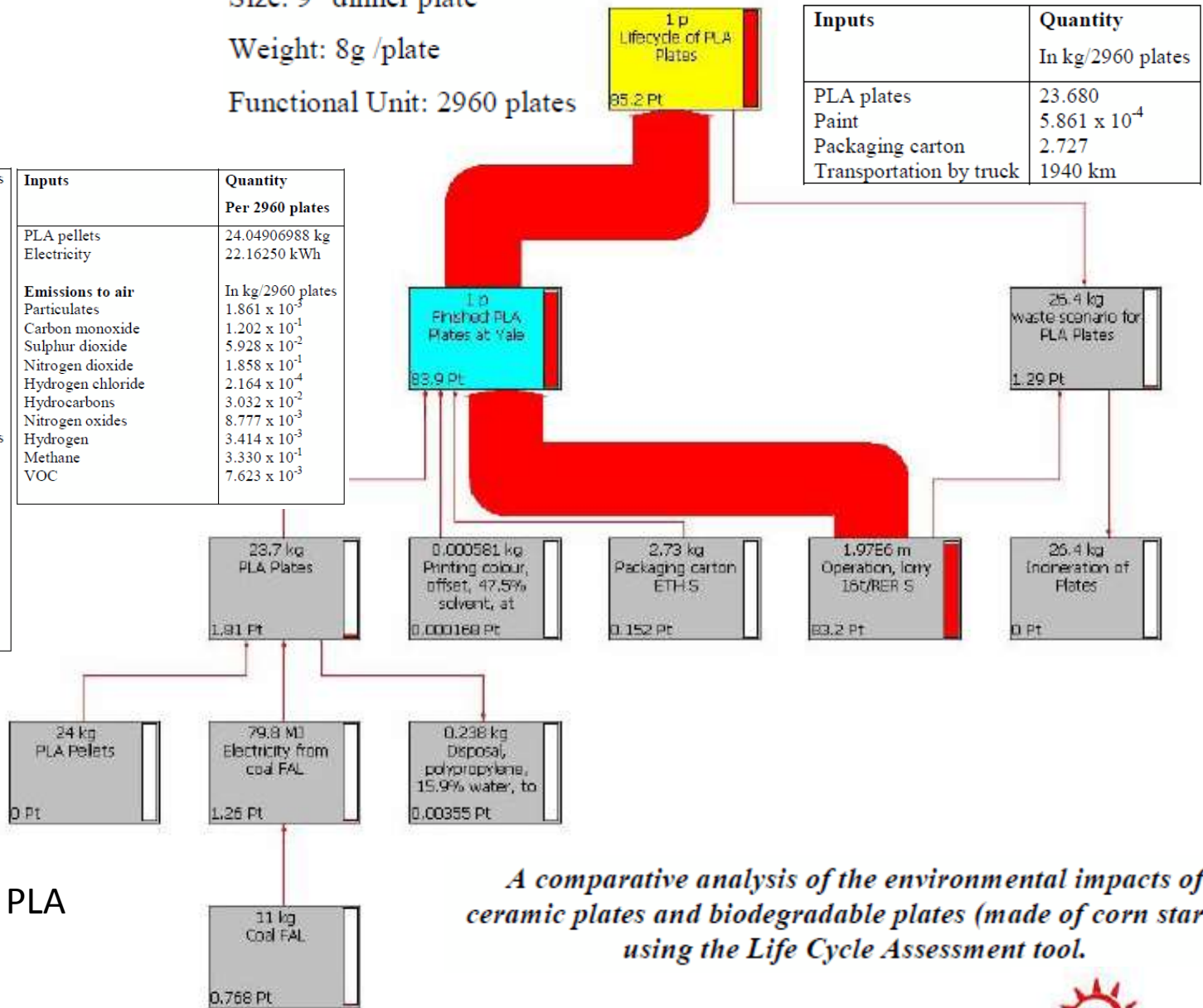
Weight: 8g /plate

Functional Unit: 2960 plates

Inputs	Quantity
	In kg/2960 plates
PLA plates	23.680
Paint	5.861×10^{-4}
Packaging carton	2.727
Transportation by truck	1940 km

Emissions to water	In kg/2960 plates
COD	1.419×10^{-1}
BOD	2.587×10^{-2}
Sodium ion	1.755×10^{-2}
Nitrogen oxides	2.900×10^{-2}
Chloride	3.915×10^{-2}
Suspended solids	7.440×10^{-2}
Phosphorous compounds	2.881×10^{-4}
Nitrogen	2.031×10^{-3}
Sulphate	3.318×10^{-3}
Calcium compounds	3.061×10^{-3}
TOC	3.770×10^{-2}
Calcium ion	5.956×10^{-3}
Solid waste	In kg/2960 plates
Plastics	2.404×10^{-2}
Unspecified waste	2.404×10^{-2}
Mineral waste	4.425×10^{-1}
Slags and ashes	2.195×10^{-2}
Chemical waste, regulated	1.049×10^{-1}
Chemical waste, unregulated	2.404×10^{-2}
Waste returned to mine	7.844×10^{-2}
Mineral waste from mining	2.339×10^{-1}
PLA waste generated	2.381×10^{-1}

Inputs	Quantity
	Per 2960 plates
PLA pellets	24.04906988 kg
Electricity	22.16250 kWh
Emissions to air	In kg/2960 plates
Particulates	1.861×10^{-3}
Carbon monoxide	1.202×10^{-1}
Sulphur dioxide	5.928×10^{-2}
Nitrogen dioxide	1.858×10^{-1}
Hydrogen chloride	2.164×10^{-4}
Hydrocarbons	3.032×10^{-2}
Nitrogen oxides	8.777×10^{-3}
Hydrogen	3.414×10^{-3}
Methane	3.330×10^{-1}
VOC	7.623×10^{-3}



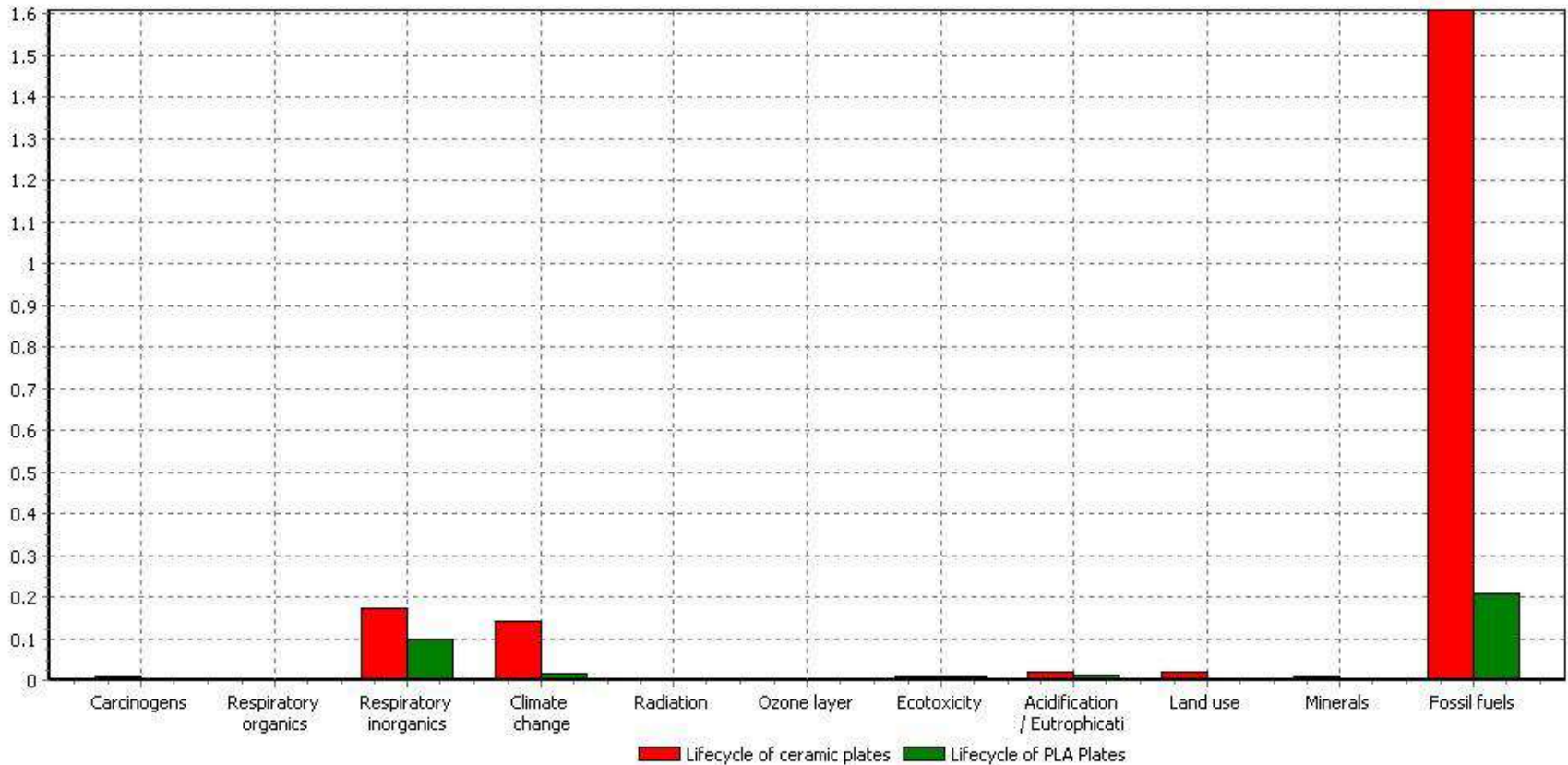
Start from PLA

A comparative analysis of the environmental impacts of ceramic plates and biodegradable plates (made of corn starch) using the Life Cycle Assessment tool.

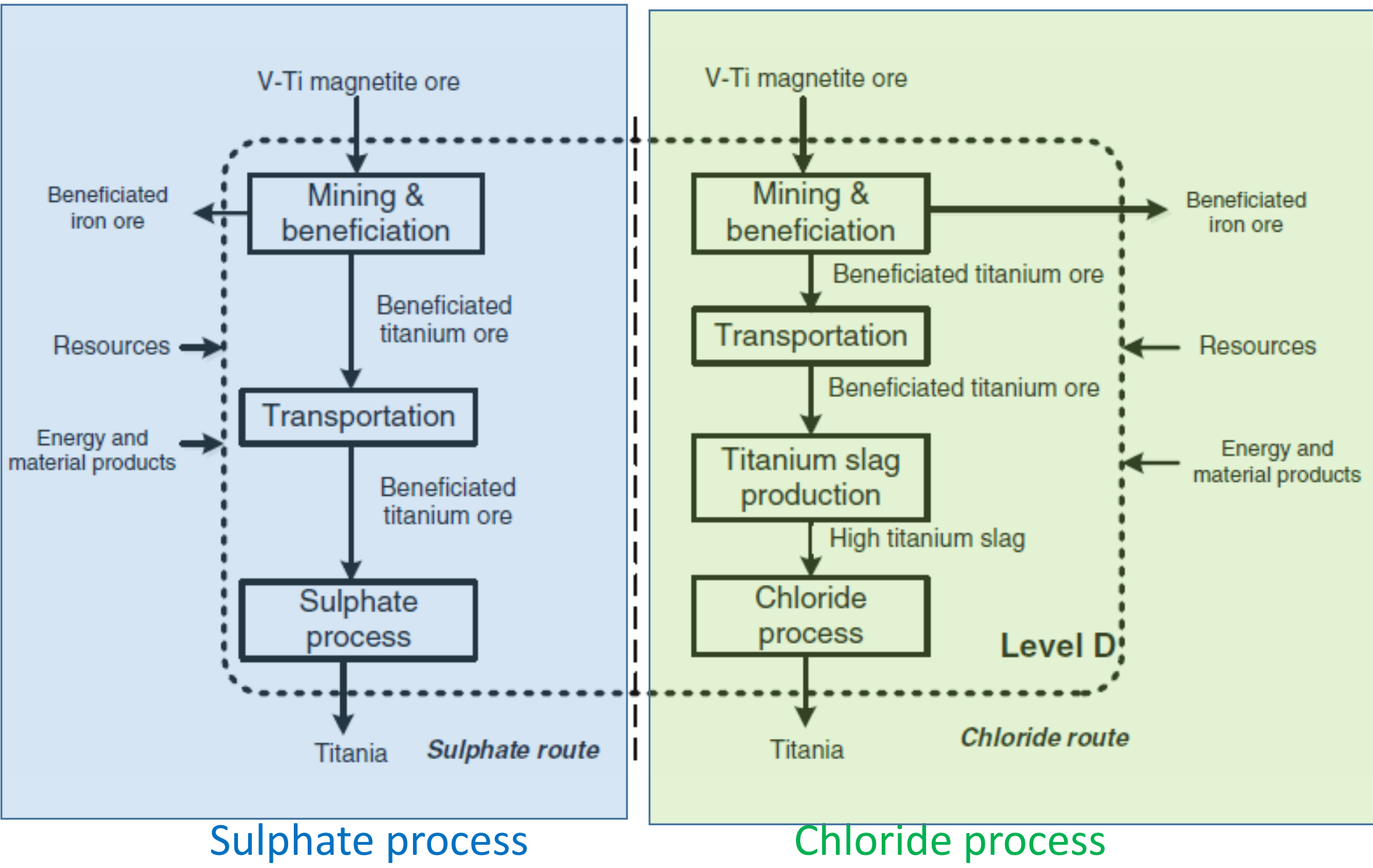
Life-cycle of corn-starch biodegradable plate



Impact analysis of ceramic plate and biodegradable plate



Comparing 1 p 'Lifecycle of ceramic plates' with 1 p 'Lifecycle of PLA Plates'; Method: Eco-indicator 99 (E) v2.04 / Europe EI 99 E/E / normalization



No.	Input	Type	Unit	Chloride route	Sulfate route
1	V–Ti magnetite ore	Resource	kg	5.071	5.576
2	Steel ball	Product	kg	0.001	0.002
3	Anthracite	Product	MJ	10.076	–
4	Coke	Product	kg	0.693	–
5	Liquid chlorine	Product	kg	0.25	–
6	Iron powder	Product	kg	–	0.09
7	Aluminum powder	Product	kg	0.006	–
8	Oxygen	Resource	kg	0.643	–
9	Liquid caustic soda (30 %)	Product	kg	0.3	0.35
10	Sulfuric acid (98 %)	Product	kg	–	4.05
11	Saturated steam (1.3 MPa)	Product	kg	5.5	8
12	Coal	Resource	kg	–	2
13	Petrol	Product	kg	0.017	0.018
14	Diesel	Product	kg	0.011	0.111
15	Process water ^a	Resource	kg	53.758	101.787
16	Electricity	Product	kWh	2.85	1.578

■ Chloride route ■ Sulphate Route

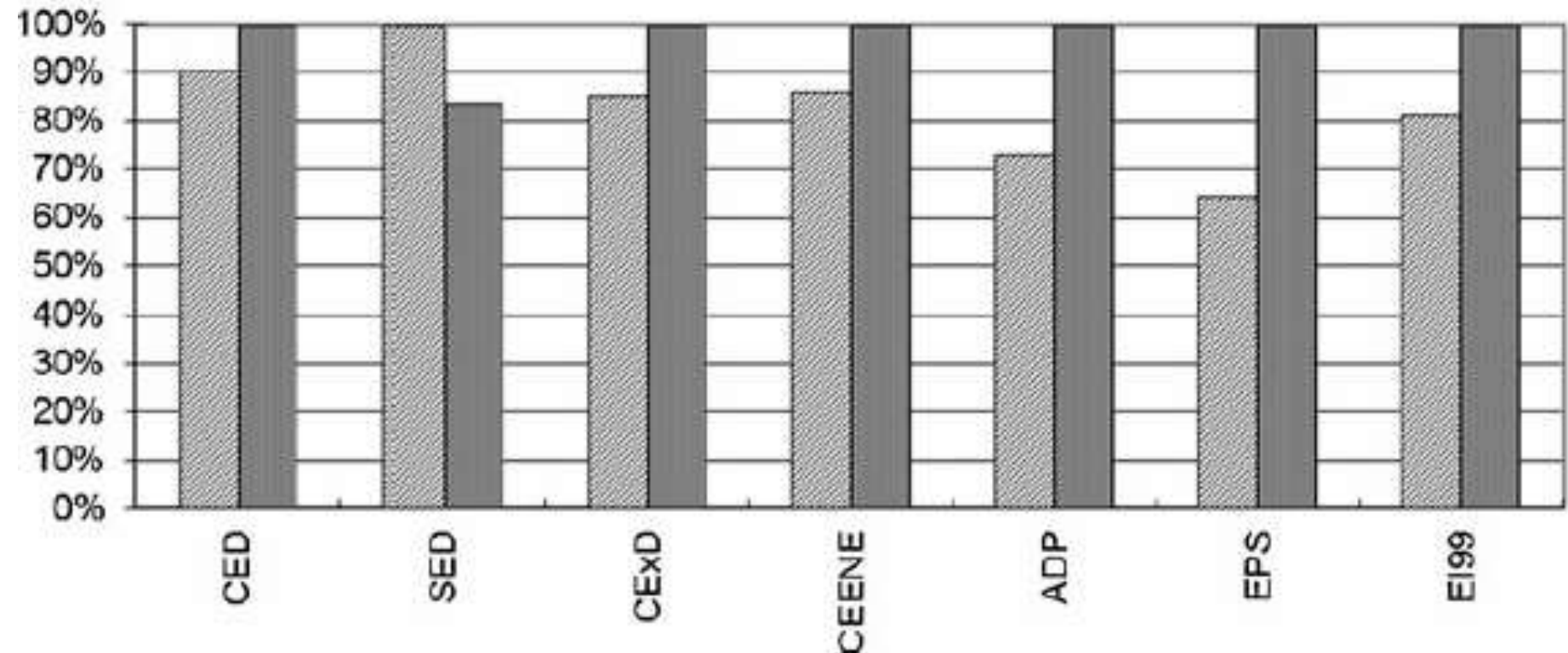


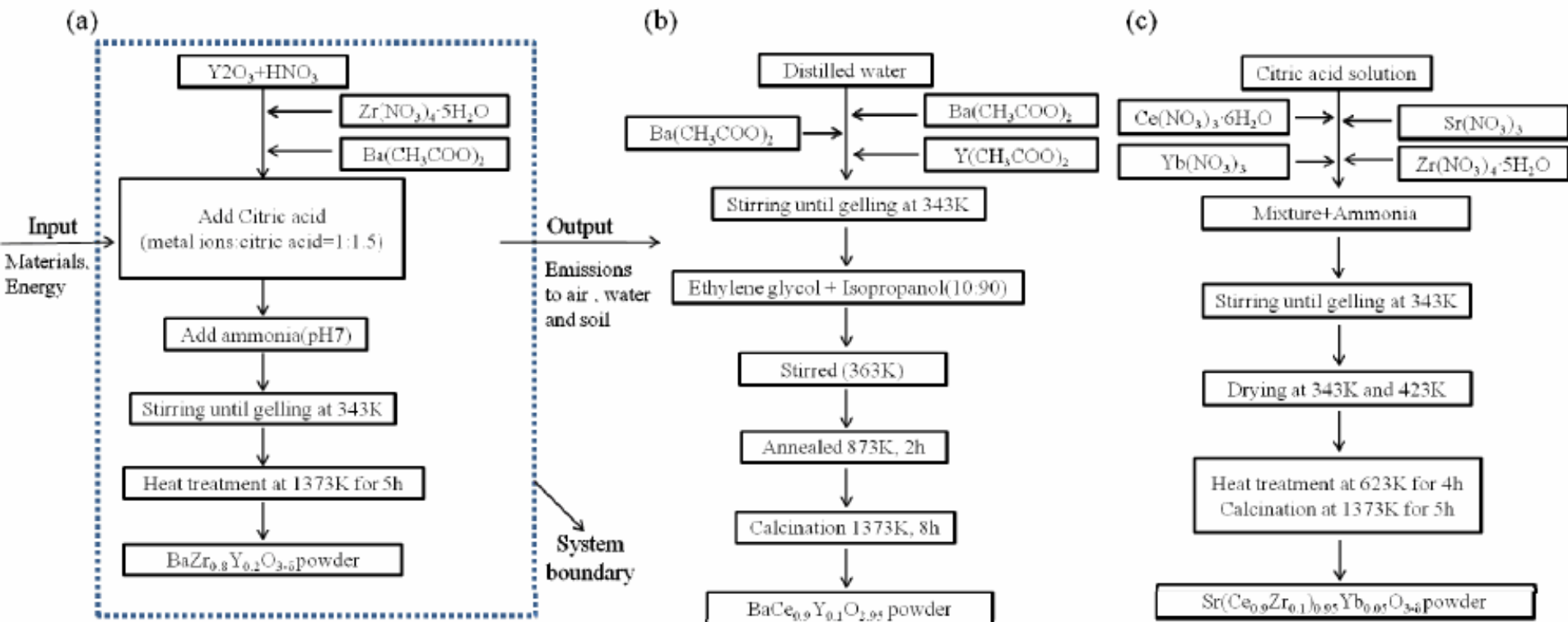
Table 2 Synthesis of resource indicators and resource groups addressed in this study

Resource group ^a	Type ^b	CED	SED	CExD	CEENE	ADP	EI99	EPS
Atmospheric	n.d.		×		×			
Fossil	NRR	×	×	×	×	×	×	×
Land	n.d.		×		×			
Metal ores	NRR		×	×	×	×	×	×
Minerals	NRR		×	×	×	×	×	×
Nuclear	NRR	×	×	×	×			×
Renewable energy	RR	×	× ^c	×	× ^d			
Water ^c	RR		×	×	×	×		

Life Cycle Assessment for Proton Conducting Ceramics Synthesized by the Sol-Gel Process :Soo-Sun Lee and Tae-Whan Hong, *Materials* 2014, 7, 6677-6685;

- The proton conducting ceramics $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ (BZY), $\text{BaCe}_{0.9}\text{Y}_{0.1}\text{O}_{2.95}$ (BCY10), and $\text{Sr}(\text{Ce}_{1-x}\text{Zr}_x)_{0.95}\text{Yb}_{0.05}\text{O}_{3-\delta}$ (SCZY)

Figure 1. Flow charts of experimental procedures of (a) BZY20 ($\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$); (b) BCY10 ($\text{BaCe}_{0.9}\text{Y}_{0.1}\text{O}_{2.95}$); (c) SCZY ($\text{Sr}(\text{Ce}_{1-x}\text{Zr}_x)_{0.95}\text{Yb}_{0.05}\text{O}_{3-\delta}$).



Life Cycle Assessment for Proton Conducting Ceramics Synthesized by the Sol-Gel Process :Soo-Sun Lee and Tae-Whan Hong, *Materials* 2014, 7, 6677-6685;

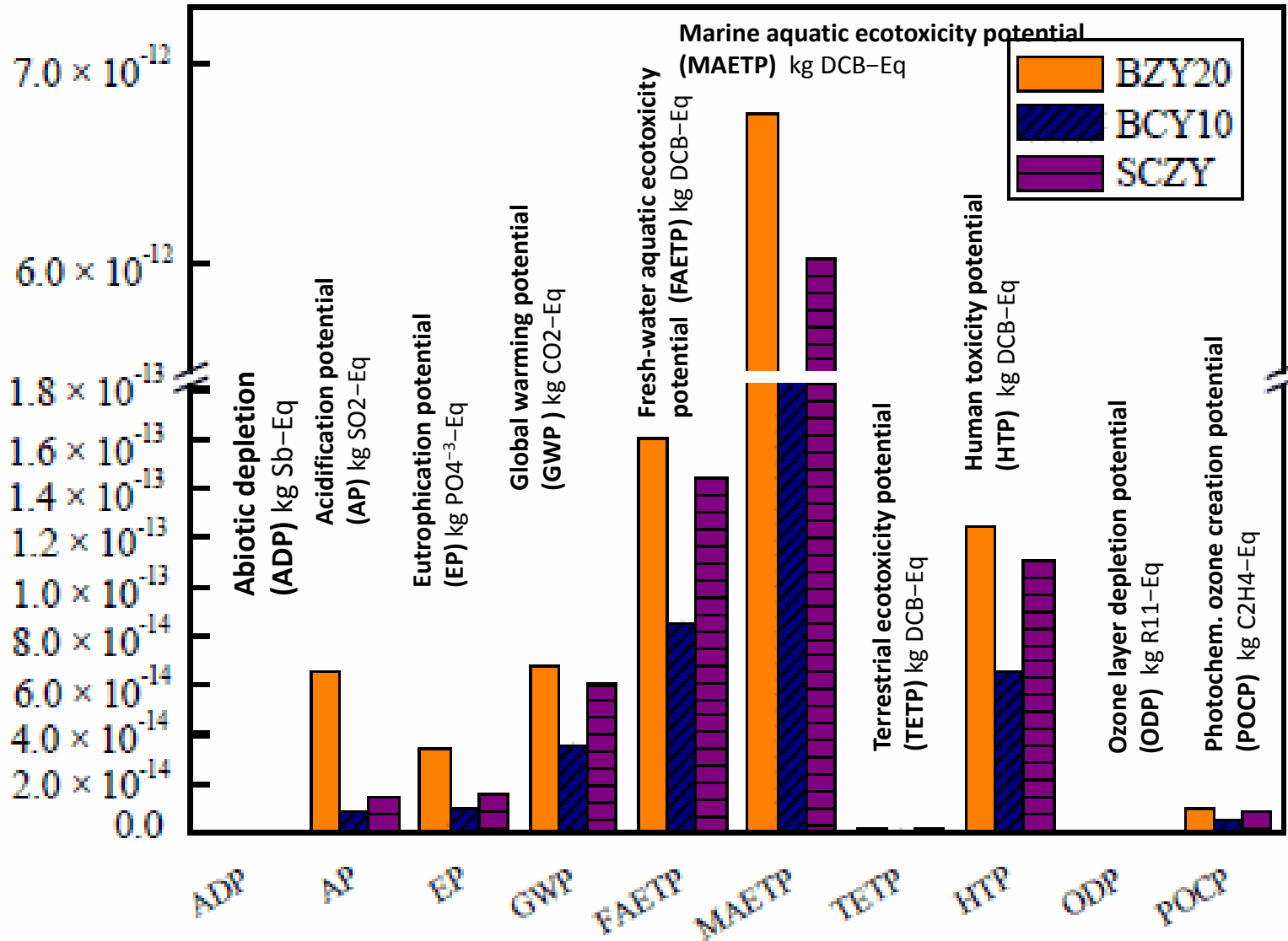
- The proton conducting ceramics BaZr_{0.8}Y_{0.2}O_{3-δ} (BZY), BaCe_{0.9}Y_{0.1}O_{2.95} (BCY10), and Sr(Ce_{0.9}Zr_{0.1})_{0.95}Yb_{0.05}O_{3-δ} (SCZY)

Table 1. Life cycle inventory for the synthesis of BZY20 (BaZr_{0.8}Y_{0.2}O_{3-δ}), BCY10 (BaCe_{0.9}Y_{0.1}O_{2.95}) and SCZY (Sr(Ce_{1-x}Zr_x)_{0.95}Yb_{0.05}O_{3-δ}).

Classification	BZY20	BCY10	SCZY
Raw-materials	Ammonia (3.6 g)	Barium (0.6 g)	Ammonia (0.6 g)
	Barium (1.3 g)	Cerium (1.0 g)	Cerium (1.9 g)
	Nitric acid (13.7 g)	Yttrium (0.1 g)	Citric acid (2.9 g)
	Yttrium (0.2 g)	Ethylene glycol (0.2 g)	Strontium (1.1 g)
	Zirconium (0.9 g)	Isopropanol (1.5 g)	Ytterbium (1.1 g)
	Distilled water (21.5 g)		Zirconium (1.1 g)
Electrical equipment	Running time (h)	Running time (h)	Running time (h)
Drying oven	48 (57.6 kWh)	2 (3.8 kWh)	24 (28.8 kWh)
Electric furnace	5 (25 kWh)	8 (40 kWh)	9 (945 kWh)
Emissions to water	Ammonia (3.6 g) Nitric acid (13.7 g)	Ethylene glycol (0.2 g)	Ammonia (0.6 g)
Emissions to air		Isopropanol (1.5 g)	

Life Cycle Assessment for Proton Conducting Ceramics Synthesized by the Sol-Gel Process :Soo-Sun Lee and Tae-Whan Hong, *Materials* 2014, 7, 6677-6685;

Figure 2. Normalized impacts of BZY20, BZY10 and SCZY the according to CML 2001.



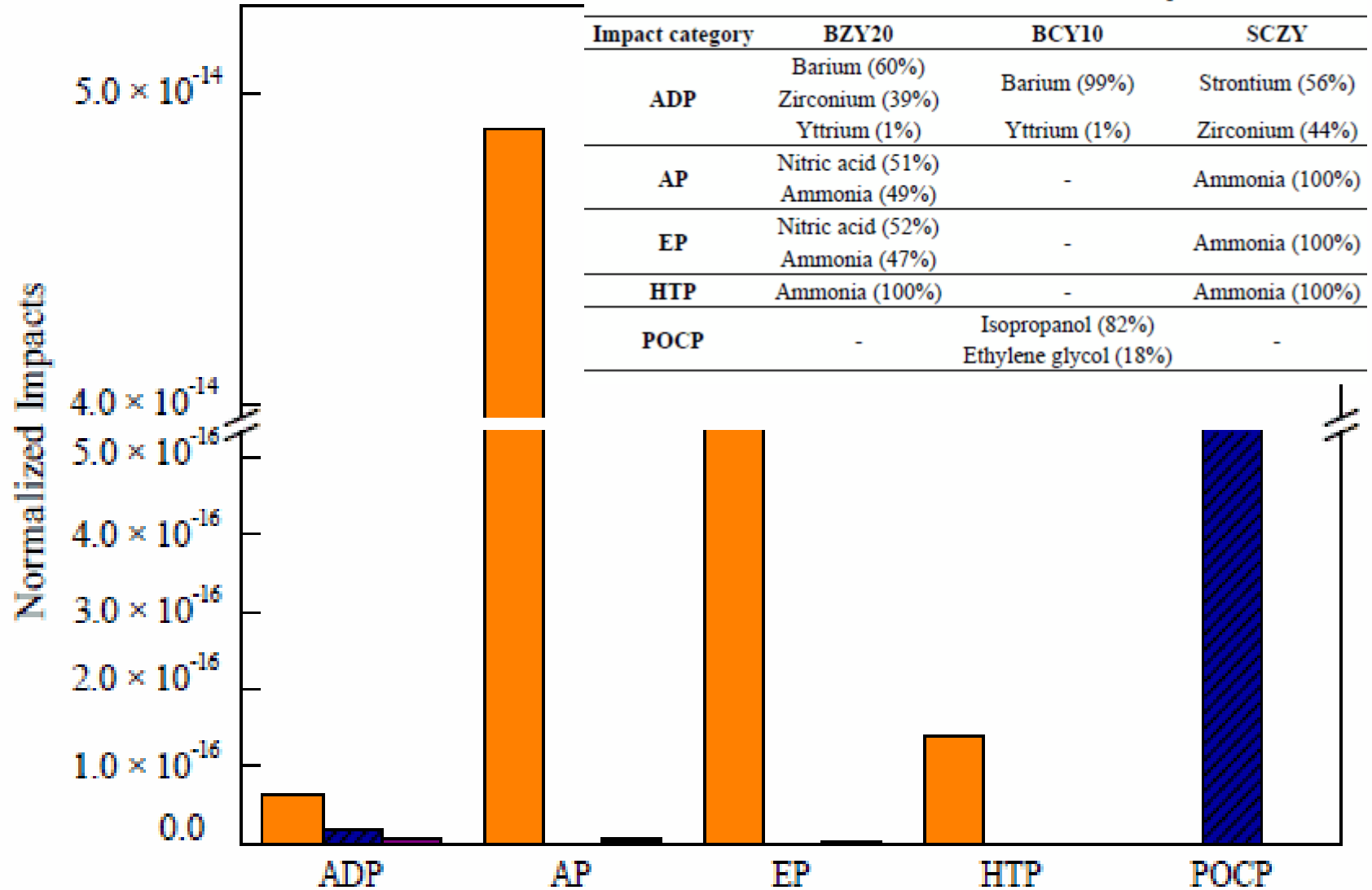
Life Cycle Assessment for Proton Conducting Ceramics Synthesized by the Sol-Gel Process :Soo-Sun Lee and Tae-Whan Hong, *Materials* 2014, 7, 6677-6685;



Figure 3. Normalized impacts of BZY20, BZY10 and SCZY (without electricity)

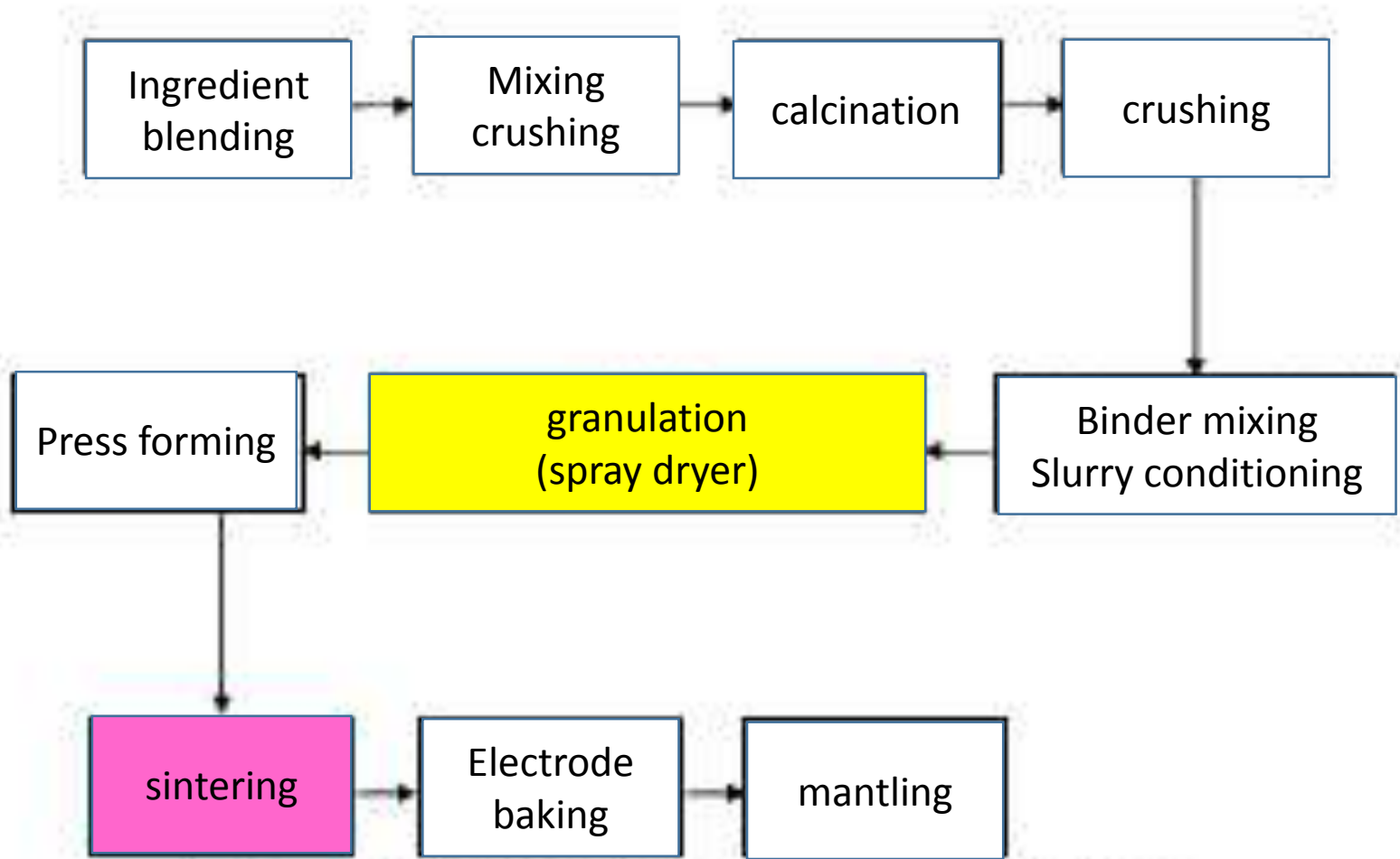
Table 3. Contribution to Environmental Impacts.

Impact category	BZY20	BCY10	SCZY
ADP	Barium (60%)	Barium (99%)	Strontium (56%)
	Zirconium (39%)		Zirconium (44%)
	Yttrium (1%)	Yttrium (1%)	
AP	Nitric acid (51%)	-	Ammonia (100%)
	Ammonia (49%)		
EP	Nitric acid (52%)	-	Ammonia (100%)
	Ammonia (47%)		
HTP	Ammonia (100%)	-	Ammonia (100%)
POCP	-	Isopropanol (82%)	-
		Ethylene glycol (18%)	



The study on LCA of the zirconia device

Koji Noda, Ruilu Liang, Takao soma, Eiji Kikuti, Hiroto Kawashima



Manufacturing process of ZrO₂ sensor

Spry dryer process

微粒化方式：ロータリーアトマイザ
 空気が熱方式：ブタン直接燃焼方式

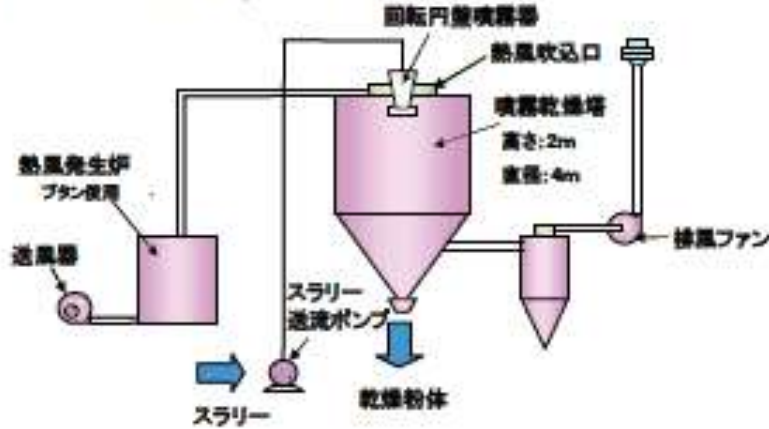


表1 スプレードライヤーの運転条件

capacity	300kg/h
composition	solid 50% aqua 50%
drying temperature	inlet 230°C outlet 105°C
evaporation rate	150kg/h
electric power	18KWh
butane consumption	18kg/h
Process yield	85%

sintering process

Electric furnace 200*200*250mm

ZrO₂ 7.8kg

Heat-up to 600°C at 50°C/h

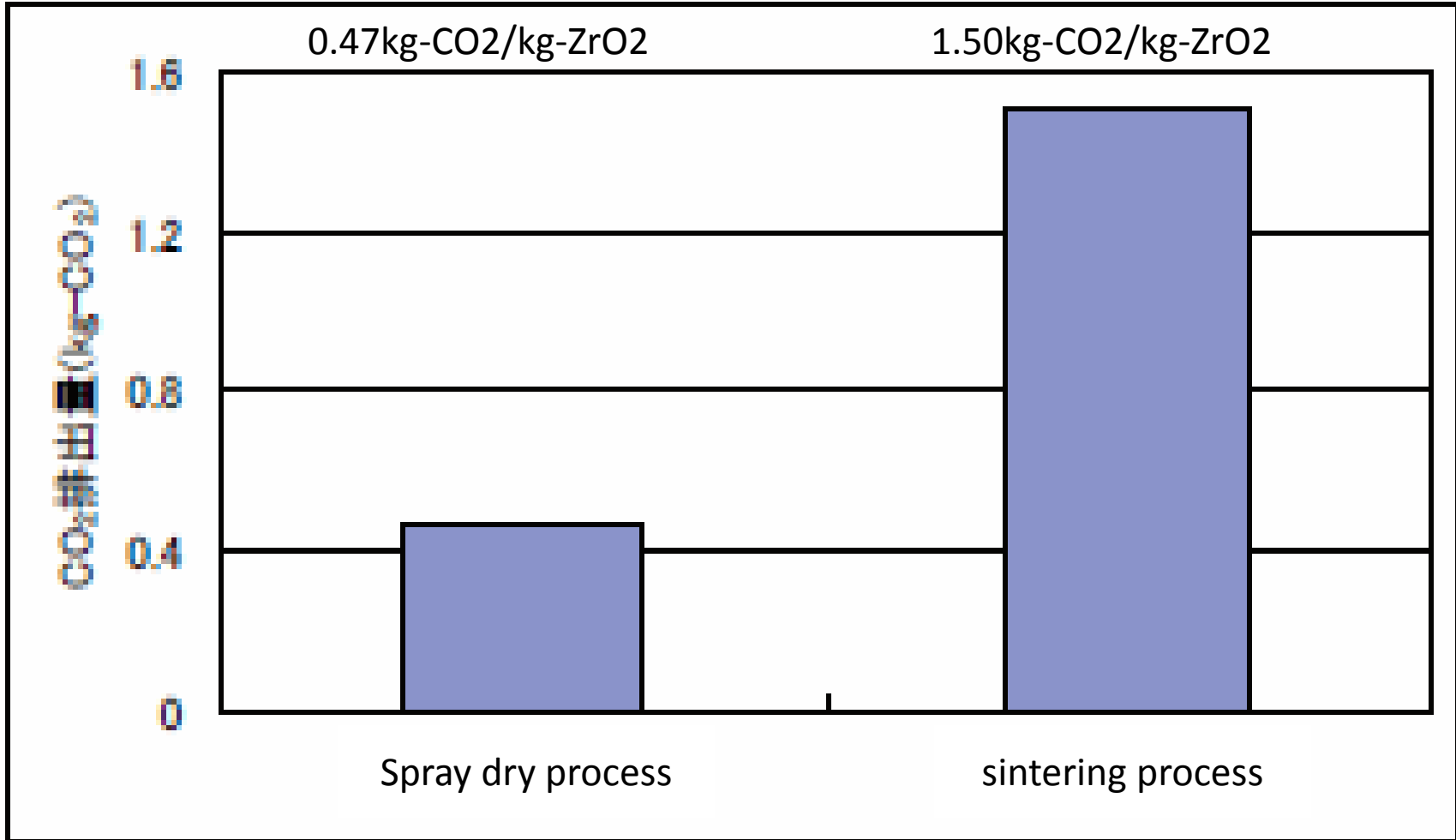
De-binder 600°C 2h

Heat-up to 1400°C at 100°C/h

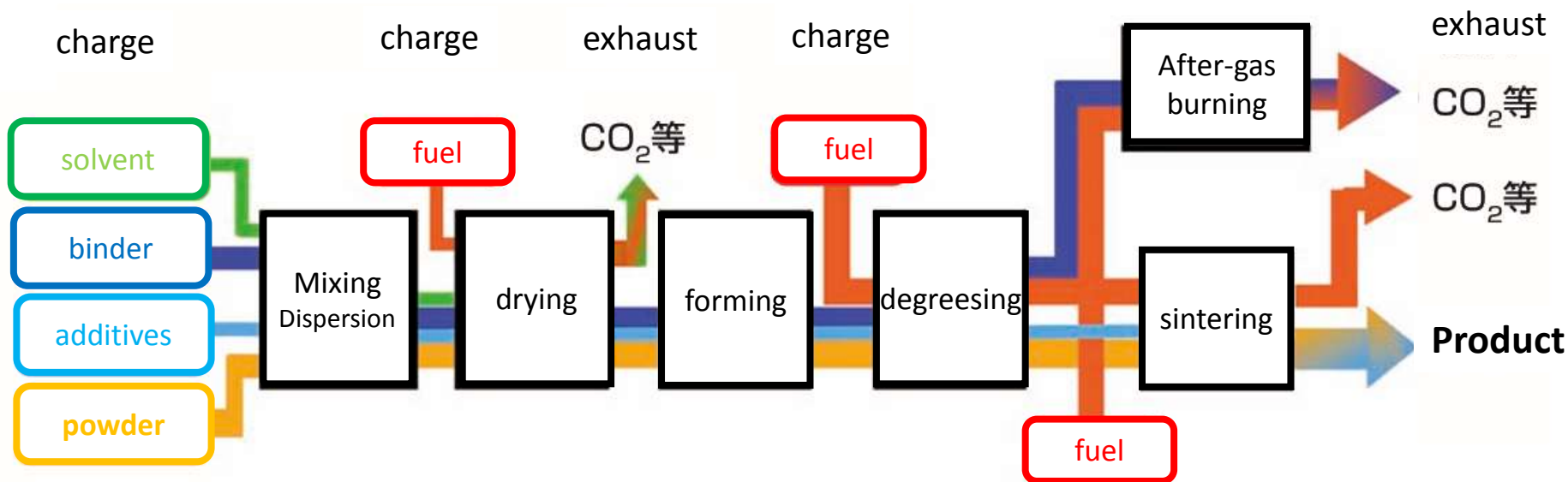
Calcining 1400°C 2h

The study on LCA of the zirconia device

Koji Noda, Ruilu Liang, Takao soma, Eiji Kikuti, Hiroto Kawashima



Comparison of CO2 emission in each process



Process and material flow in ceramic production

1kg sintering of Al₂O₃

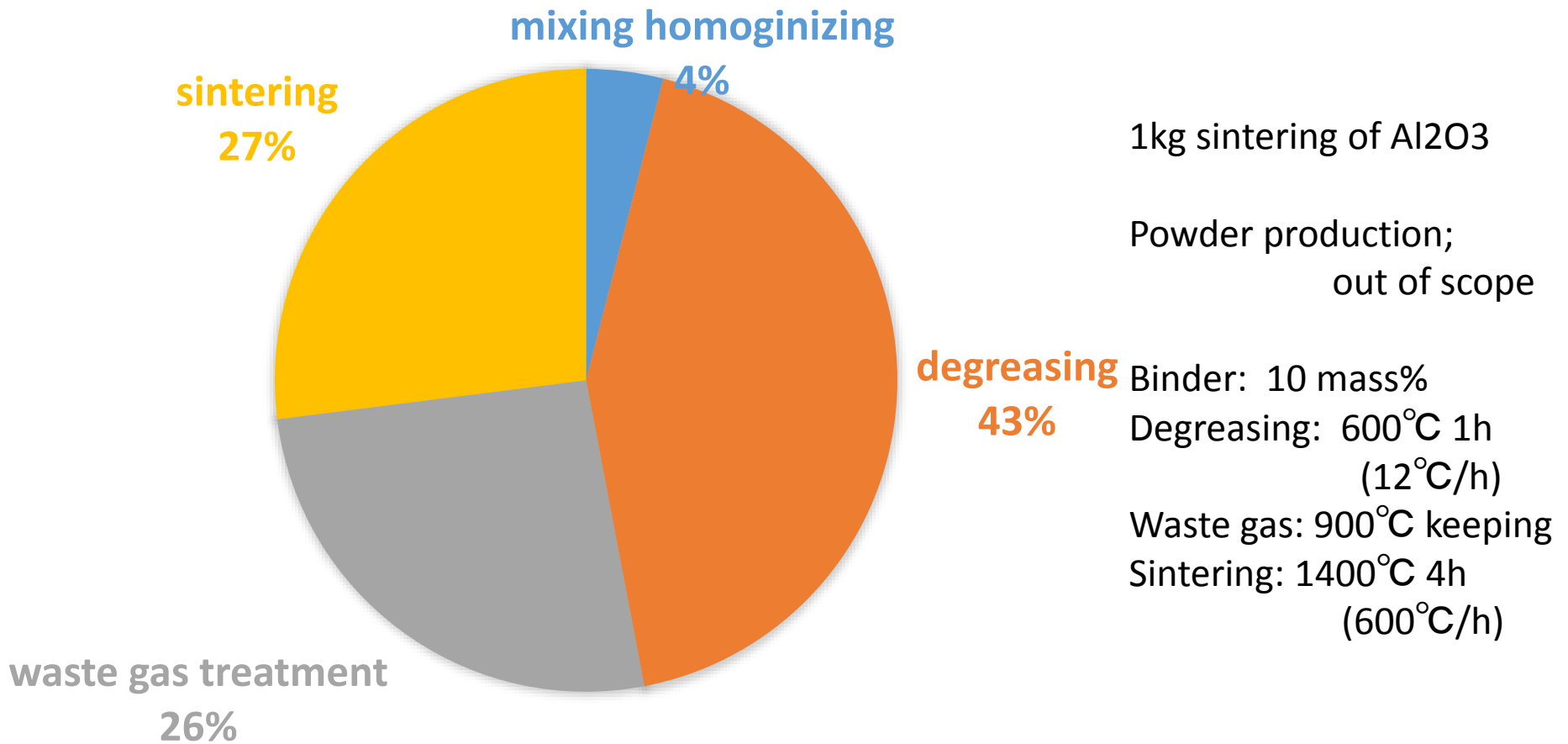
Powder production; out of scope

Binder: 10 mass%

Degreasing: 600°C 1h (12°C/h)

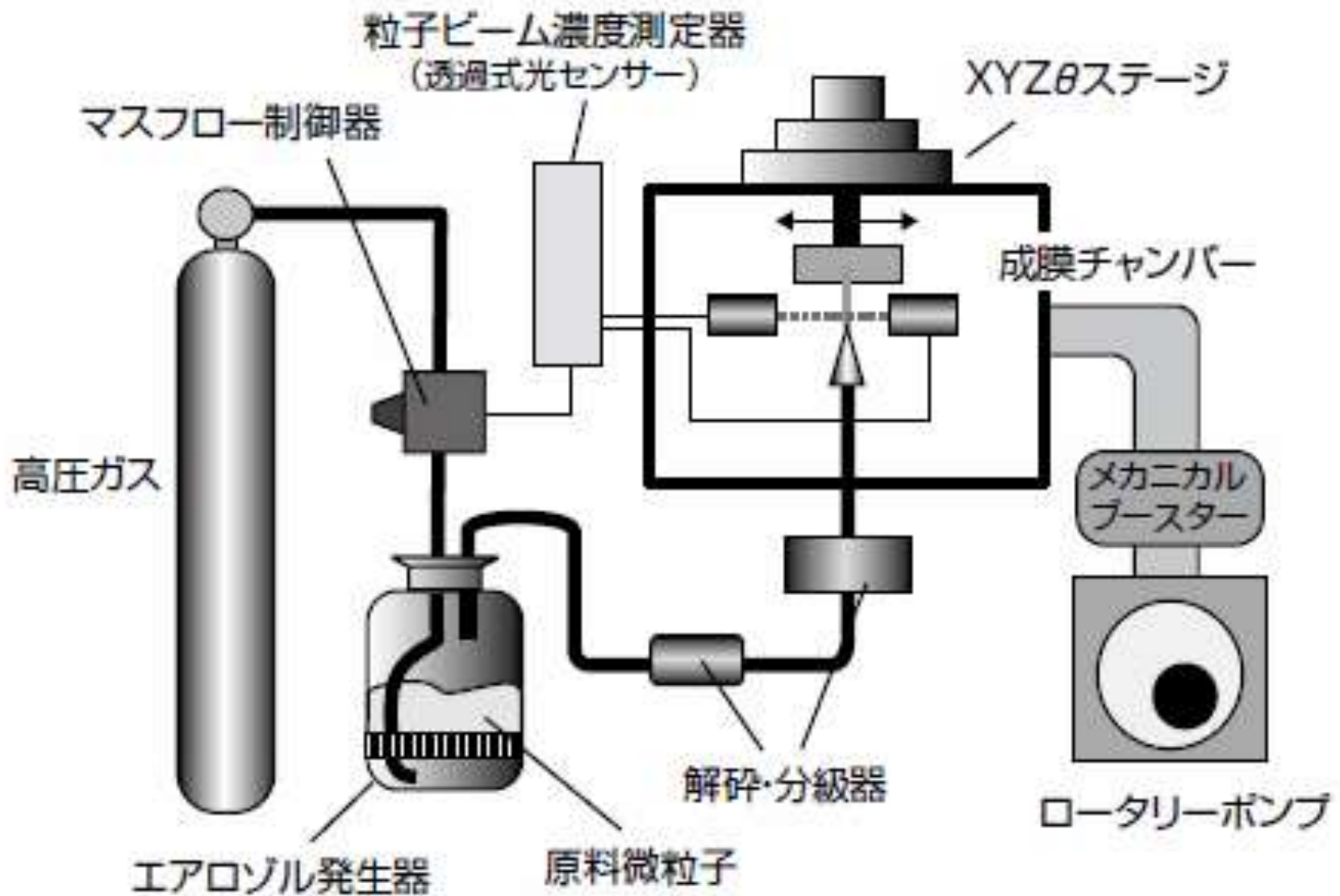
Waste gas: 900°C keeping

Sintering: 1400°C 4h (600°C/h)



Aerosol deposition method

Akedo, AIST



	構 成	工 程
焼成法 (従来製法)		
AD法		
AD法の 特 長	<ul style="list-style-type: none"> ● アルミナ100%でウェハへの重金属污染がない。 ● 薄いセラミック膜であるため、熱伝導性に優れる。 	<ul style="list-style-type: none"> ● 従来の焼成プロセスに対してAD成膜プロセスは、投入エネルギーで82%の削減効果(7500 kWh → 1300 kWh) 時間で96%の削減効果(288時間 → 10時間)

図4 静電チャックの構造とAD法導入による製造工程のエネルギー消費比較

COMPARATIVE STUDY OF NANOPARTICLE PRODUCTION TECHNOLOGIES FOCUSED ON ENVIRONMENTAL ASPECTS

Barbora STIEBEROVÁ, Miroslav ŽILKA, Marie TICHÁ, František FREIBERG, Jan HOŠEK

Czech Technical University in Prague, Faculty of Mechanical Engineering, Technická 4, 16607 Praha 6, Barbora.Stieberova@fs.cvut.cz

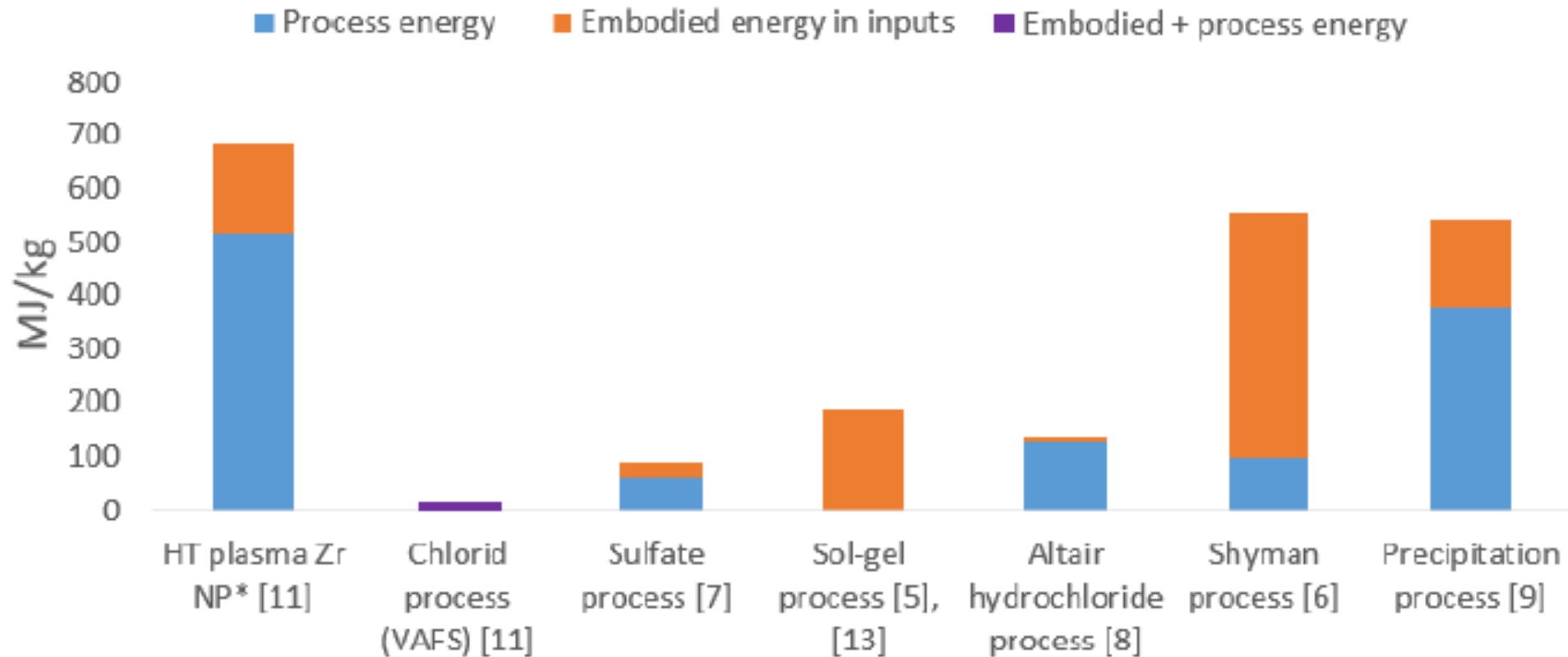
Technology/criterion	Productivity	Quality	Variability	Cost of inputs	Cost of equipment	Energy Consumption process	Energy Consumption embodied	CO2 emissions	Important sources
HT plasma	High/Medium	Good	High	Different	High	Very high	Different	Very high	[11], [13], [14], [15]
LT plasma	Low	Very good	Medium	Different	Medium	N/A	Different	N/A	[13]
VAFS	High	Good	Low	Low	High	Low	Low	Low	[16], [17]
FSP	Medium	Good	High	High	High	Low	High	Medium?	[4], [17], [18], [19]
CS solution	Low	Very good	High	High	Low	N/A	High	N/A	[20], [21]
Sol-gel	Low	Very good	Very high	Different	Low	N/A	Different	N/A	[5], [8], [22], [23]
Solvothermal	Low	Very good	Very high	High	Medium	N/A	High	N/A	[23]
Hydrothermal	Low	Very good	Very high	High	Medium	N/A	High	N/A	[23]
Altair	High	Good	Low	Low	? Medium	Low	Low	Low	[9], [24]
Shyman	Medium	Very good	Very high	High	Medium	Low	Very high	High	[6], [8]
Precipitation	Low	Very good	Very high	Different	Low	High	Medium	High	[10], [23]

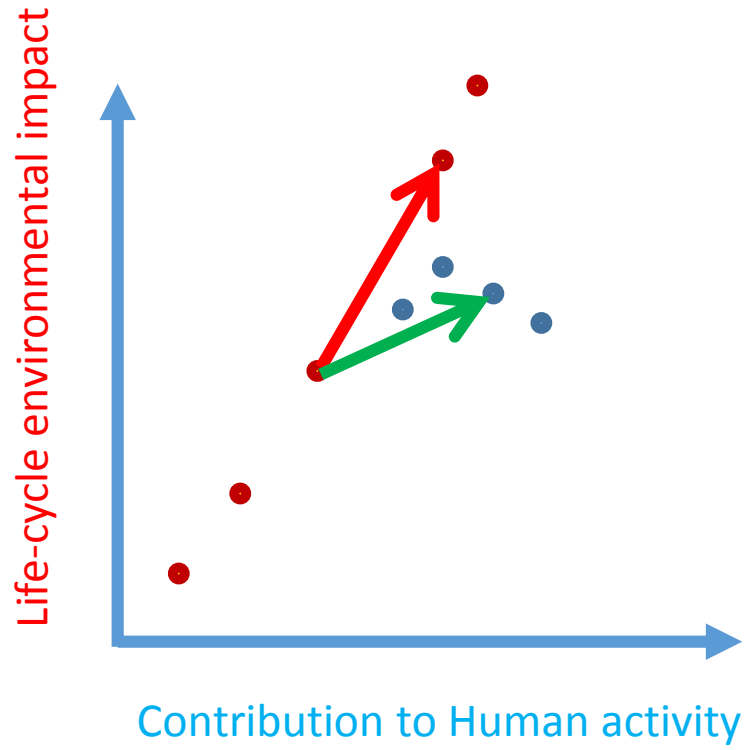
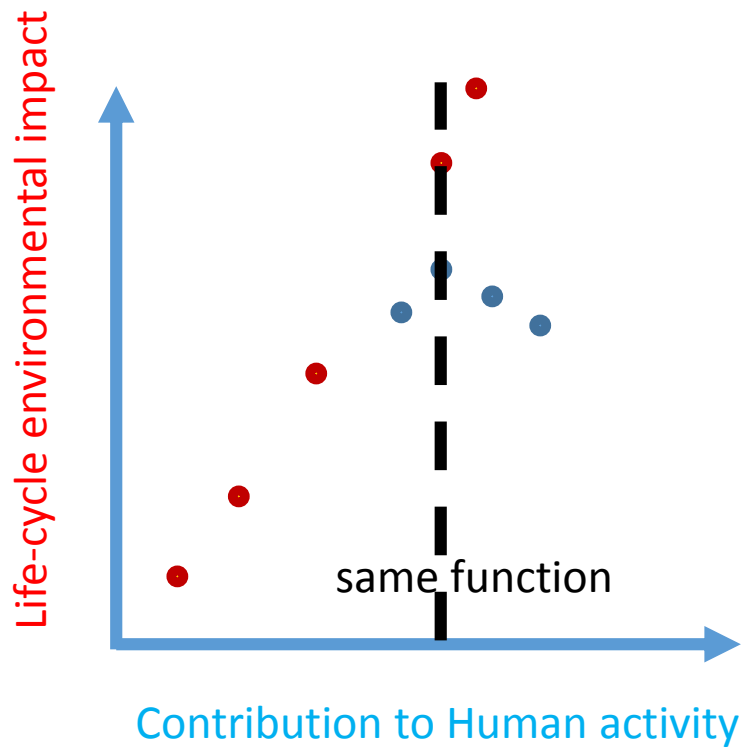
Fig. 1 The overall comparison of the different NP production technologies (HT- high temperat., LT- low temperat., VAFS – vapour-fed aerosol flame synth., FSP – flame spray pyrolysis, CS – combustion synth., Altair – Altair hydrochloride process, SHYMAN – continuous supercritical hydrothermal syntheses)

COMPARATIVE STUDY OF NANOPARTICLE PRODUCTION TECHNOLOGIES FOCUSED ON ENVIRONMENTAL ASPECTS

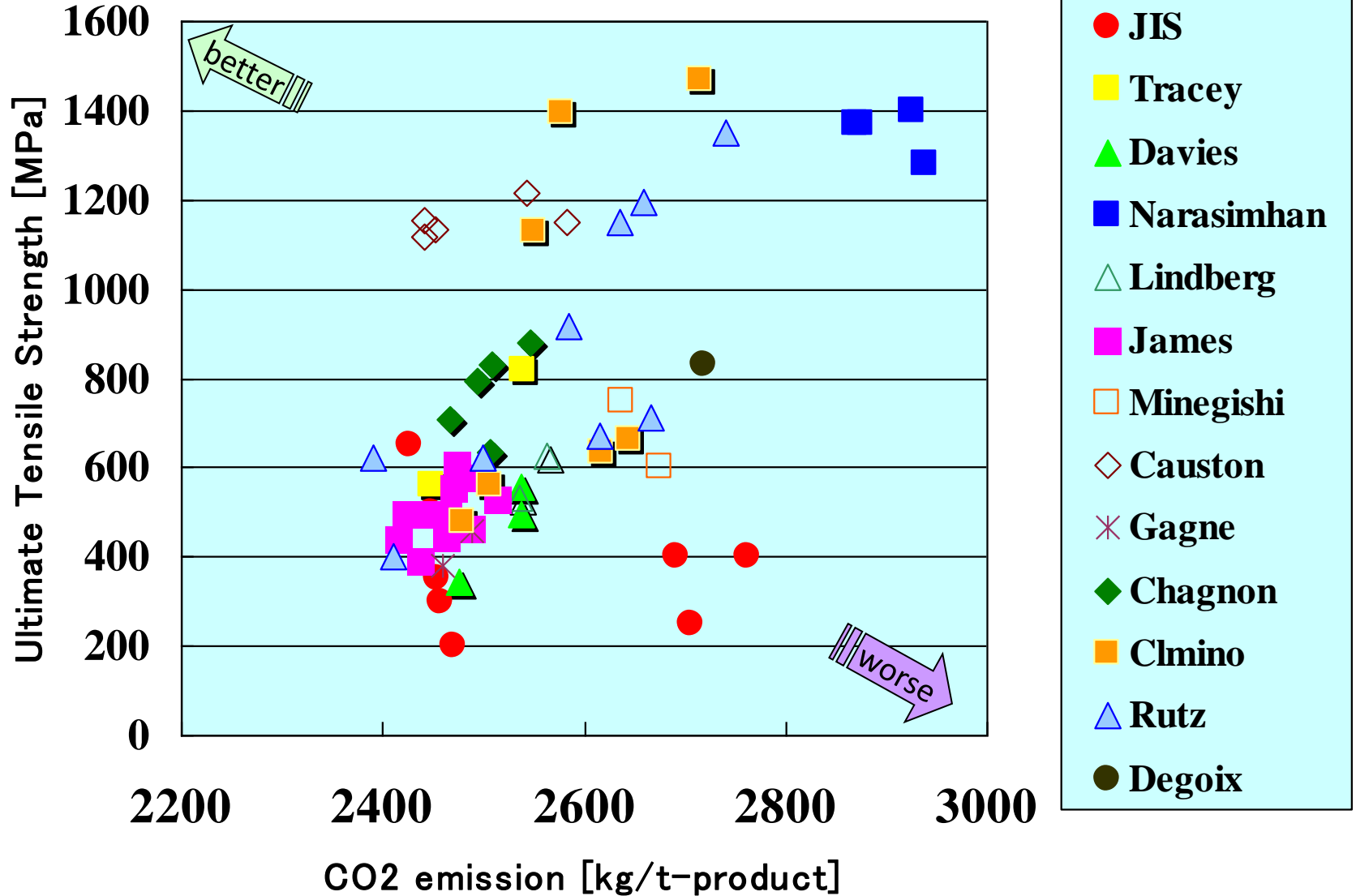
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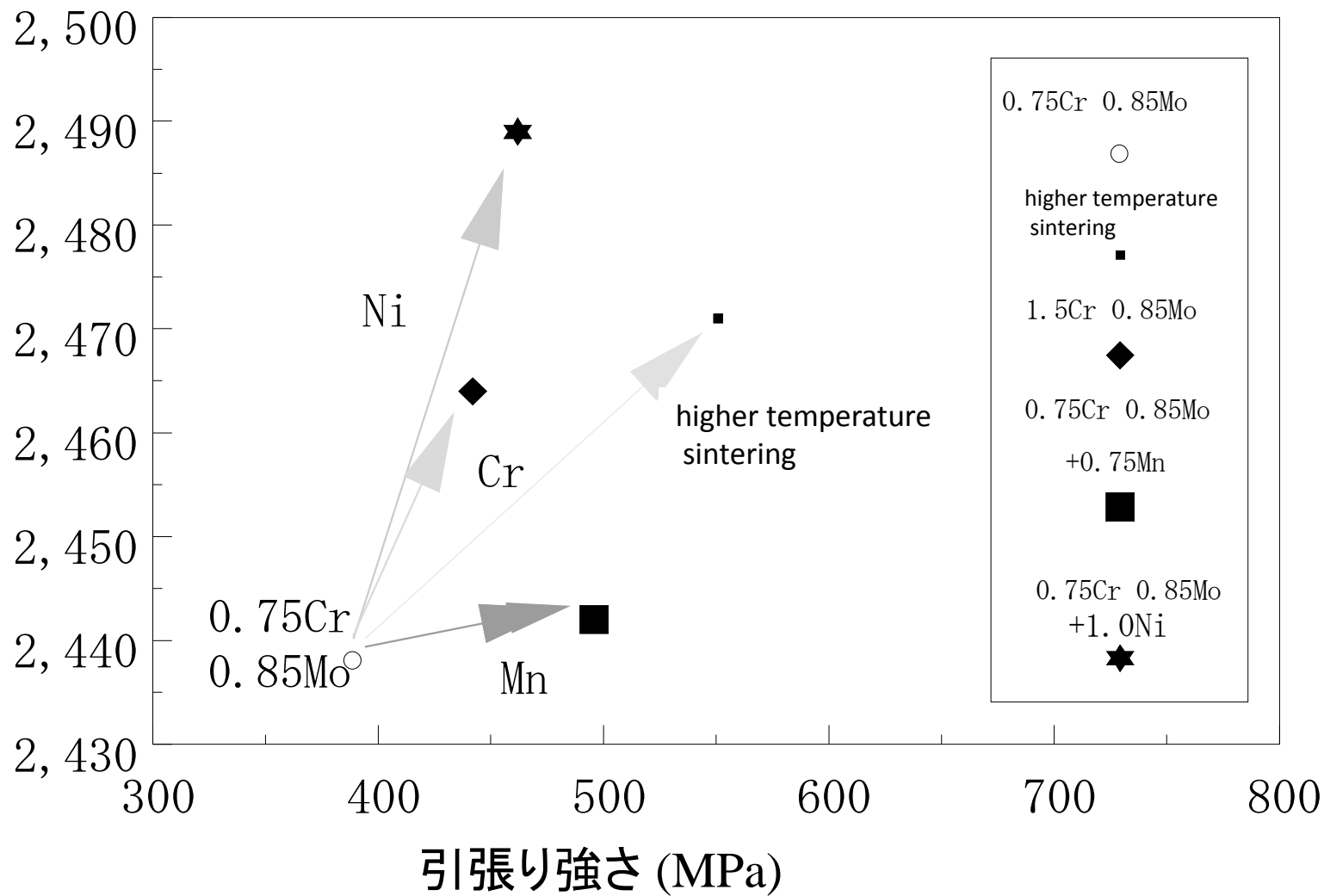




Life-cycle Emission v.s. Strength of PM Products



CO2 排出量 (g/kg-product)



service=func.(work, provide, support)

service=func.(work, provide, support)

burden in production + burden through usage + burden of End-of-Life - deduction by recycling

life-cycle environmental burden =
burden in production + burden through usage
+ burden of End-of-Life - deduction by recycling

Resource Efficiency

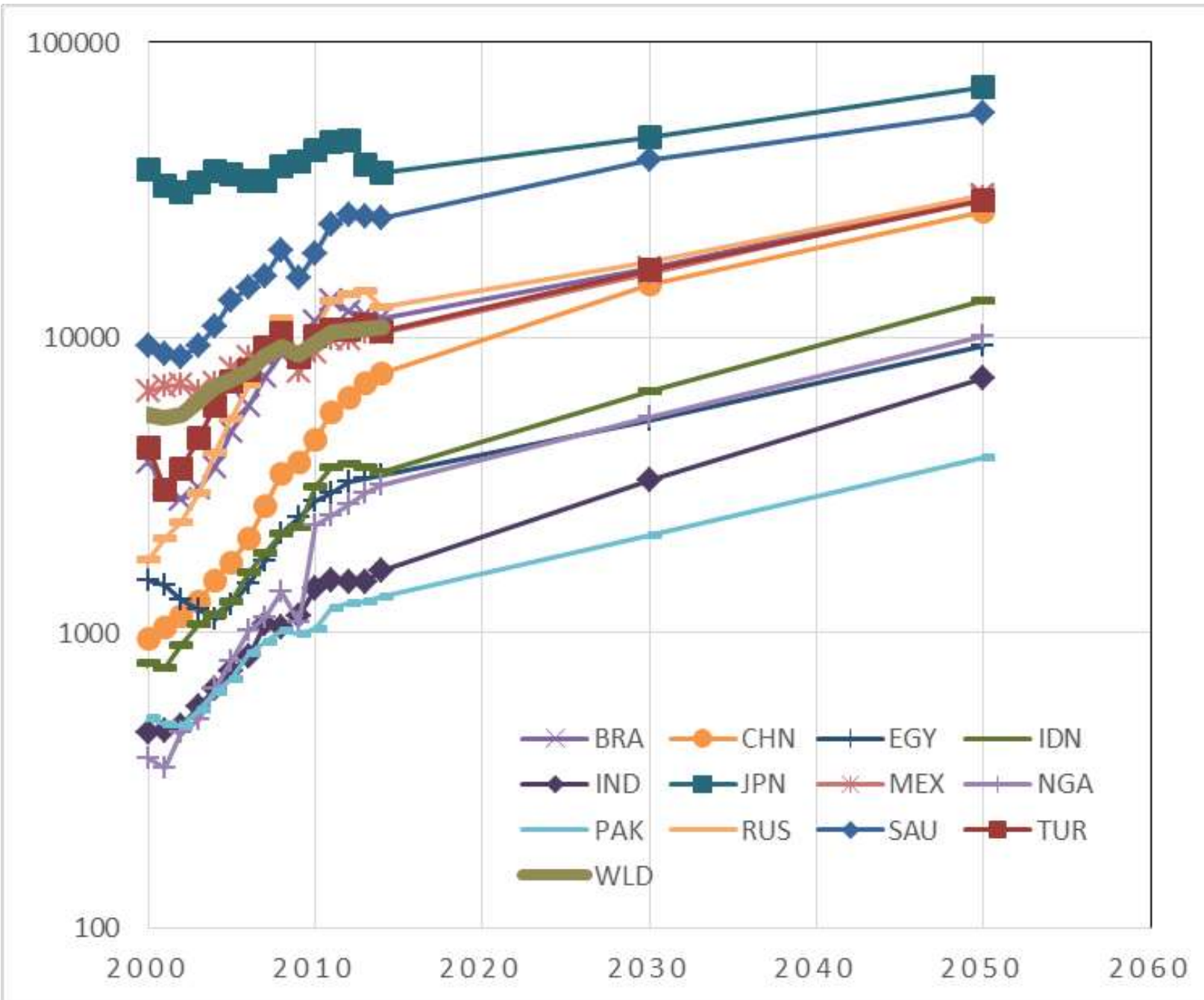
The Elements with sustainability parameters

- Durable years: (reserve)/(annual consumption)
- Resource-view weight: tons of TMR for 1kg of metal production
- Share % Of top country of production, country code
- Increase of production from 1999 to 2009, (%)

H scarcity TMR domination acceleration																		He					
Li 194 1.5 41CL 120	Be 2.5 86US 42	<ul style="list-style-type: none"> Magnet, motor Batteries IC tips and parts Electric wiring lightning Optical function Information media Thermoelectric, Catalyst, electrode Structural material Display & its porishing Fire retardant Solar cell 																B 0.14 47TK 101	C	N	O	F	Ne
Na 0.4 56 100	Mg 5500 0.07 82CN 215																	Al 164 0.05 31CN 163	Si 0.03 65CN 169	P 124 35CN 114	S	Cl	Ar
K	Ca	Sc	Ti 1300 0.04 23AU 220	V 208 1.5 37CN 135	Cr 60 0.03 42ZA 180	Mn	Fe	Co 122 0.61 40CG 219	Ni 41 0.26 19RU 125	Cu 31 0.36 34CL 125	Zn 22 0.04 28CN 131	Ga 7.3 157	Ge 32 71CN 241	As 0.03 47 129	Se 59 0.45 50JP 119	Br 38IL 86	Kr						
Rb	Sr 0.51 48ES 133	Y6 1 2.7 271	Zr 4200 0.55 41AU 151	Nb 73 0.64 92BR 335	Mo 48 0.75 25US 155	Tc	Ru 79 79ZA 119	Rh 160 2300 79ZA 85	Pd 160 810 41ZA 156	Ag 14 4.8 18PL 134	Cd 0.07 23CN 94	In 24 12 50CN 250	Sn 22 2.5 37CN 153	Sb 0.06 91CN 136	Te 10 44JP 88	I 600 59CL 159	Xe						
Cs	Ba 31 0.51 147	(Ln)	Hf 10 151	Ta 33 6.8 48AU 245	W 40 0.2 81CN 185	Re	Os 18 540 79ZA 118	Ir 400 79ZA 40	Pt 160 530 79ZA 118	Au 17 1100 13CN 101	Hg 32 2 63CN 56	Tl	Pb 17 0.03 43CN 128	Bi 57 0.22 62CN 221	Po	At	Rn						
Fr	Ra	(An)	La 1600 8.2 371*	Ce 770 18 295*	Pr 7.9	Nd 420 12 90*	Pm	Sm 16	Eu 188 33	Gd 17	Tb 244 55	Dy 209 16	Ho	Er	Tm	Yb	Lu						
			Ac	Th	Pa	U																	

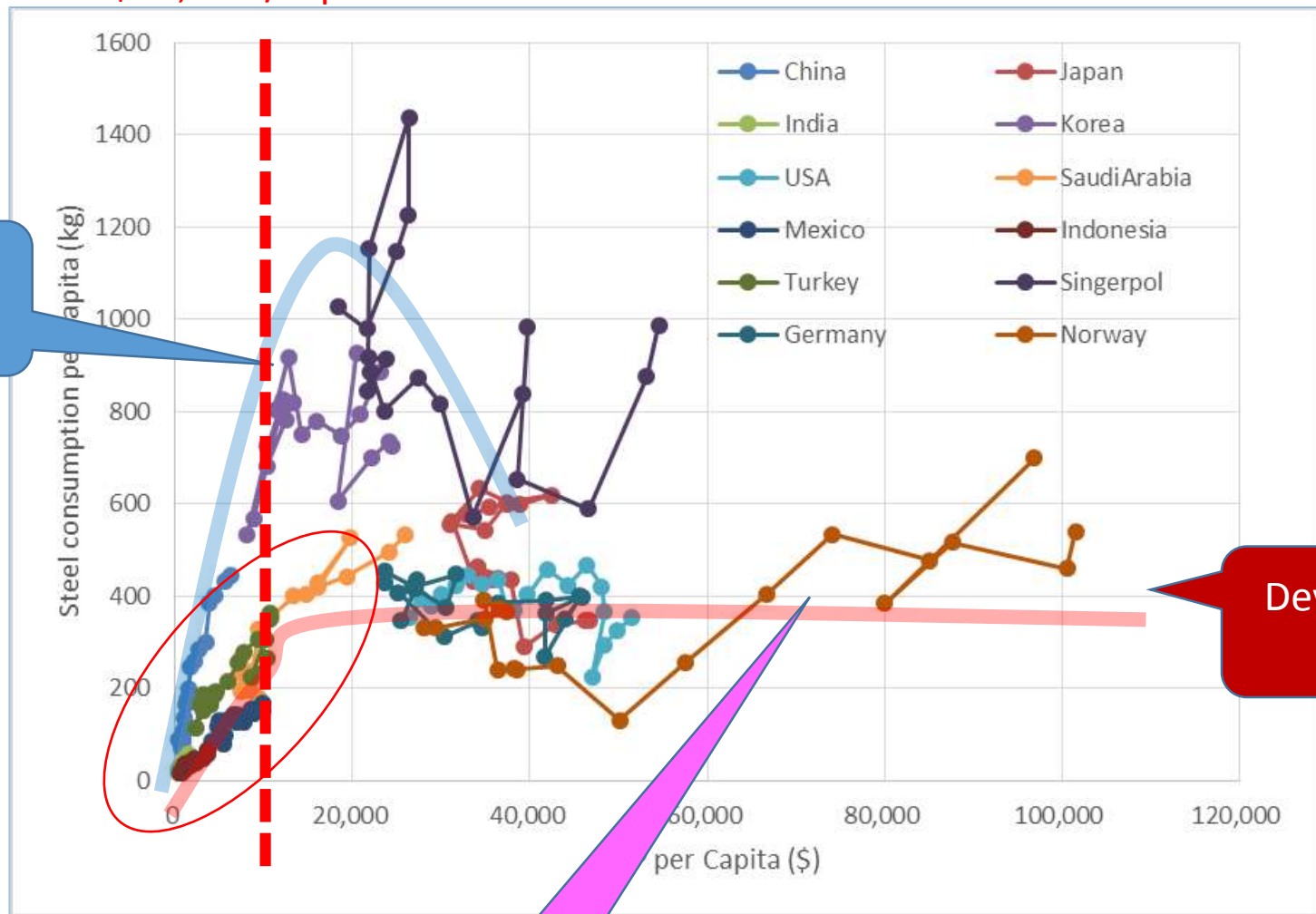
* Estimated by import of Japan, () amount in crust is less than in sea water

Data form 米国鉱山局データ USGS minerals information
 工業レアメタル (Kogyo rare metal) Japanese journal
 「概説 資源端重量」 NIMS-EMC data on mat. & env. No.18
 Halada, Katagiri, Proc. of EcoBalance 2010 p609



Fe consumption / capita v.s. GDP/ capita from 1994 to 2014

\$10,000 /capita



Exporting countries

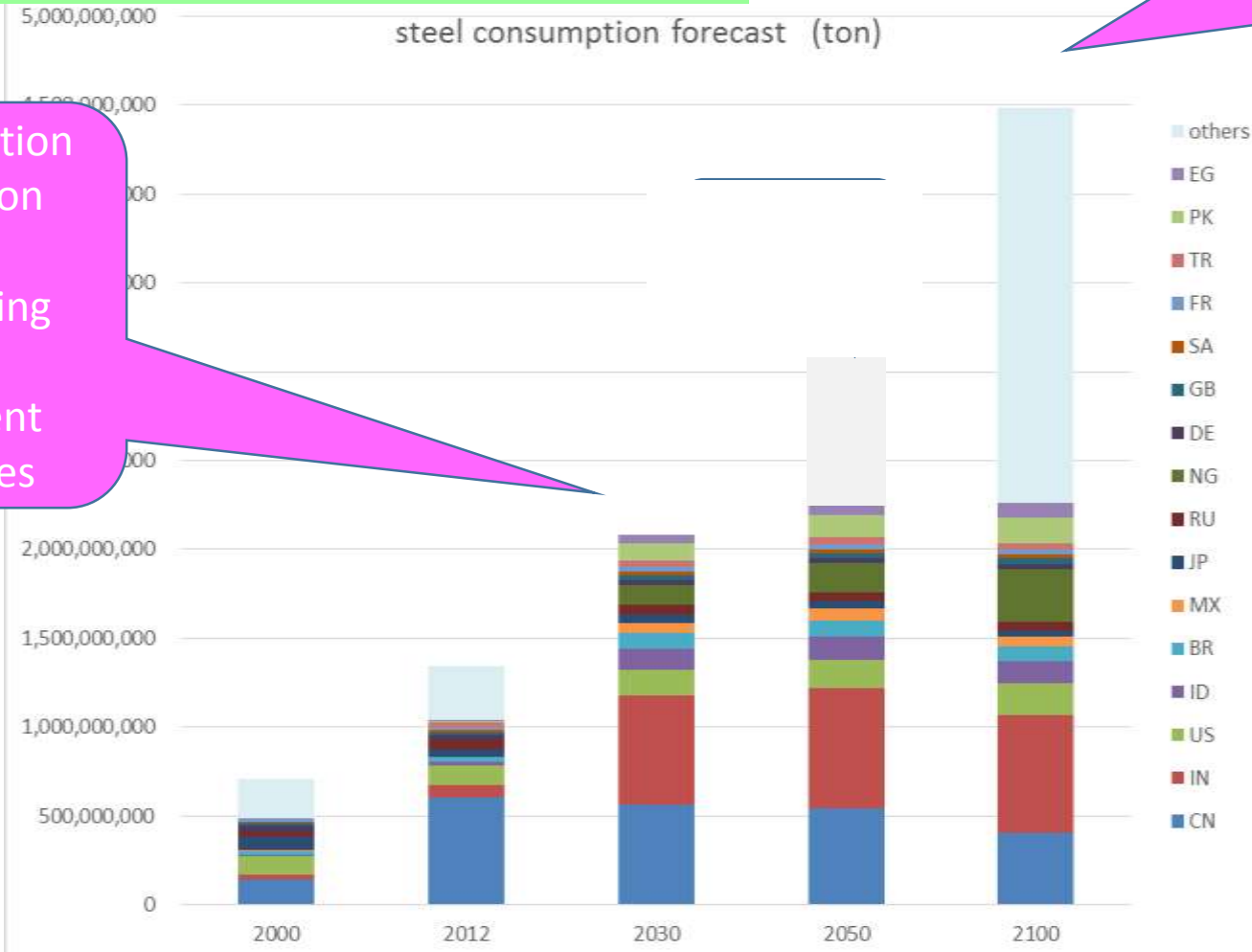
Developed level

Consuming countries

Rough forecast gets to be simpler,
 (population) x (developed consumption level)

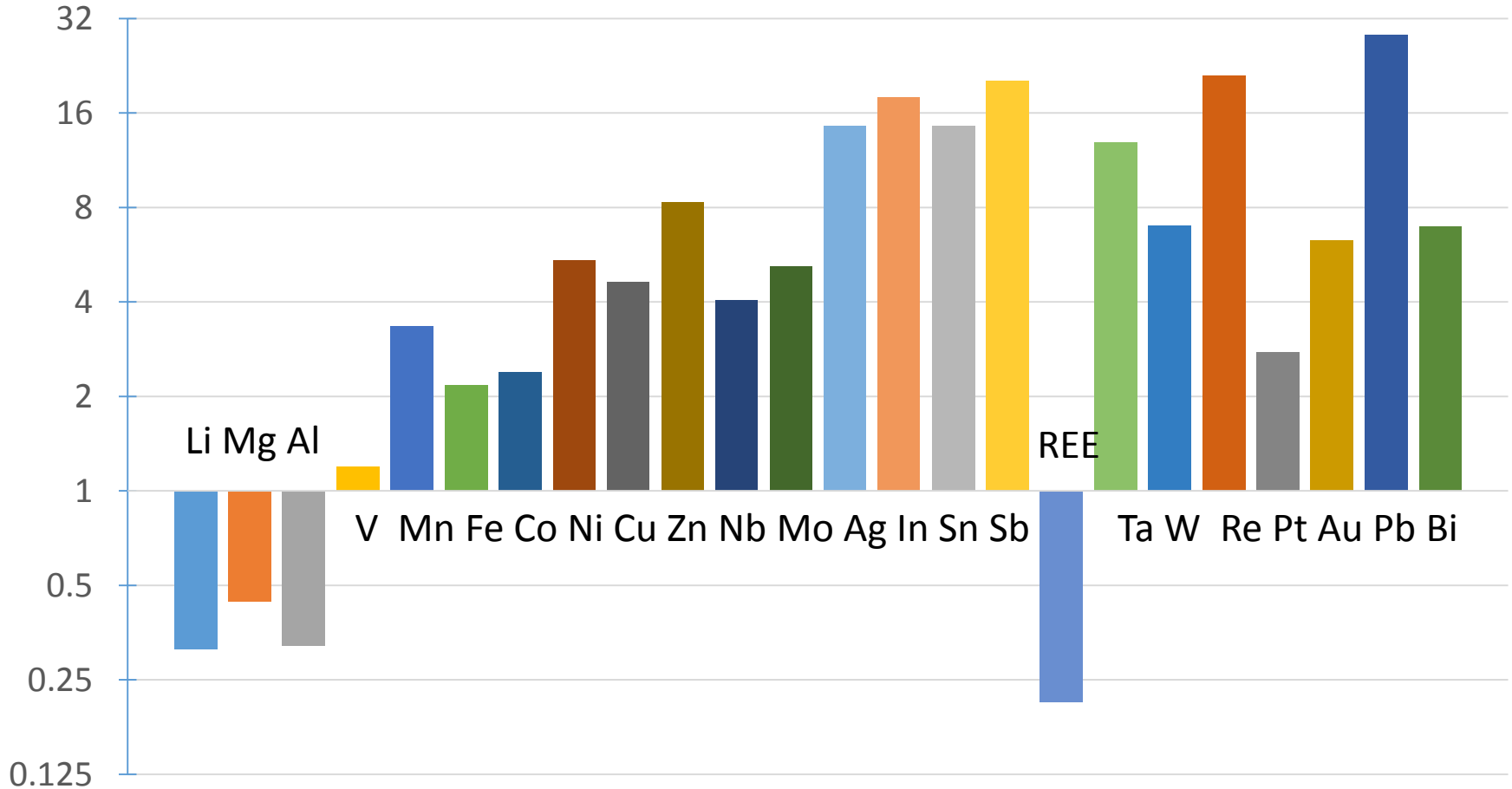
Every country reaches developed level of consumption per capita

Consumption prediction with concerning only prepotent countries

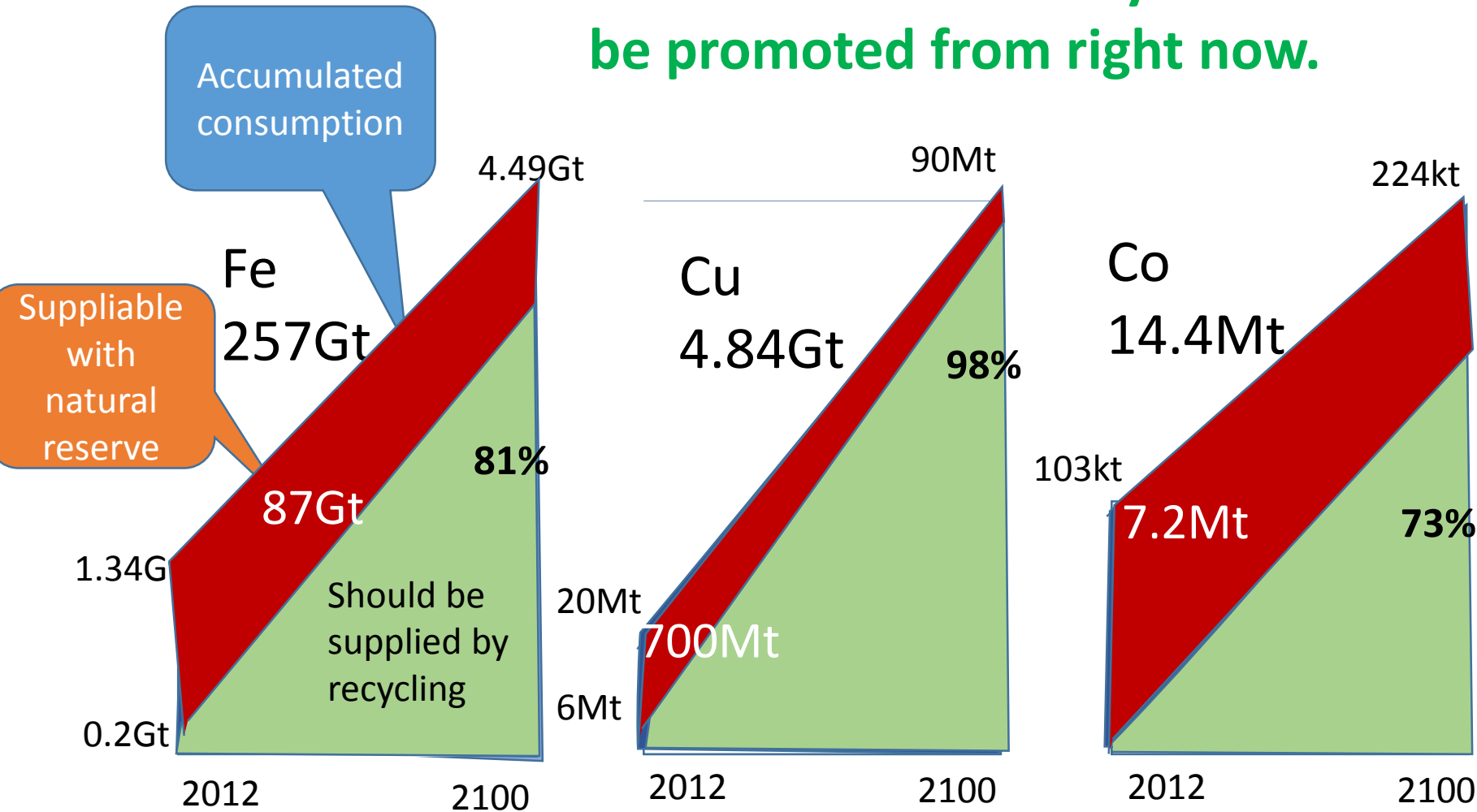


metal	Fe
Consumption/year at 10Gperson world	4.5Gton/year
Reserve	87Gton

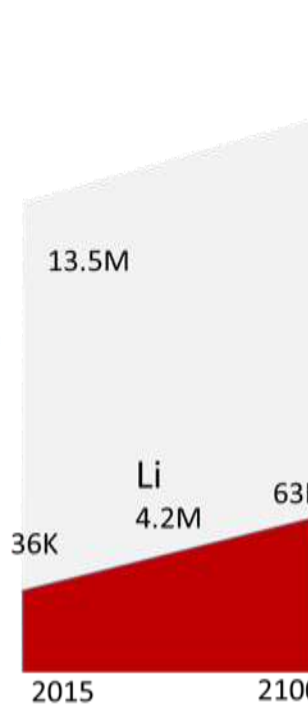
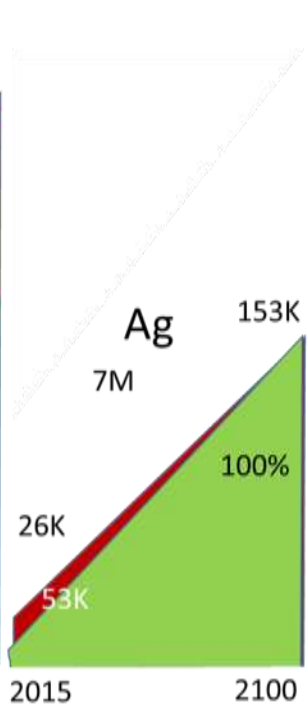
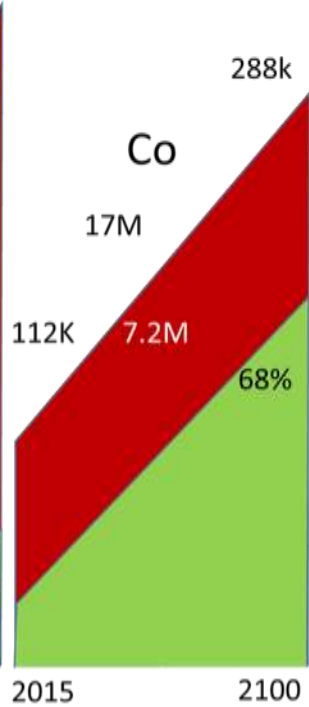
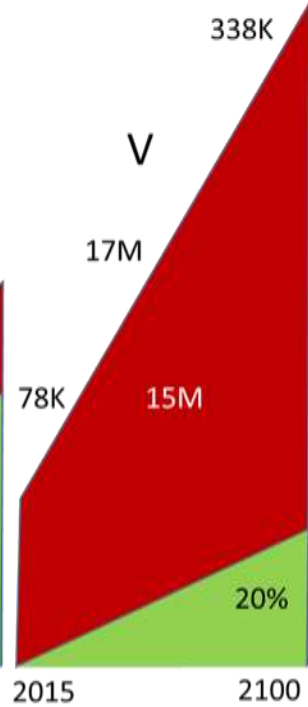
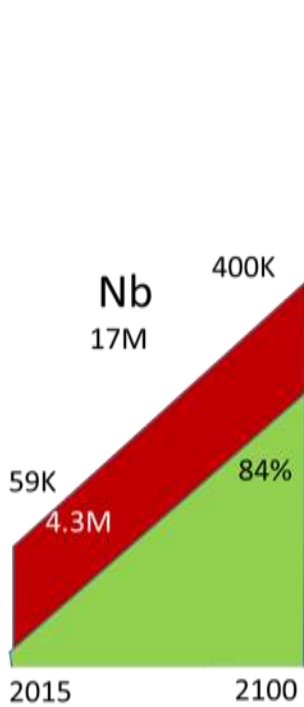
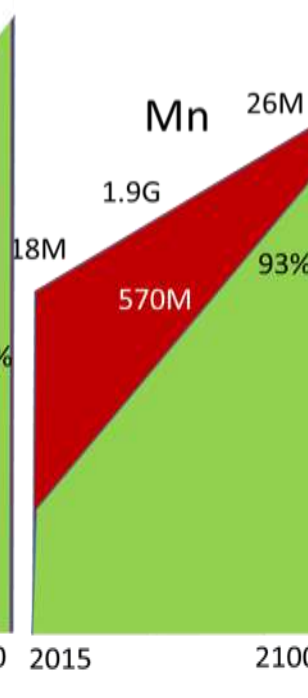
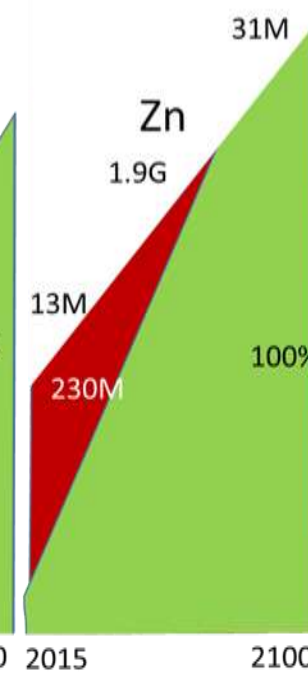
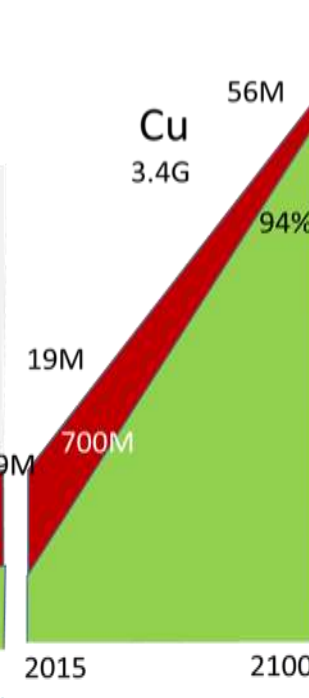
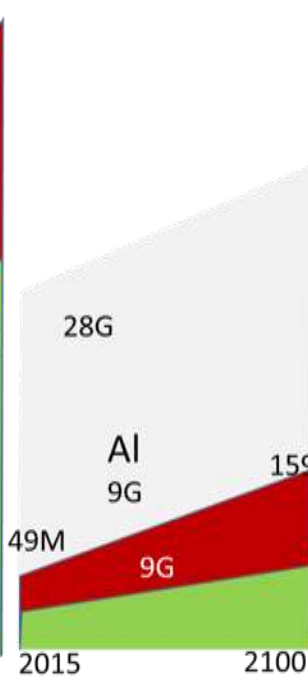
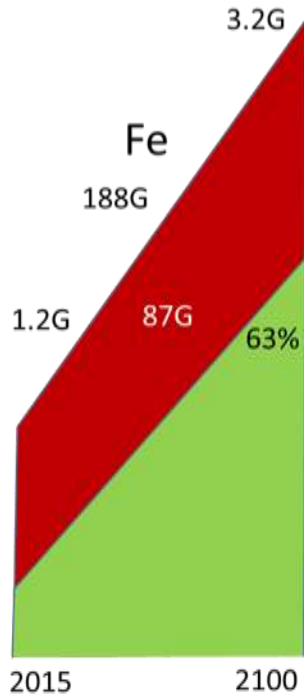
Estimated demand up to 2100 v.s. current reserve amount

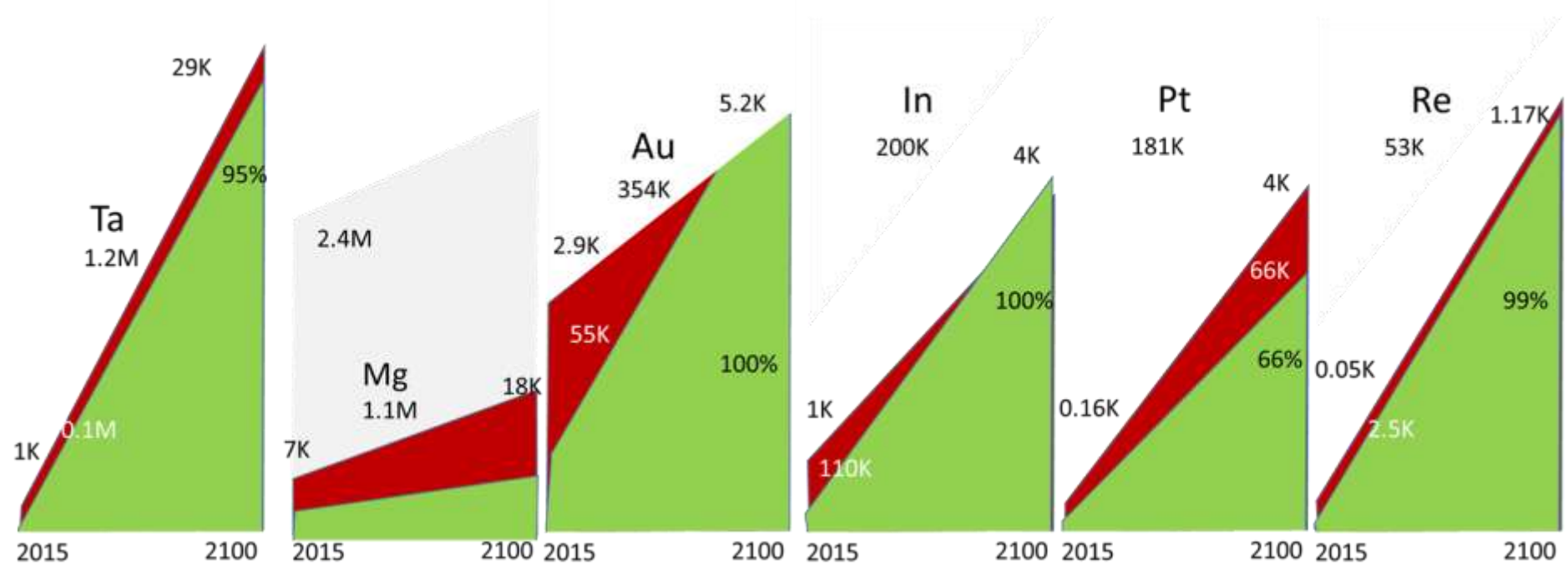
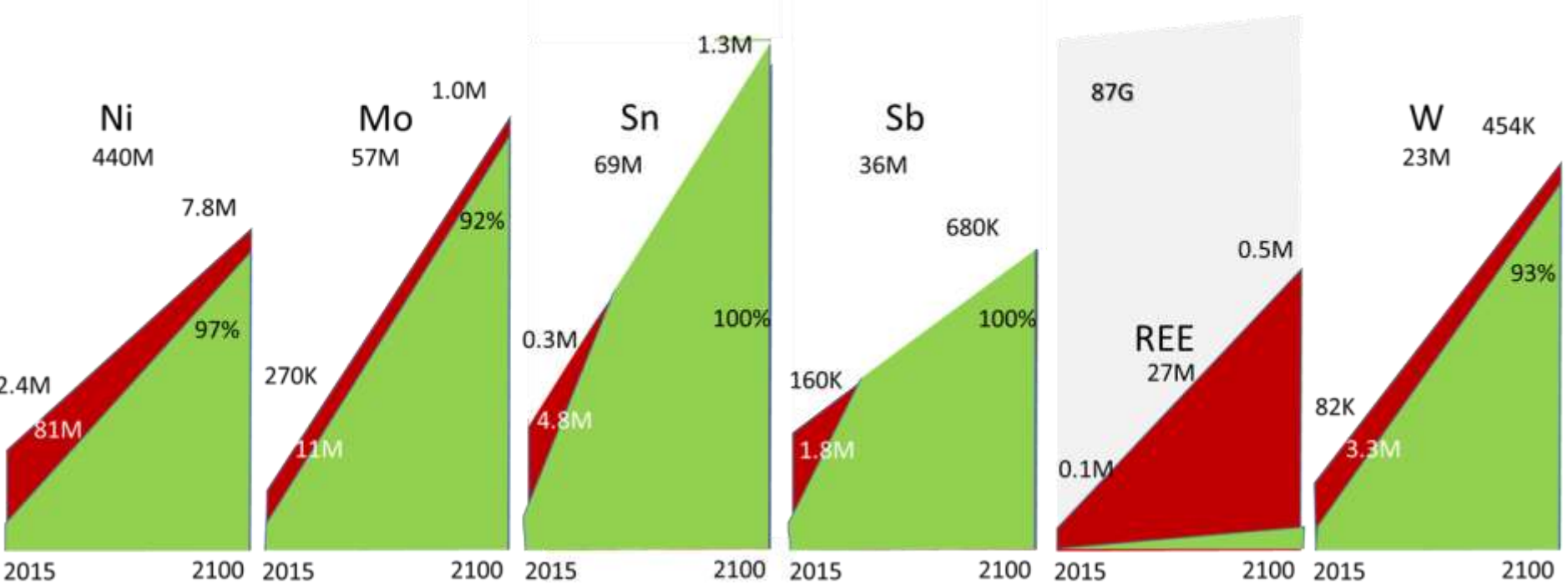


The circulation society must be promoted from right now.



Estimated accumulated consumptions till 2100 with simple assumption of linear growth







The circular economy

Walter R. Stahel

23 March 2016

A new relationship with our goods and materials would save resources and energy and create local jobs, explains Walter R. Stahel.



PDF



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Subject terms: [Economics](#) · [Society](#) · [Materials science](#) · [Policy](#)



Gaming the gamers



Can a video game company tame toxic behaviour?

Scientists are helping to stop antisocial behaviour in the world's most popular online game. The next stop could be a kinder Internet.

Naoko Okamura and 243,150 others like this.



nature
الطبعة العربية



Recent

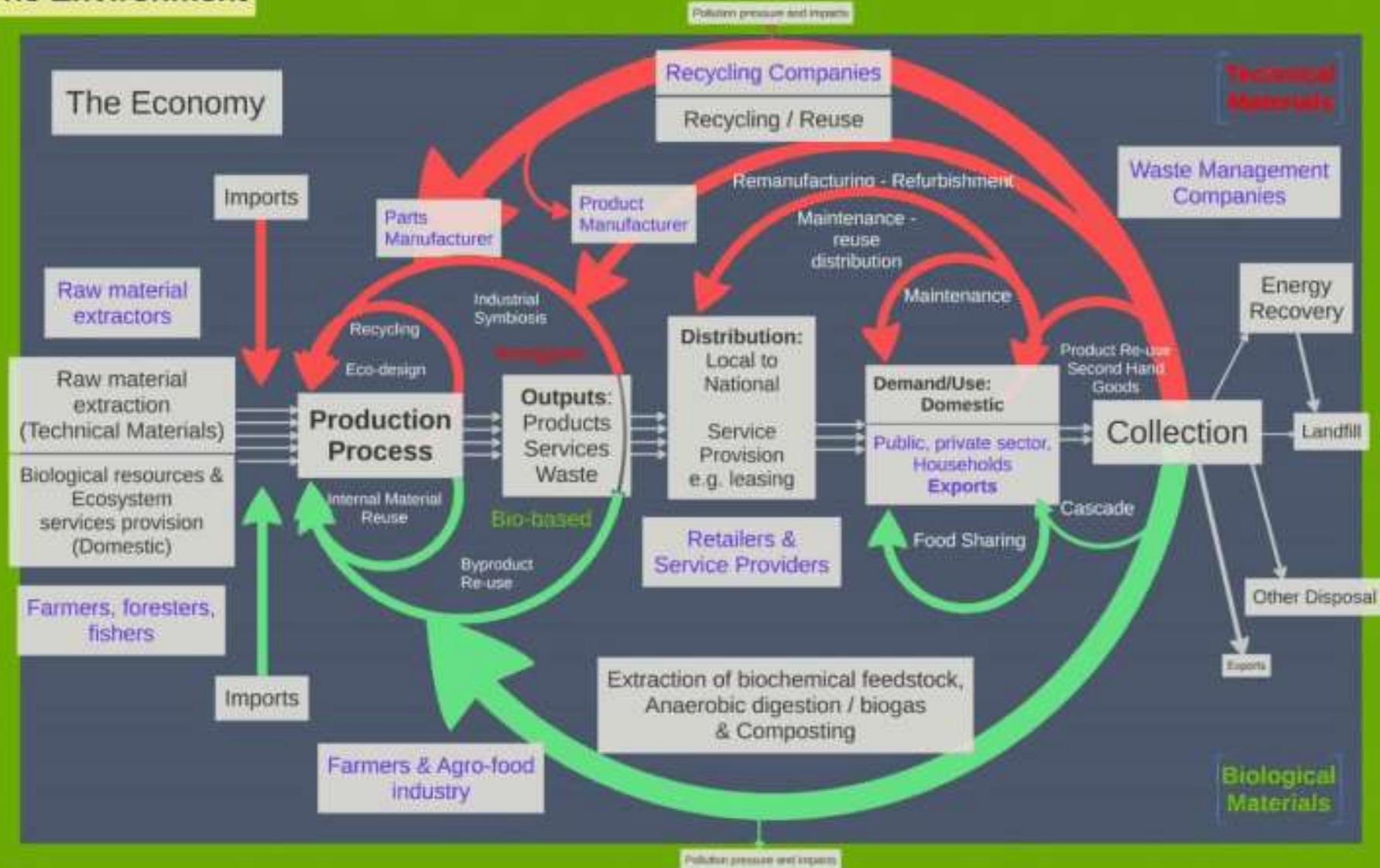
Read

Commented

Figure E2: Simplified illustration of a circular economy

like the circulatory organ, in every corner

The Environment



Source: Own representation, P ten Brink, P Razzini, S. Withana and E. van Dijk (IEEP), 2014



G7 Ise-Shima Leaders' Declaration

G7 Ise-Shima Summit, 26-27 May 2016

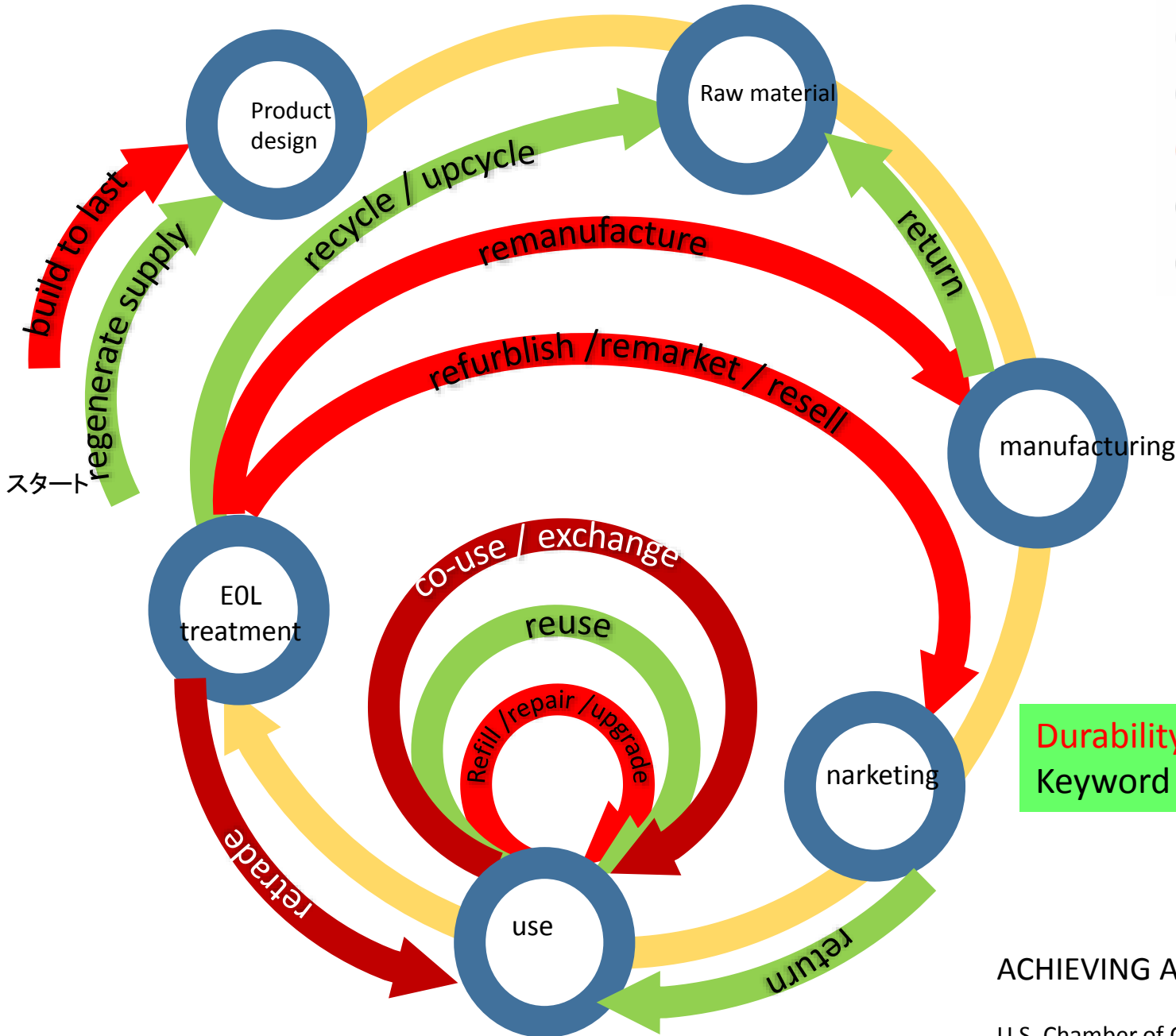
Resource Efficiency and the 3Rs

Achieving the sustainable management and efficient use of resources is addressed in the 2030 Agenda and is crucial for the protection of the environment, climate and planet. Having in mind the importance of sustainable materials management and material cycle societies, we endorse the *Toyama Framework on Material Cycles*. This new framework provides a common vision and a guide for future actions to deepen our efforts on resource efficiency and the 3Rs (Reduce, Reuse, Recycle). We will continue to cooperate through the G7 Alliance on Resource Efficiency. We will work with business and other stakeholders to improve resource efficiency with the aim of also fostering innovation, competitiveness, economic growth and job creation. We encourage all countries to join us in these efforts.

We reaffirm our commitment to address marine litter, recognizing that our efforts on resource efficiency and the 3Rs also contribute to the prevention and reduction of marine litter, particularly plastic, from land-based sources. Furthermore, we support scientific work to enhance global ocean observation and assessment for the science-based management, conservation and sustainable use of marine resources.

BUSINESS MODELS

-  CIRCULAR SUPPLY-CHAIN
-  RECOVERY & RECYCLING
-  PRODUCT LIFE-EXTENSION
-  SHARING PLATFORM
-  PRODUCT AS A SERVICE



Durability becomes the greatest Keyword of Ecodesign

ACHIEVING A CIRCULAR ECONOMY

U.S. Chamber of Commerce Foundation,
Supported by CCC's Circular Economy Network

The Elements with sustainability parameters

- Durable years: (reserve)/(annual consumption)
- Resource-view weight: tons of TMR for 1kg of metal production
- Share % Of top country of production, country code
- Increase of production from 1999 to 2009, (%)

H scarcity TMR domination acceleration												He							
Li 194 1.5 41CL 120		Be 2.5 86US 42												B	C	N	O	F	Ne
Na 0.4 56 100		Mg 5500 0.07 82CN 215												Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr		
2800	0.09	2.	1300 0.04 23AU 220	208 1.5 37CN 135	60 0.03 42ZA 180	40 0.01 22CN 163	92 0.008 39CN 165	122 0.61 40CG 219	41 0.26 19RU 125	31 0.36 34CL 125	22 0.04 28CN 131	7.3	32 0.03 71CN 241	0.03 47 129	0.45 50JP 119	38IL 86			
Rb	Sr	Y6	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe		
0.13	0.51 48ES 133	1 2.7 271	4200 0.55 41AU 151	73 0.64 92BR 335	48 0.75 25US 155		79 79ZA 119	160 2300 79ZA 85	160 810 41ZA 156	14 4.8 18PL 134	0.07 23CN 94	24 12 50CN 250	22 2.5 37CN 153	0.06 91CN 136	10 44JP 88	600 59CL 159			
Cs	Ba	(Ln)	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
0.01	31 0.51 147	800 - 97CN 162	10 151	33 6.8 48AU 245	40 0.2 81CN 185	18 48CL 118	540 79ZA	400 79ZA 40	160 530 79ZA 118	17 1100 13CN 101	32 2 63CN 56	0.4 67	17 0.03 43CN 128	57 0.22 62CN 221					
Fr	Ra	(An)	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
			1600 8.2 371*	770 18 295*	7.9	420 12 90*		16	188 33	17	244 55	209 16	30	12	32	32	32		
			Ac	Th	Pa	U													
						22													

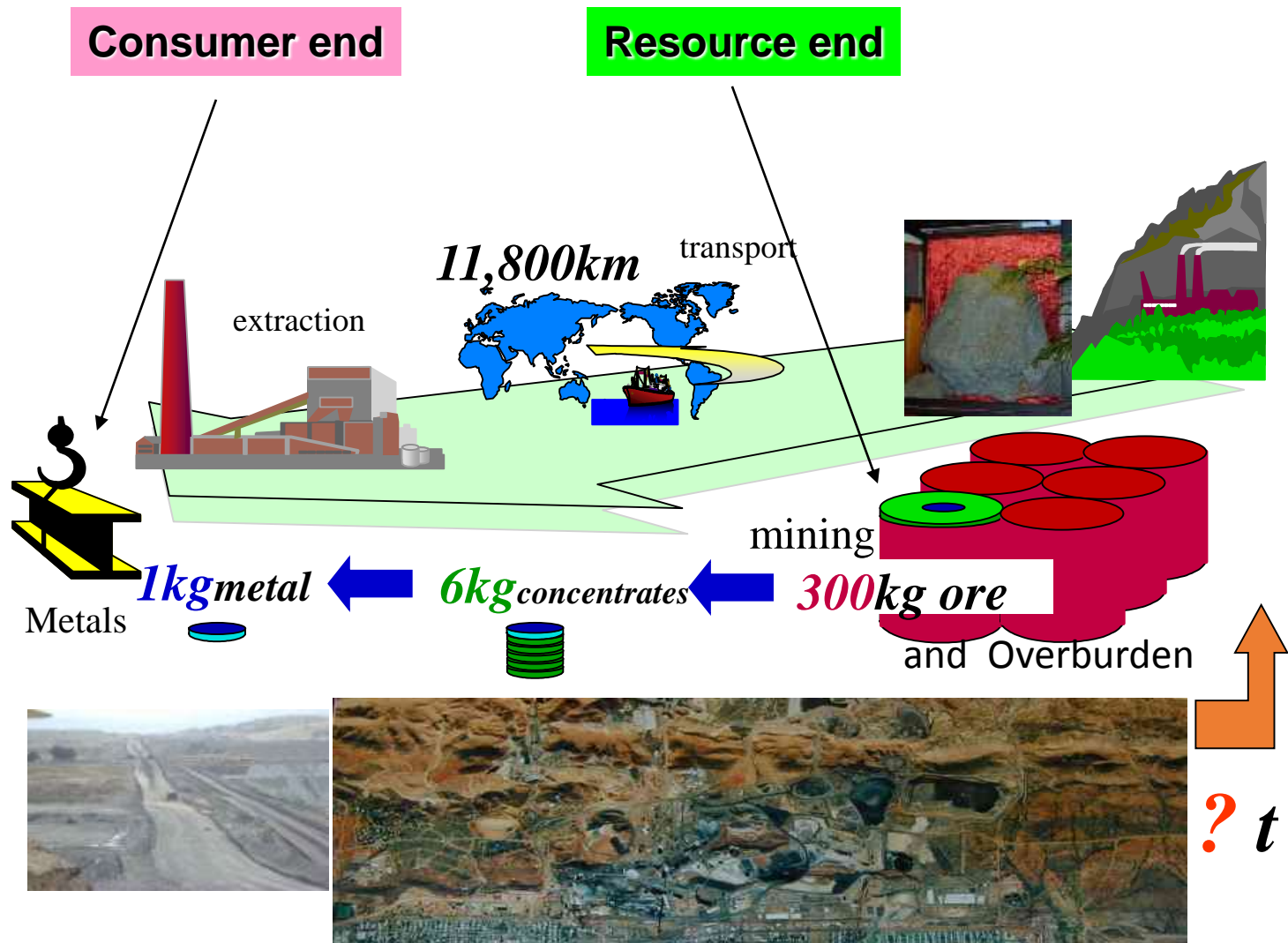
- Magnet, motor
- Batteries
- IC tips and parts
- Electric wiring
- lightning
- Optical function
- Information media
- Thermolectric,
- Catalyst, electrode
- Structural material
- Display & its porishing
- Fire retardant
- Solar cell



* Estimated by import of Japan, () amount in crust is less than in sea water

Data form 米国鉱山局データ USGS minerals information
 工業レアメタル (Kogyo rare metal) Japanese journal
 「概説 資源端重量」 NIMS-EMC data on mat. & env. No.18
 Halada, Katagiri, Proc. of EcoBalance 2010 p609

Resource(-end)-view weight



TMR: Total Materials Requirements, or Ecological rucksacks



Mashijing 馬石徑



Mashijing 馬石徑

垵上

24

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© 2010 Kingway Ltd.

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圖像取得日: 2005/4/14

24° 58'14.42" N 115° 03'12.45" E 標高 378 m

高度 2.13 km

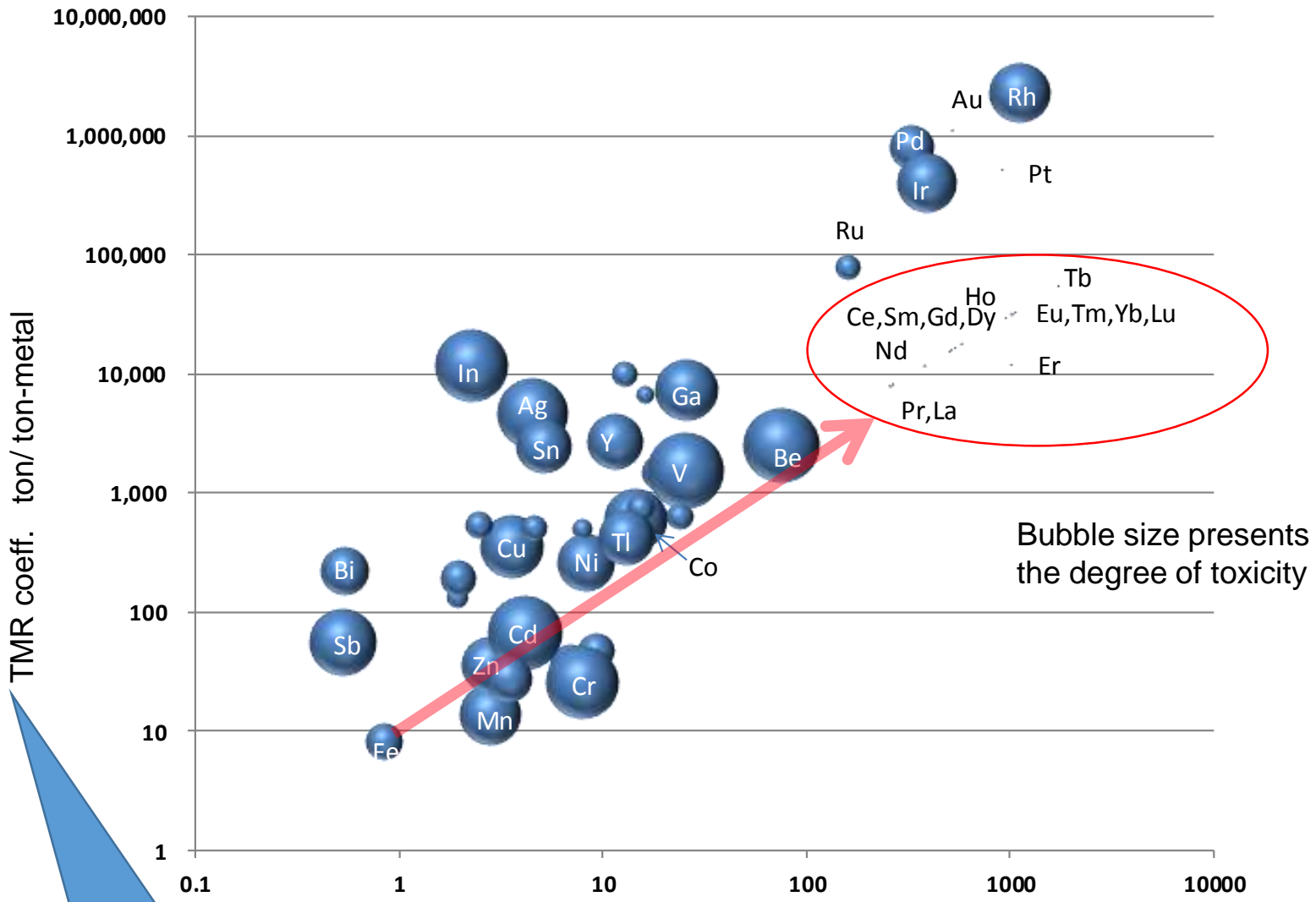
Ecological Rucksack

3g platinum ring



More than 1.5ton ecological rucksack

1kg R.E.E. is nearly equivalent to 1 ton Fe by environmental view

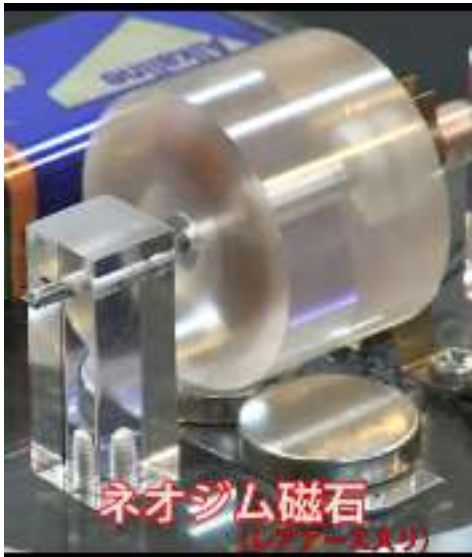


Bubble size presents the degree of toxicity

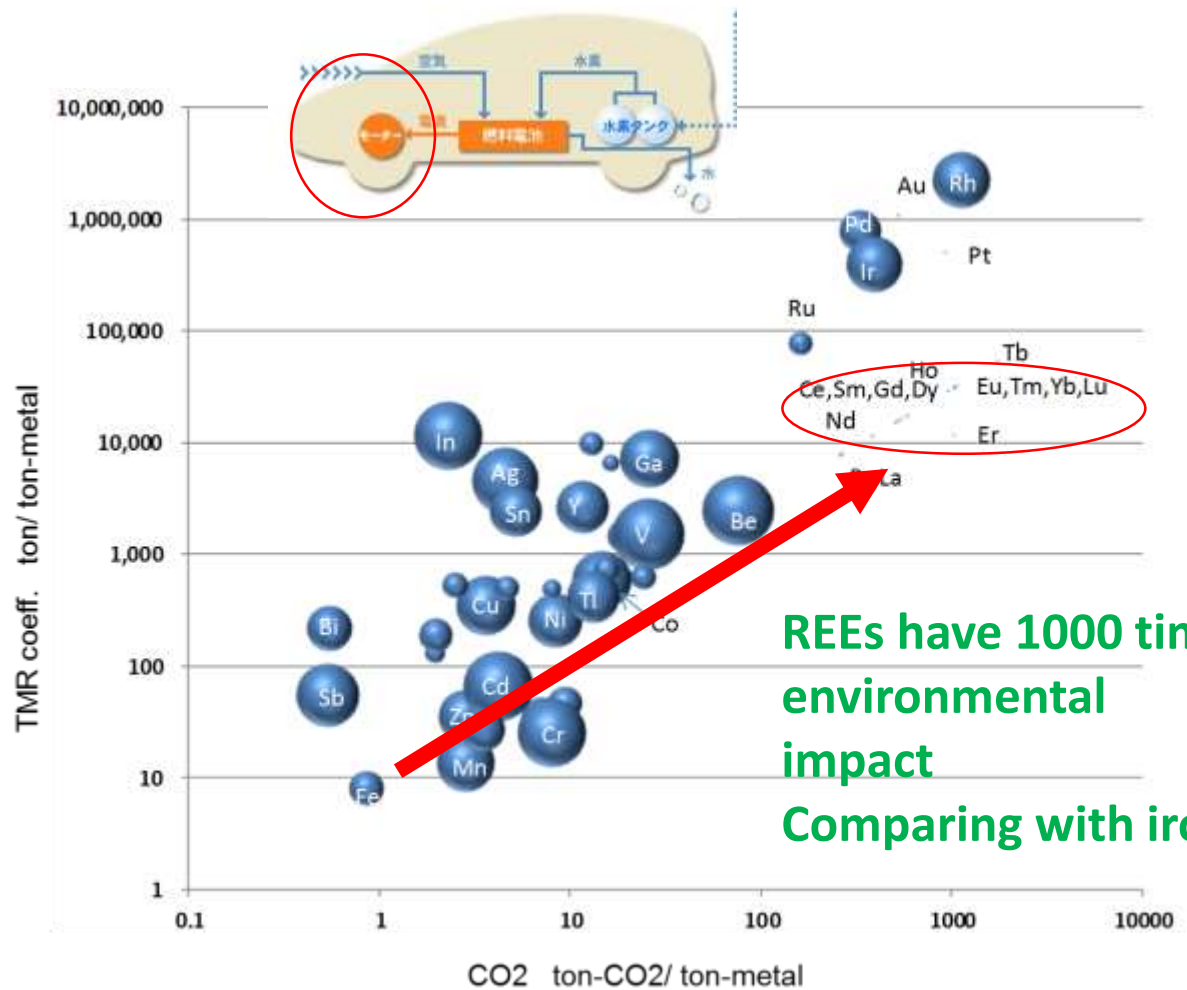
Total material requirement
 ≈
 Waist from mining

CO2 emission during mining and extraction

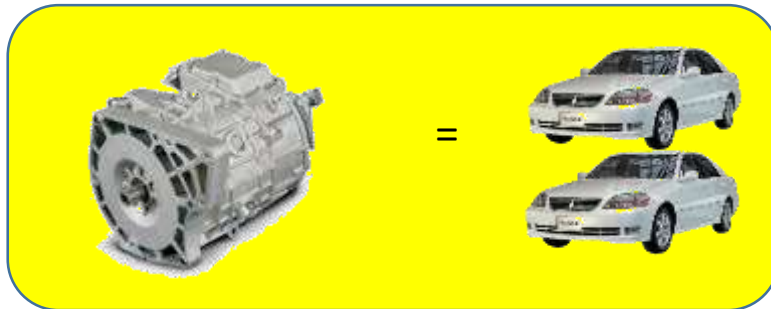
EV motor (Rare Earth)



1.2kg Nd magnet/car



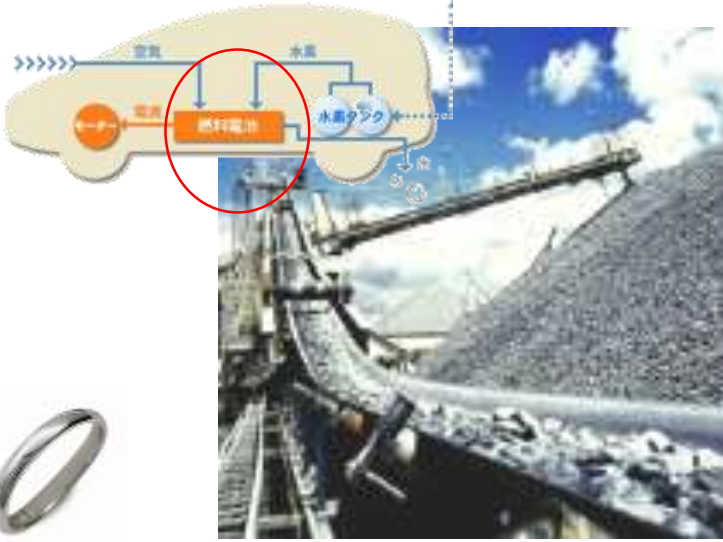
REEs have 1000 times environmental impact
Comparing with iron



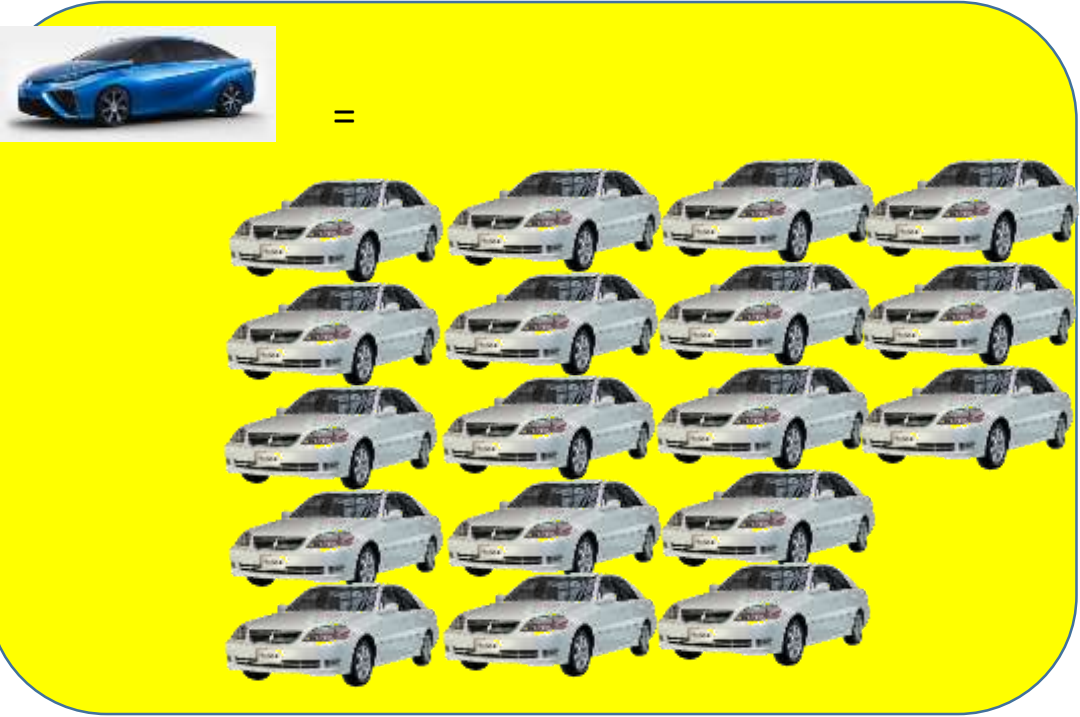
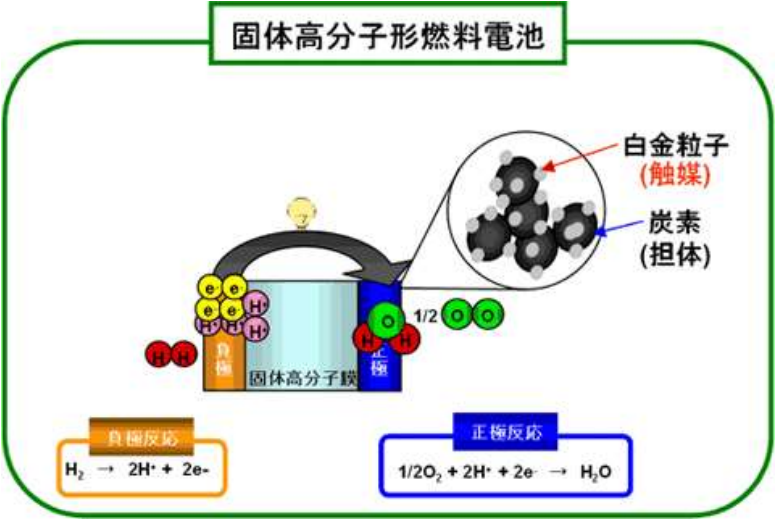
Small mass but Great impact

Fuel Cell (Pt)

	Pt consumption g/car
Small car 80kW	32
Medium car 150kW	60
Large car 250kW	150
average 120kW	50



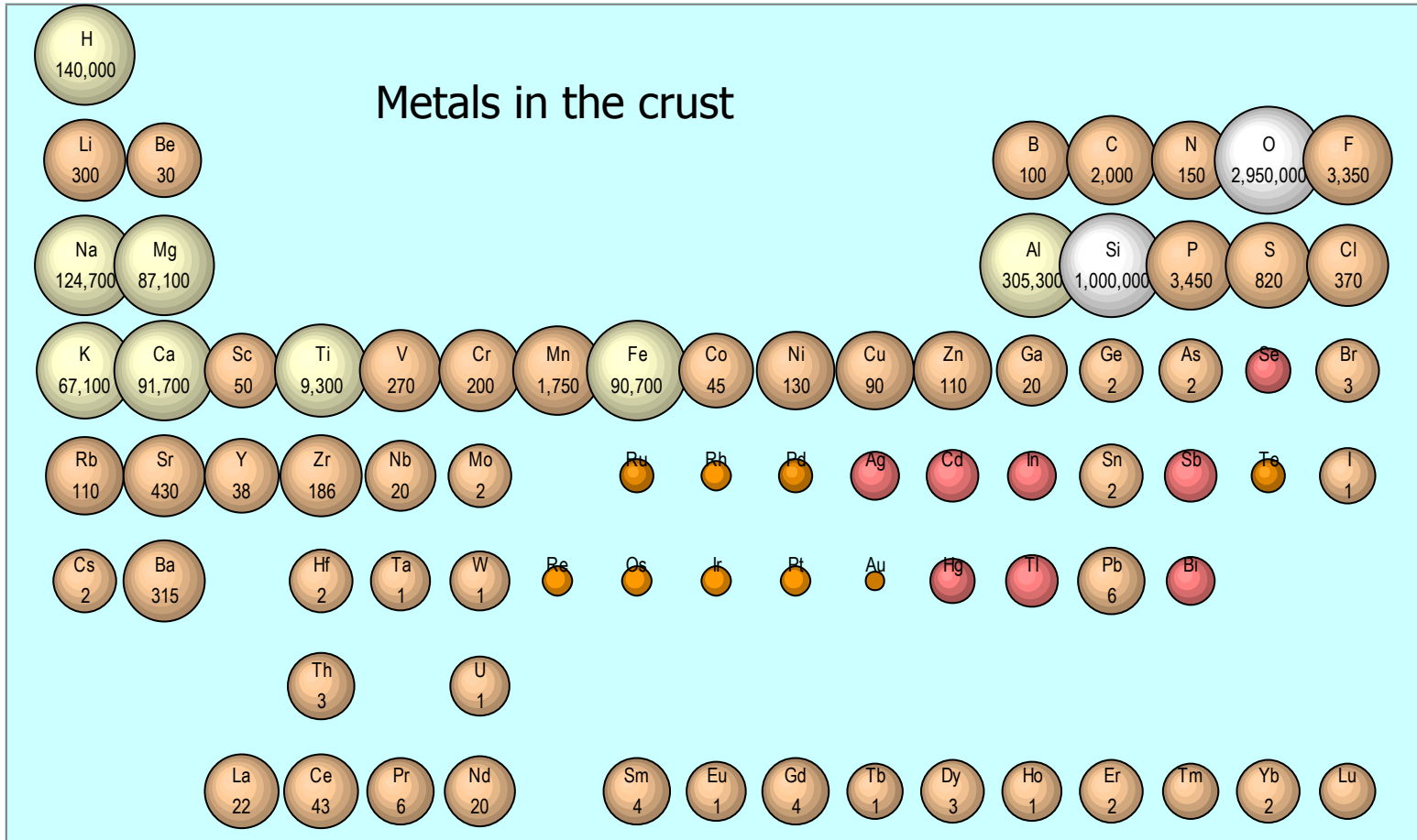
3g platinum = 3.6ton resource = iron for 1 automobile



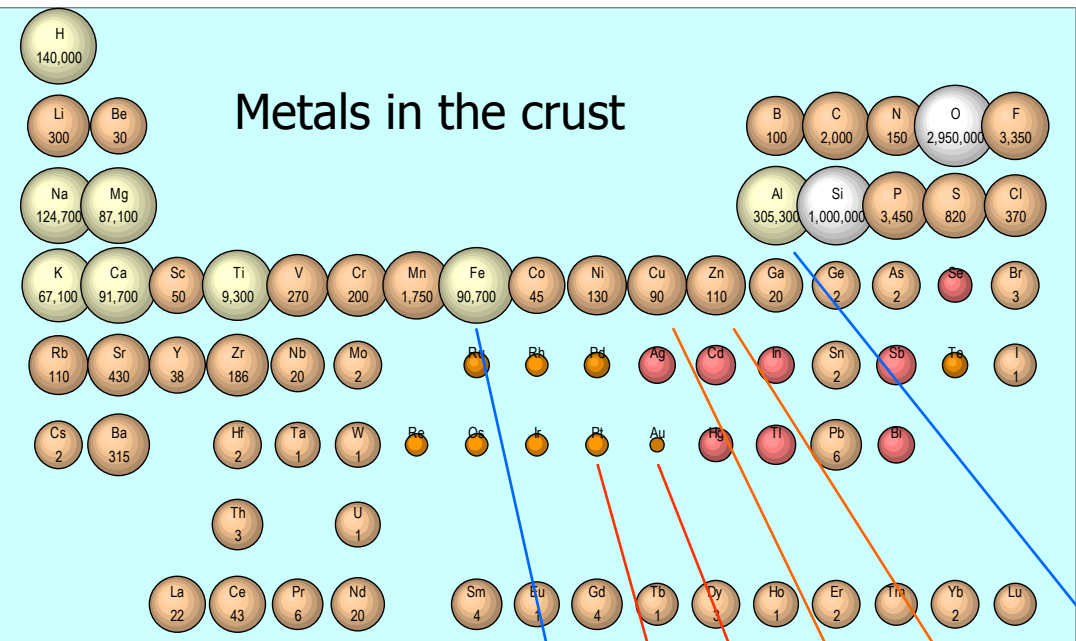
To promote the resource efficiency,

it is important to use common element in the crust.

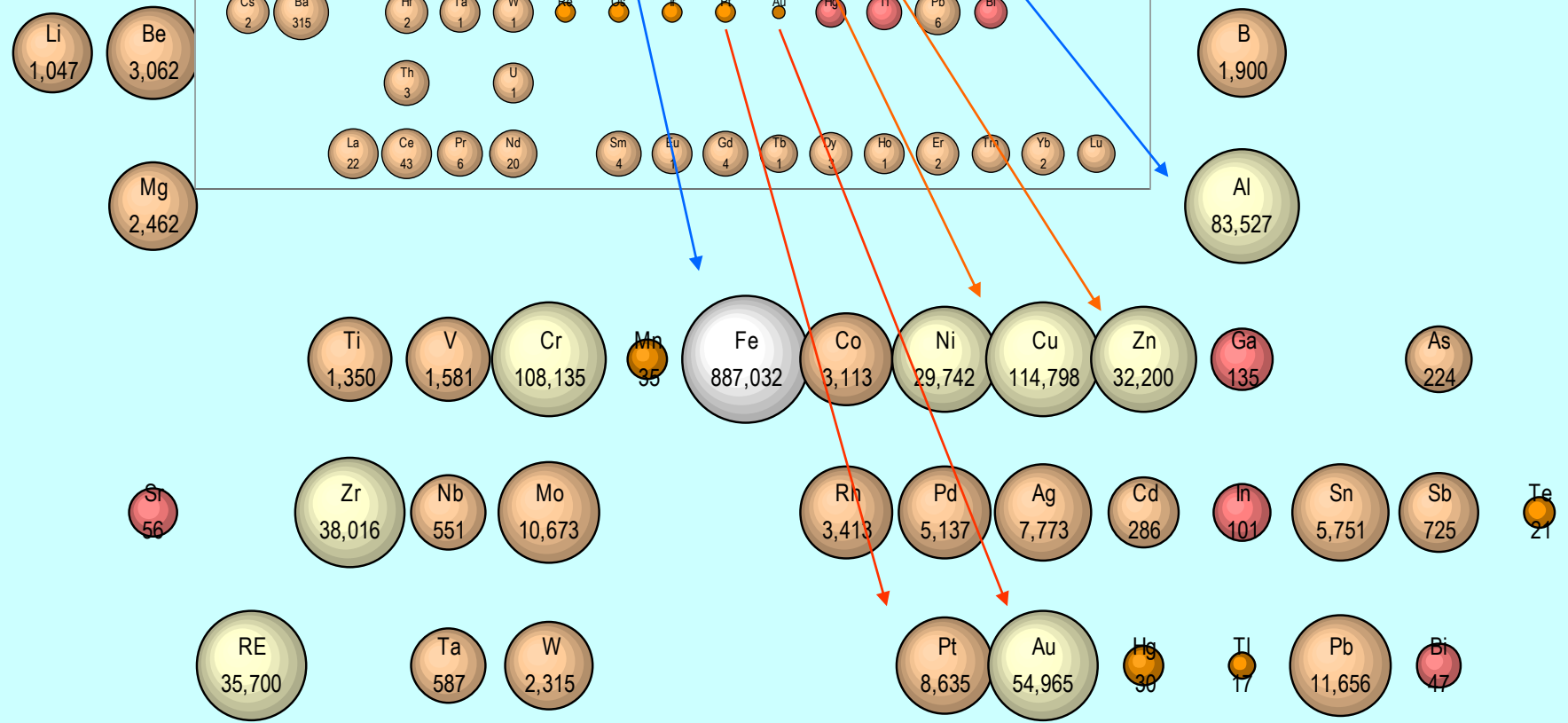
educing functions by controlling electron orbit of chemical compounds.



Metals in the crust



Market size of metals



We are still in front of the entrance of sound material use

In order to achieve the decoupling of development
and environmental impact,

**Life-cycle consideration and
the pursuit of Resource Efficiency are required.**

Ceramics have advantage of Resource Efficiency
in resource view weight and
durability.

,but have to consider the environmental impact
in processing processes.

Grazie per l' attenzione !!